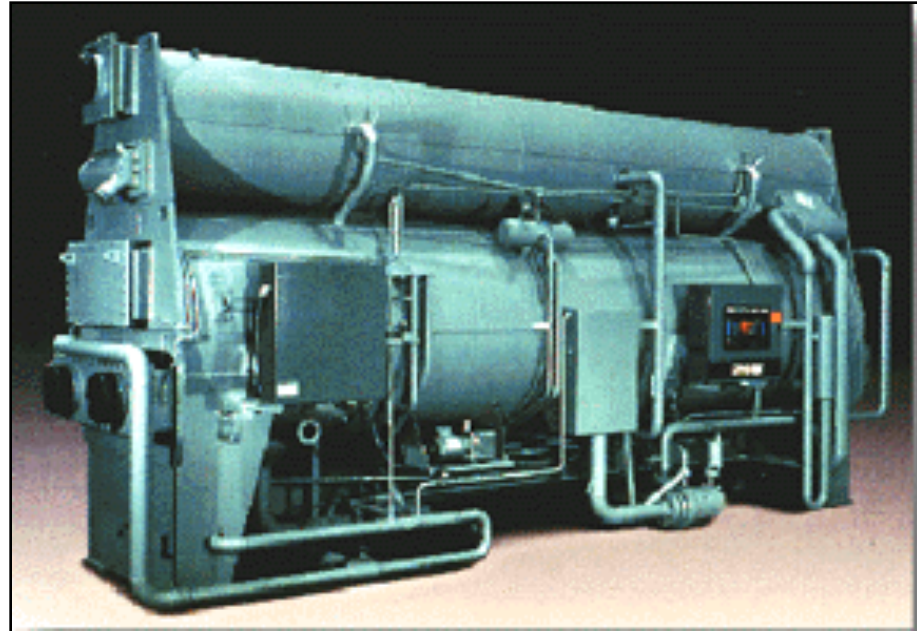




# Basic Absorption Training Course



Robert Smith,

Yo

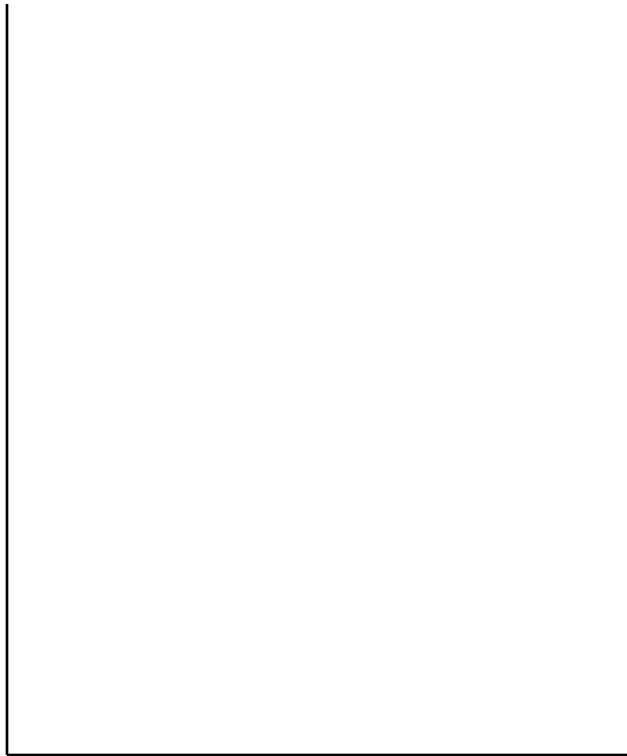
Service Training Department

# Absorption Cooling Basics

# Pressure - Enthalpy Chart

---

Pressure

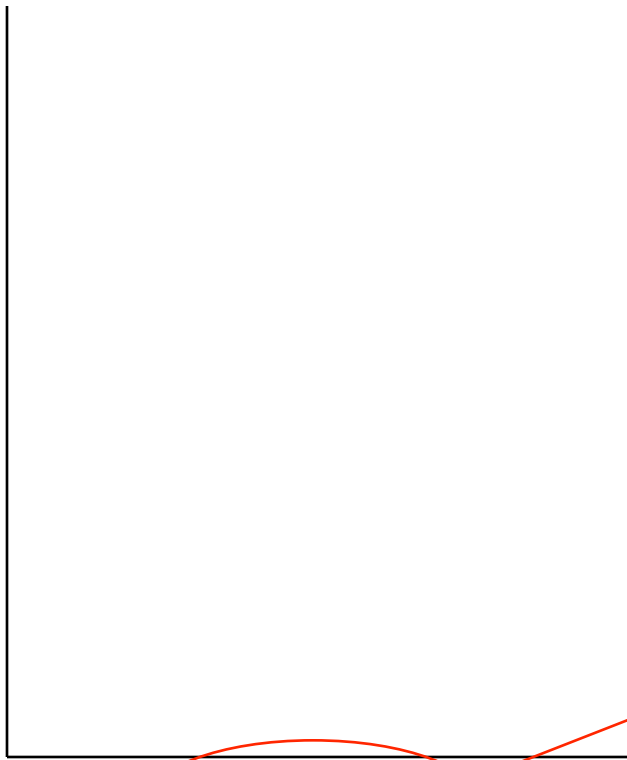


Enthalpy

# Pressure - Enthalpy Chart

---

Pressure

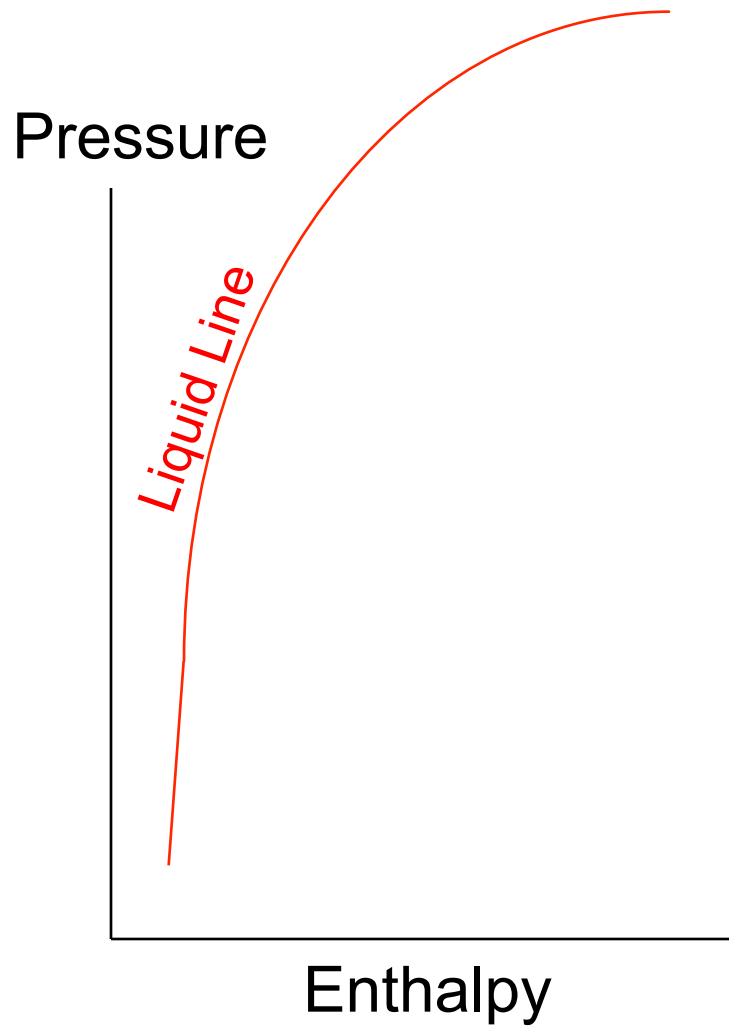


Heat Content  
(BTU / lb.)

Enthalpy

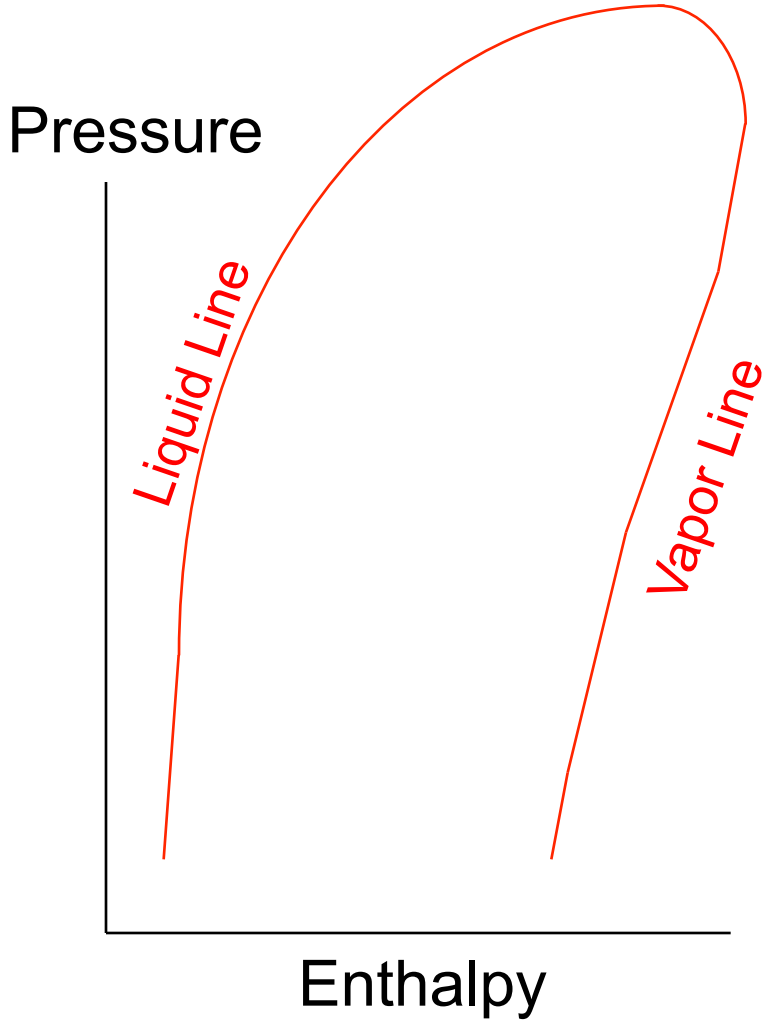
# Pressure - Enthalpy Chart

---



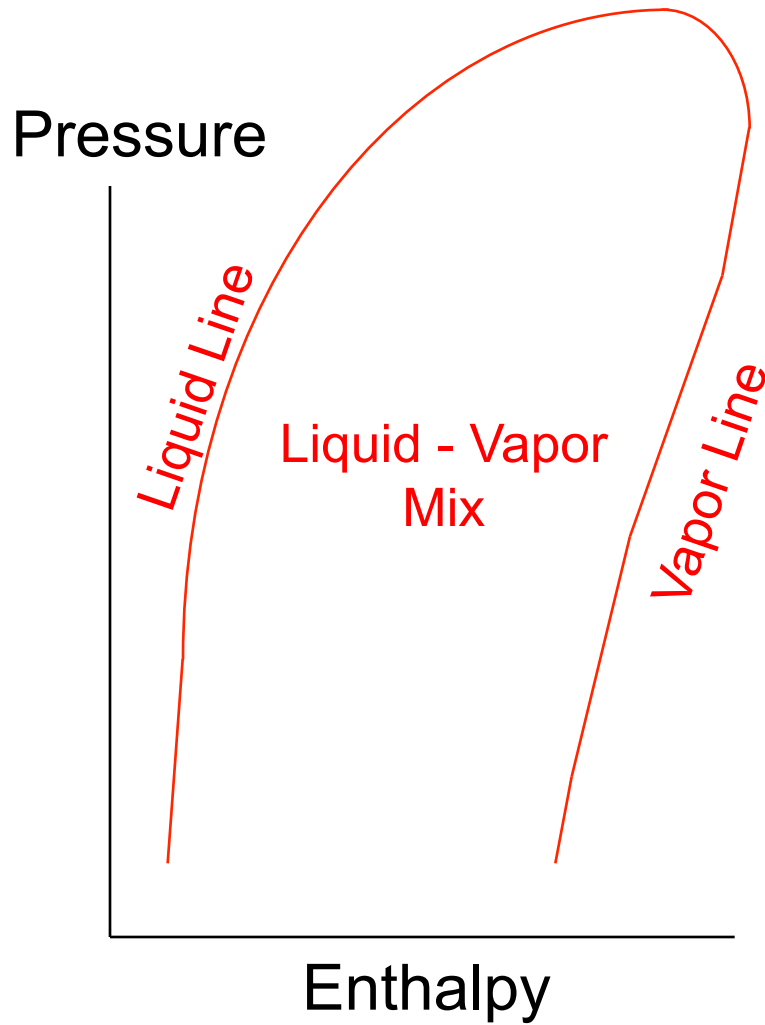
# Pressure - Enthalpy Chart

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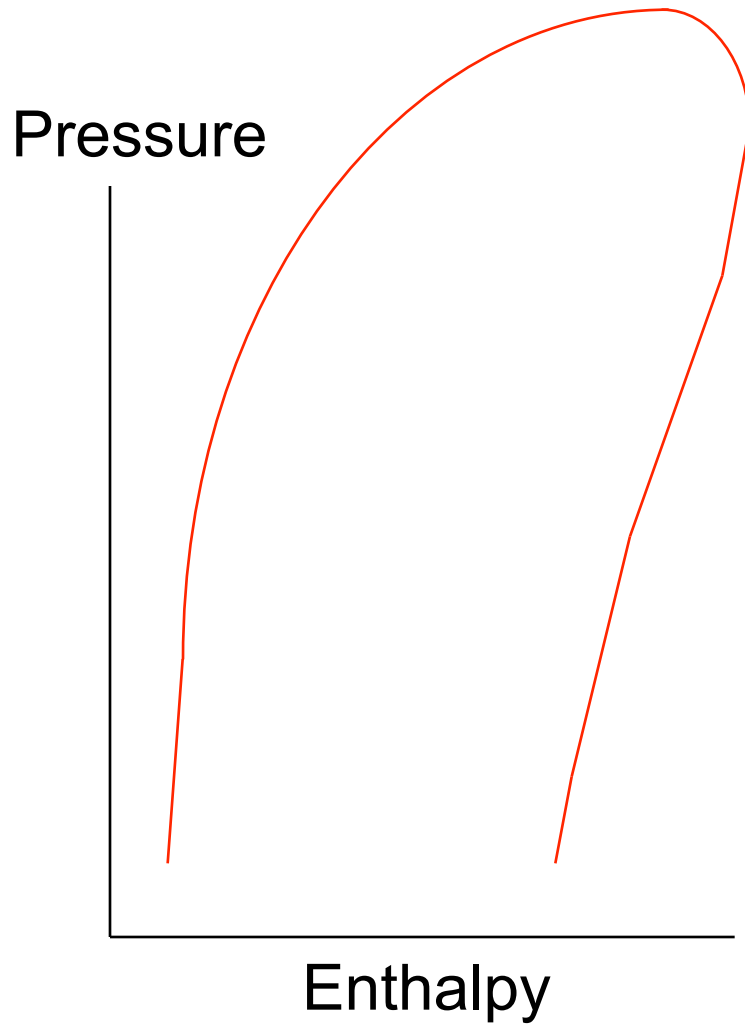
# Pressure - Enthalpy Chart

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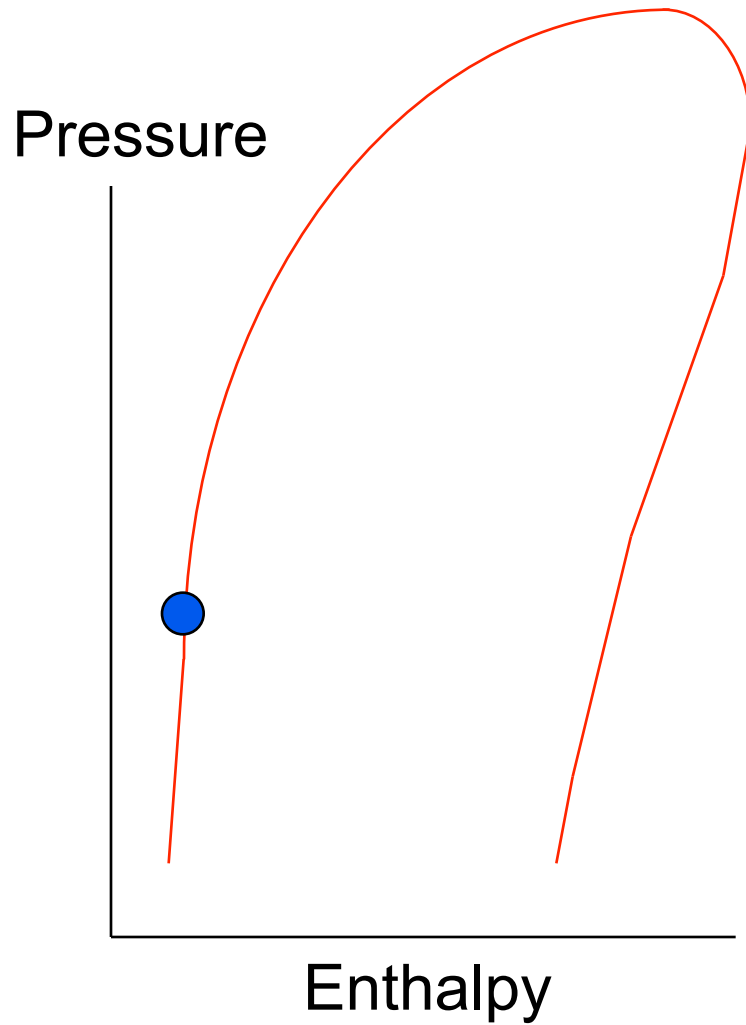
# Pressure - Enthalpy Chart

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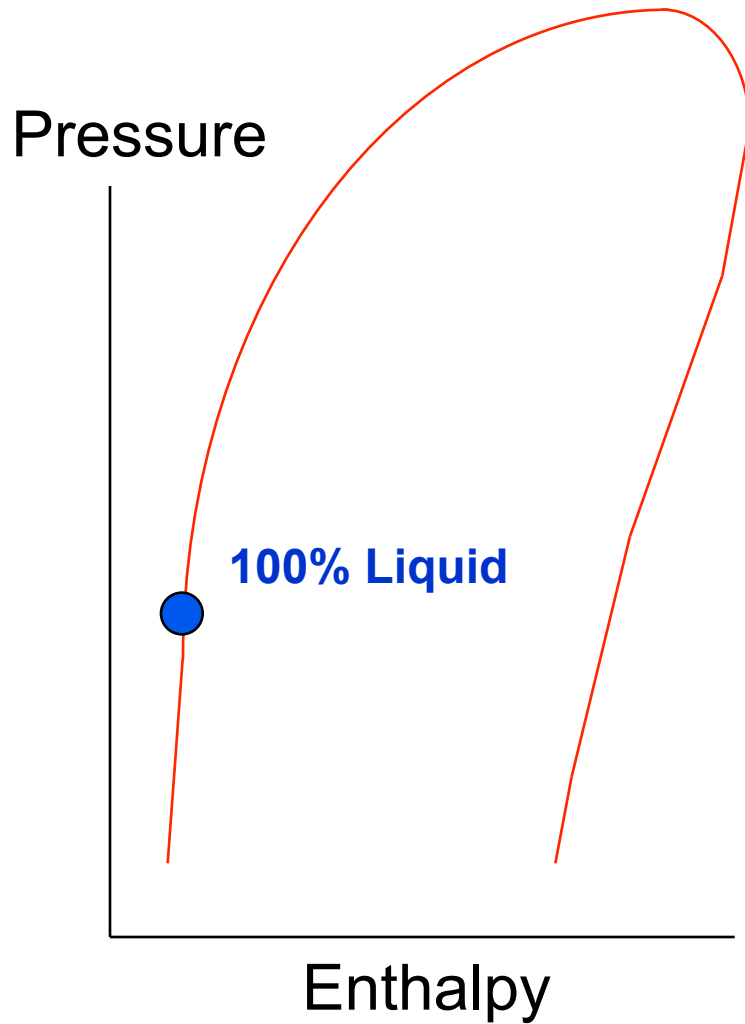
# Pressure - Enthalpy Chart

---



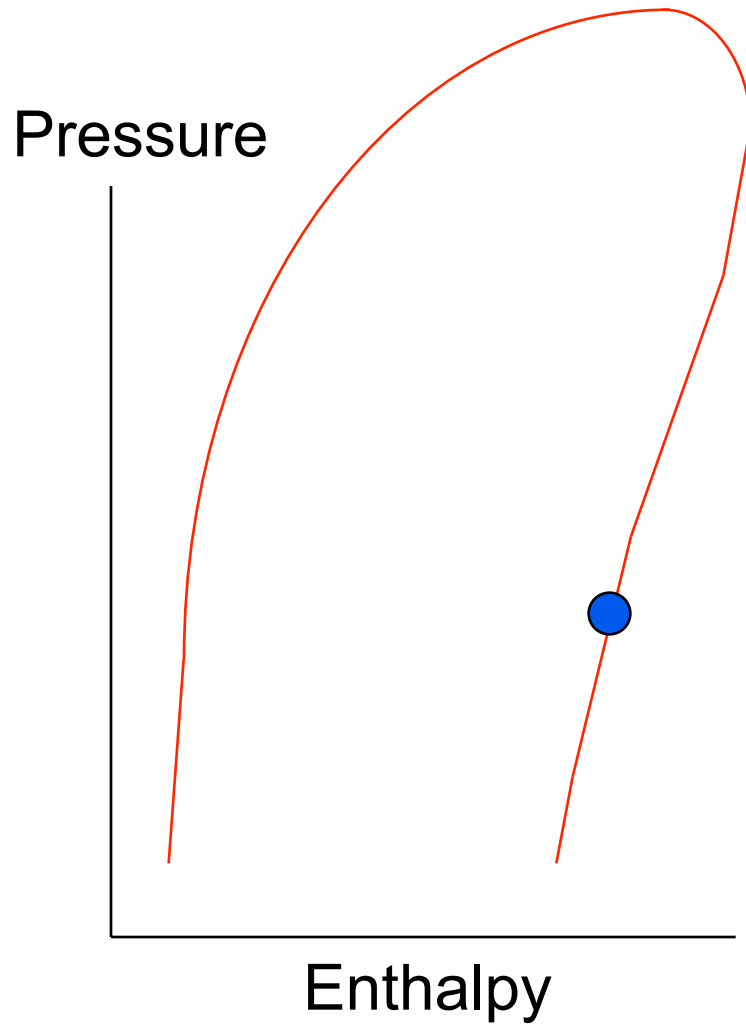
# Pressure - Enthalpy Chart

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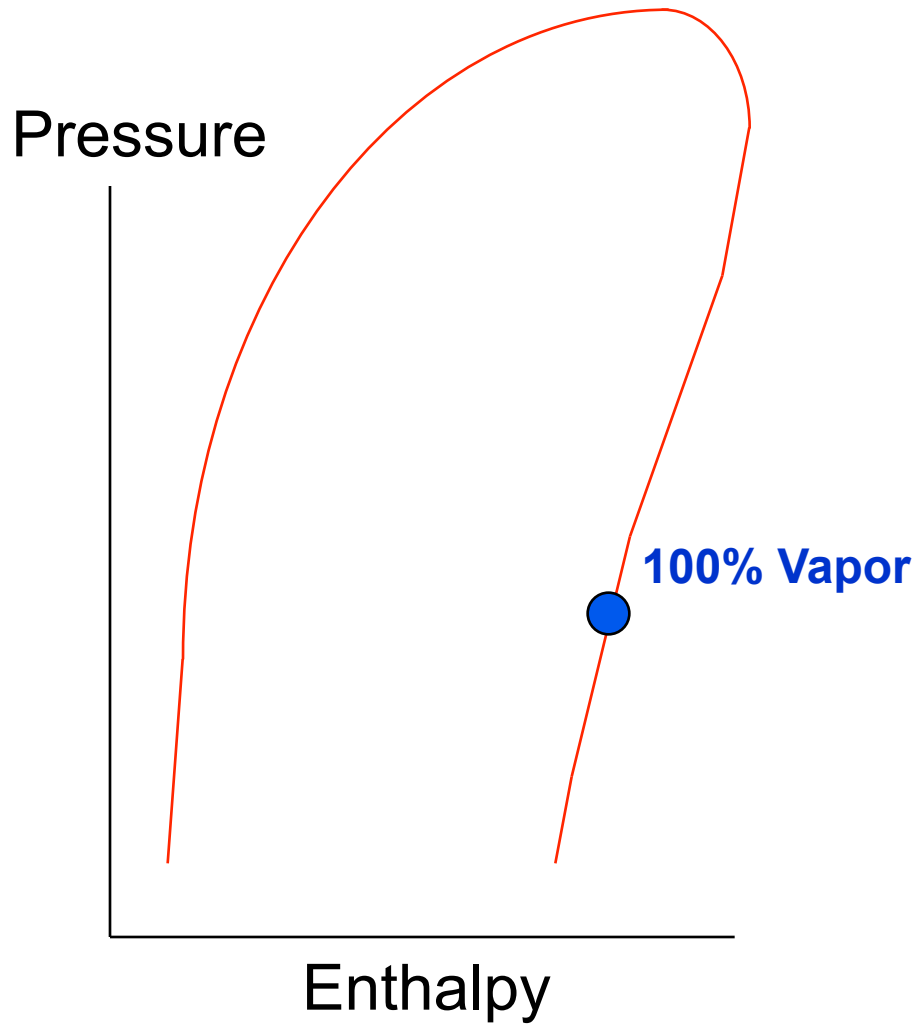
# Pressure - Enthalpy Chart

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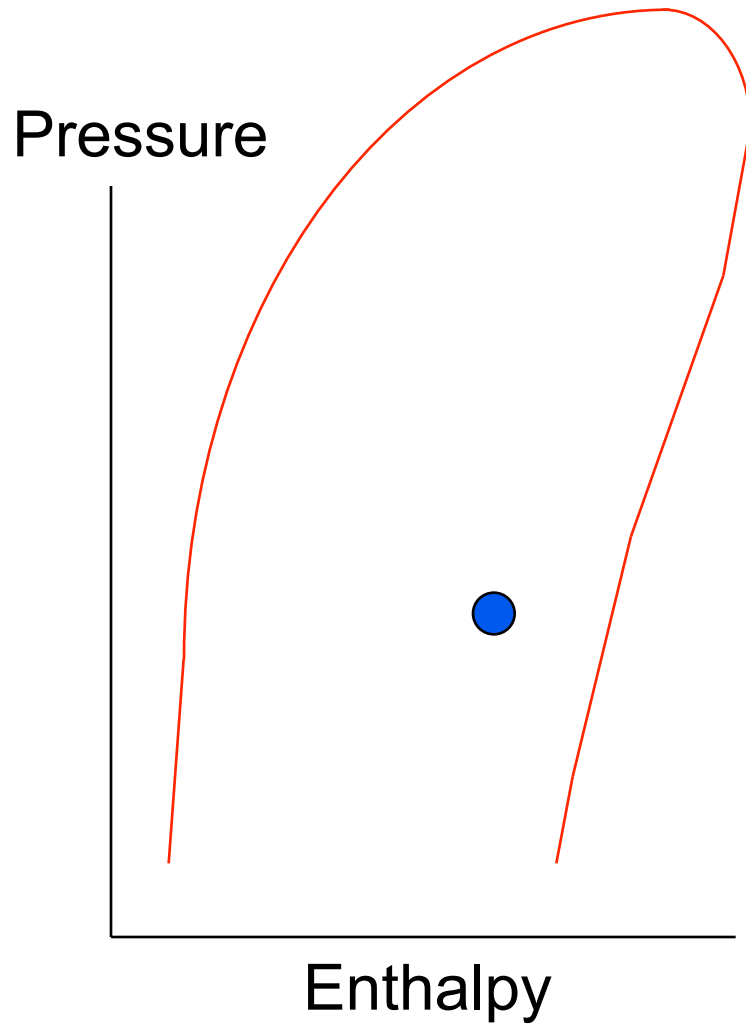
# Pressure - Enthalpy Chart

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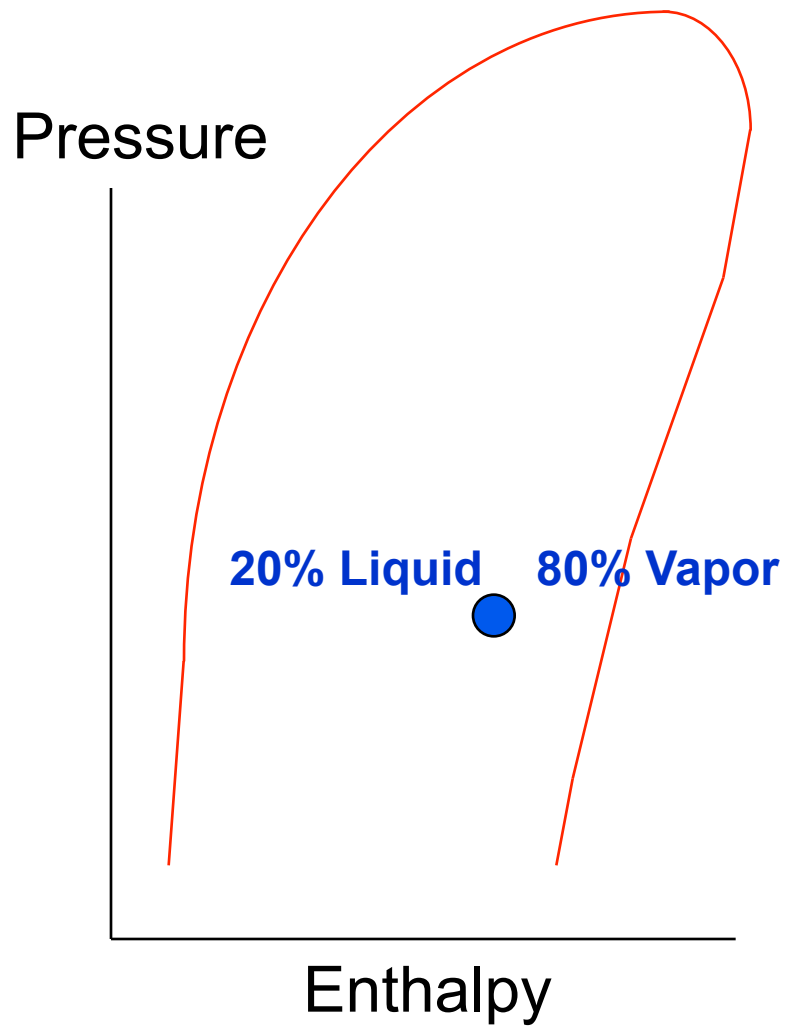
# Pressure - Enthalpy Chart

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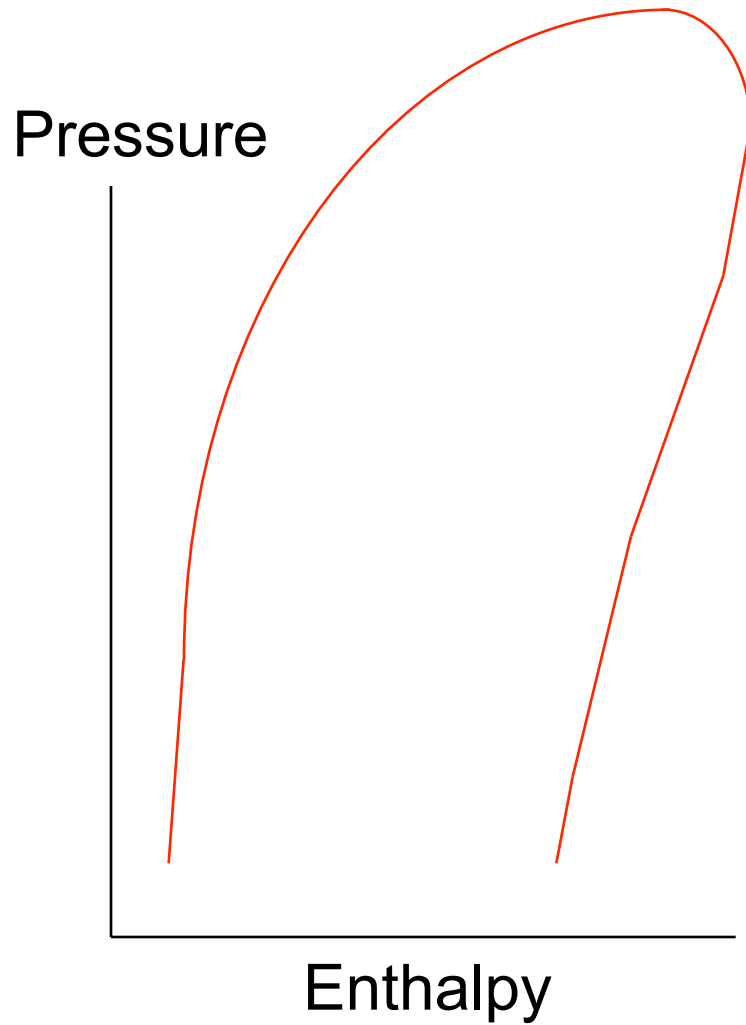
# Pressure - Enthalpy Chart

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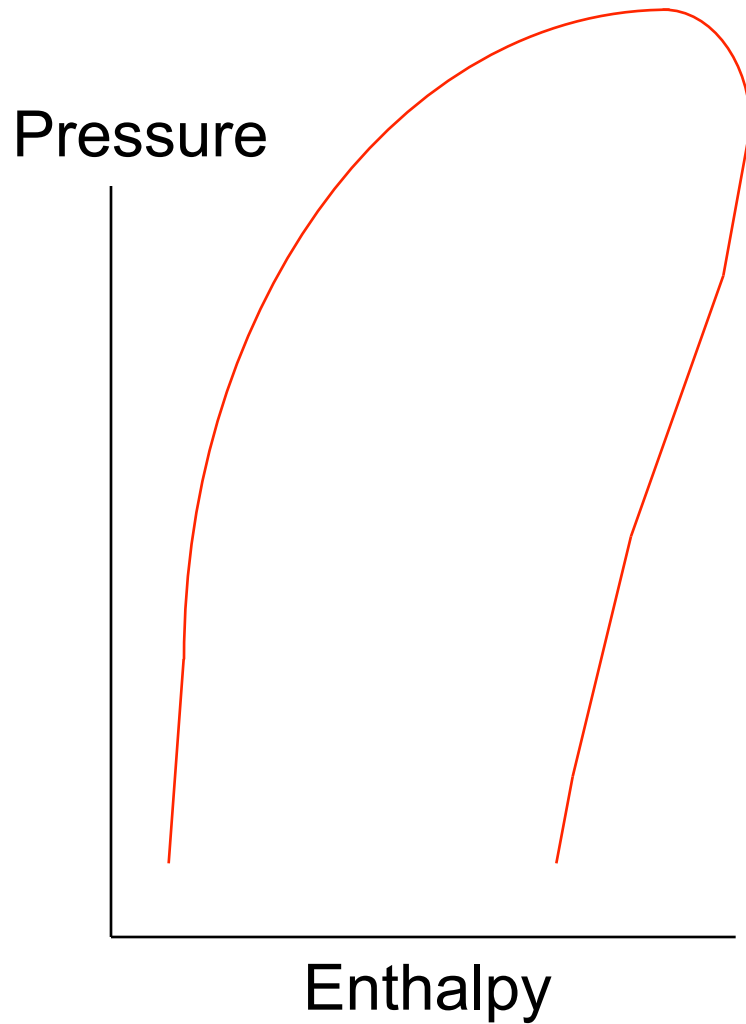
# Pressure - Enthalpy Chart

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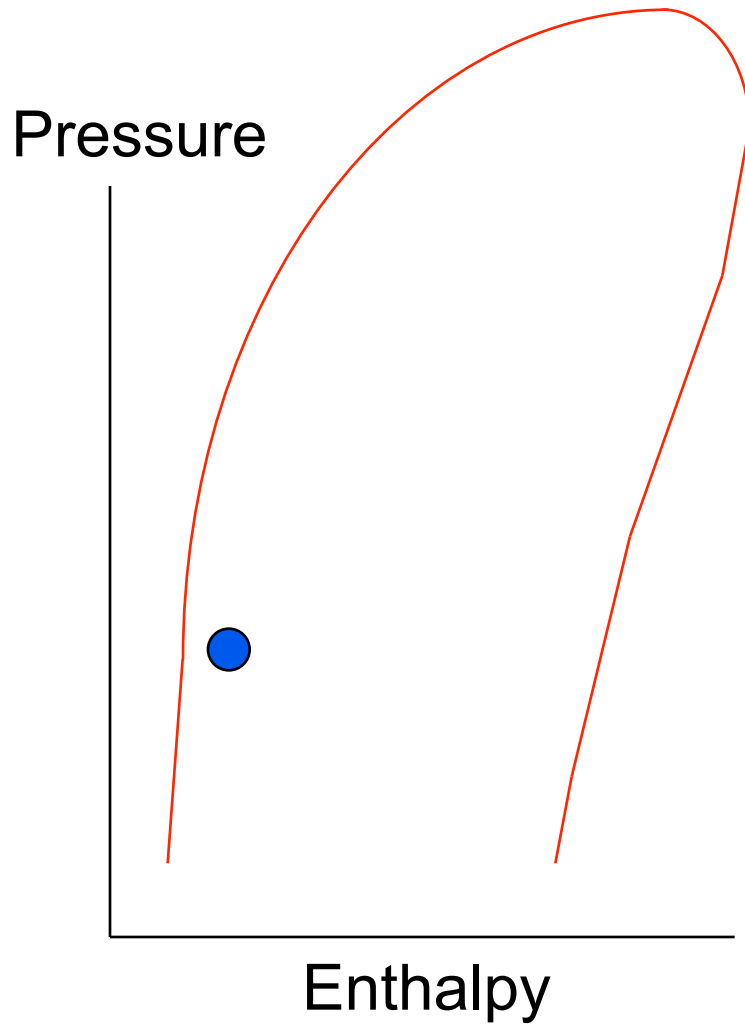
# Pressure - Enthalpy Chart

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# Pressure - Enthalpy Chart

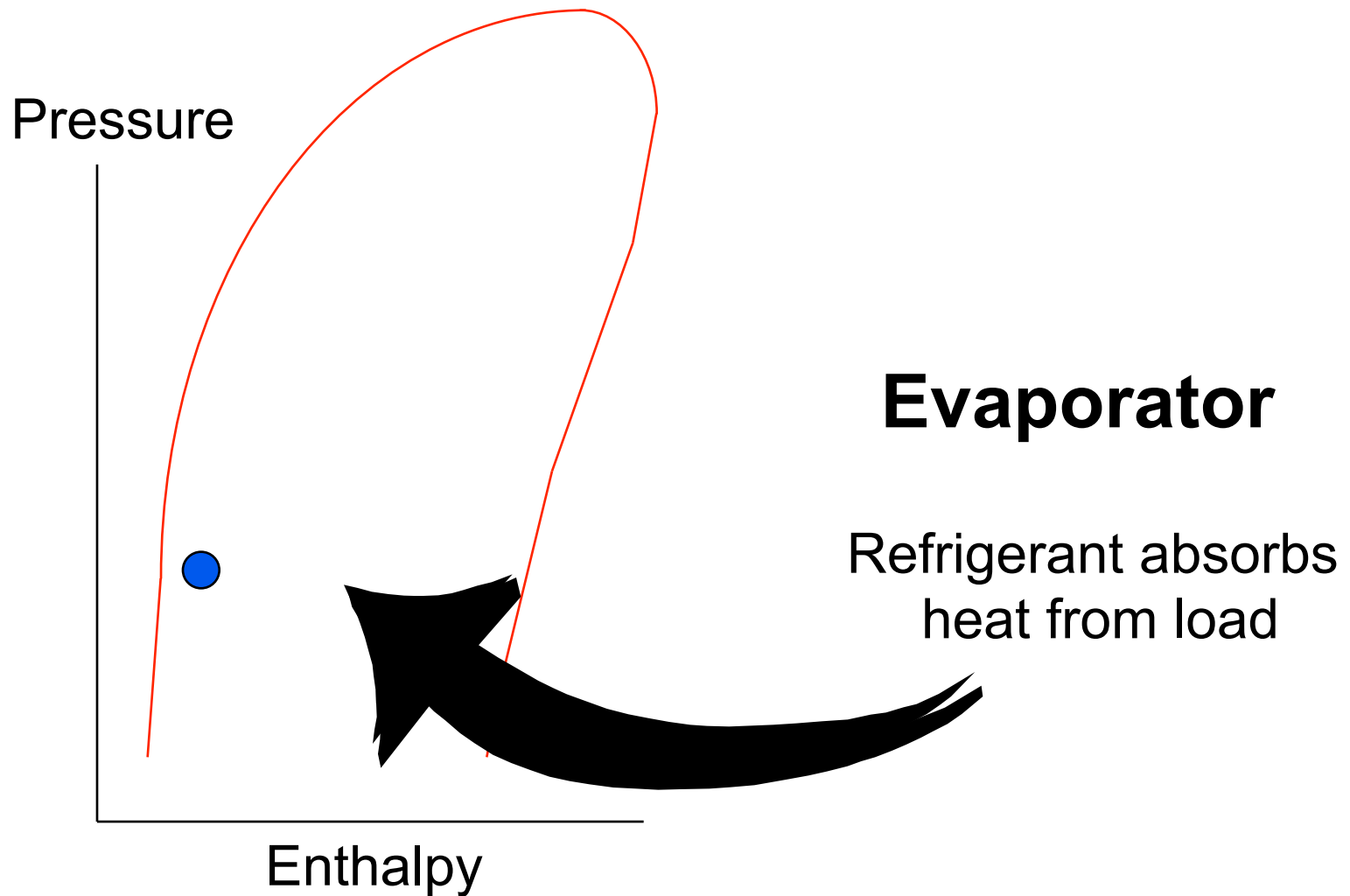
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**Evaporator**

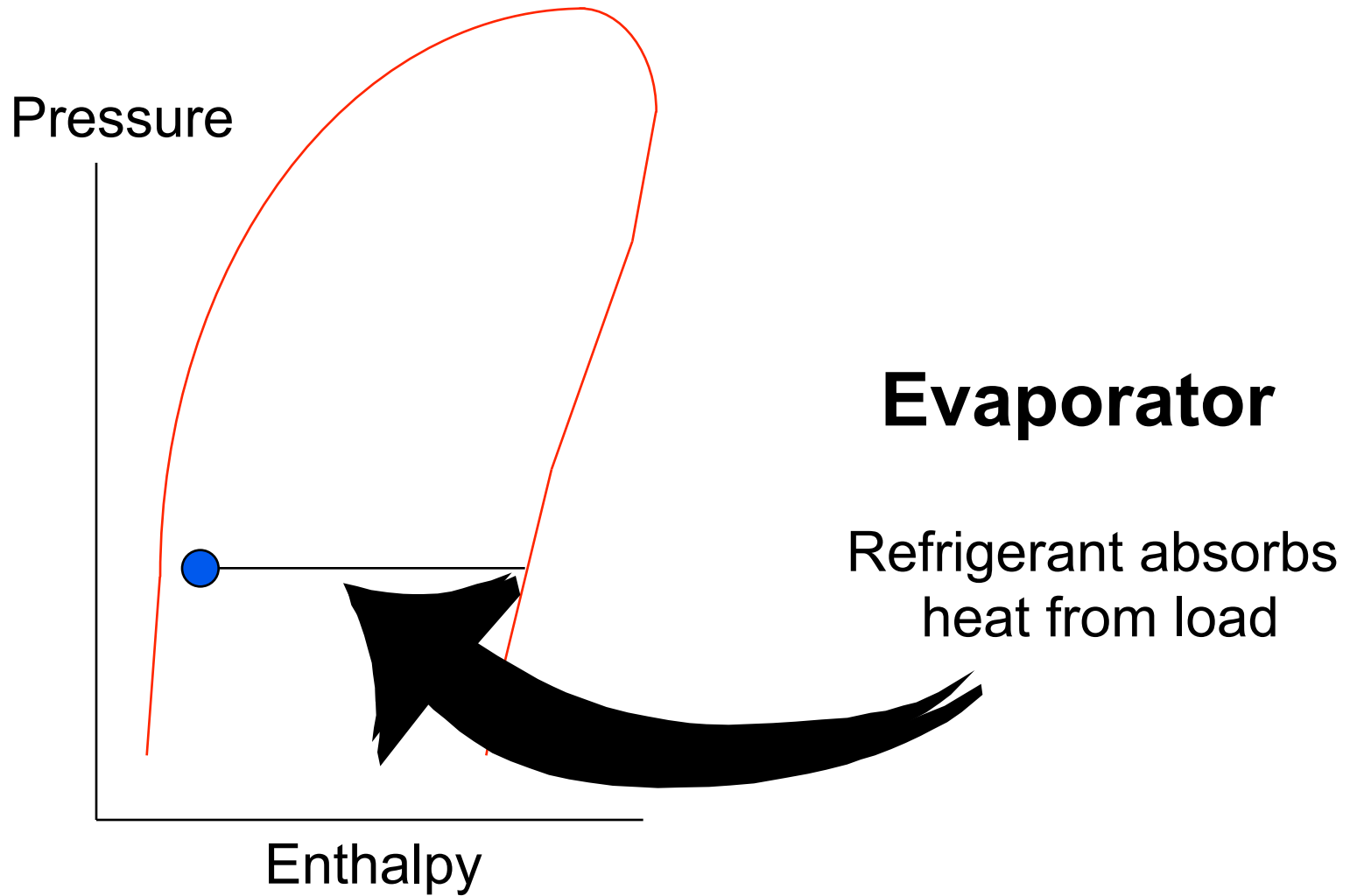
# Pressure - Enthalpy Chart

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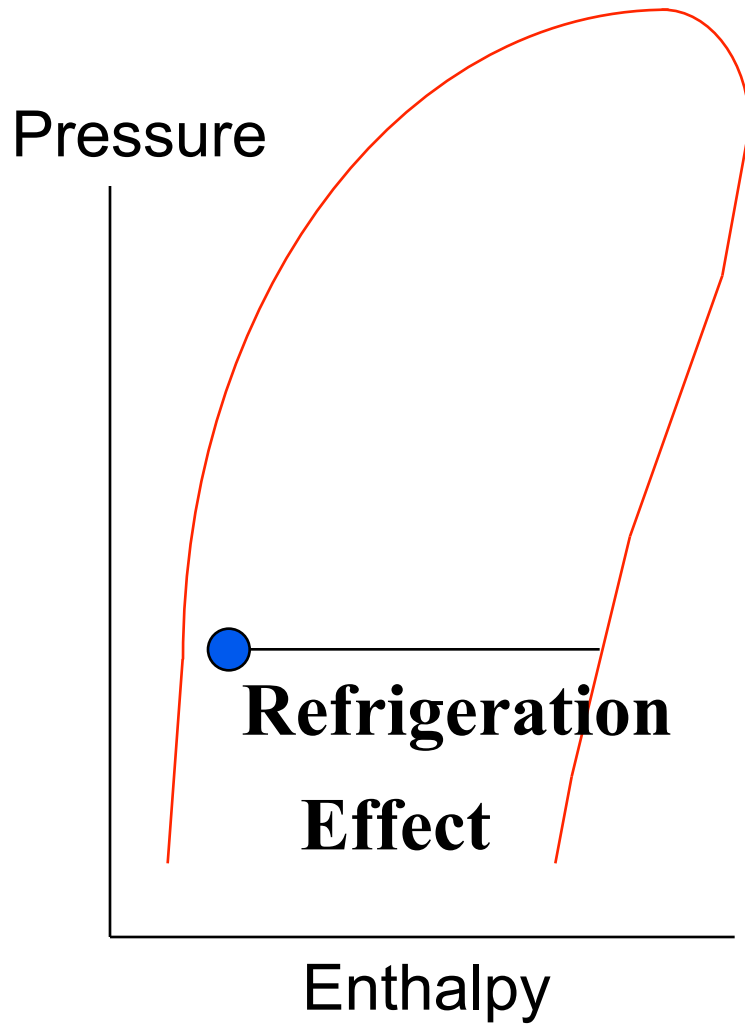
# Pressure - Enthalpy Chart

---



# Pressure - Enthalpy Chart

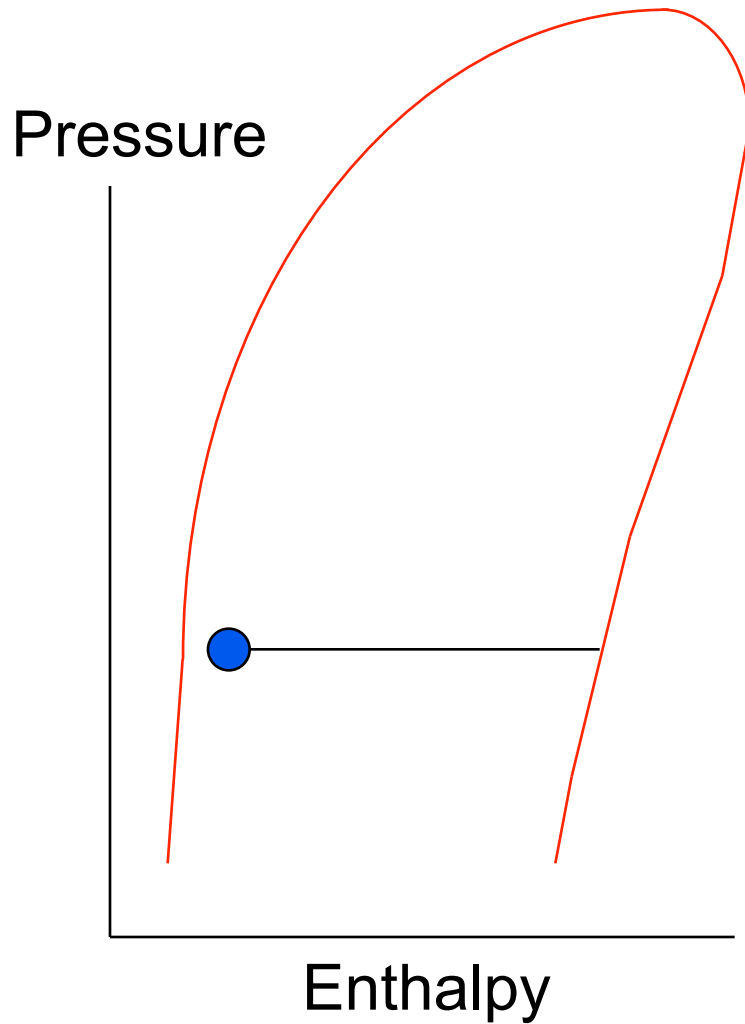
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**Evaporator**

# Pressure - Enthalpy Chart

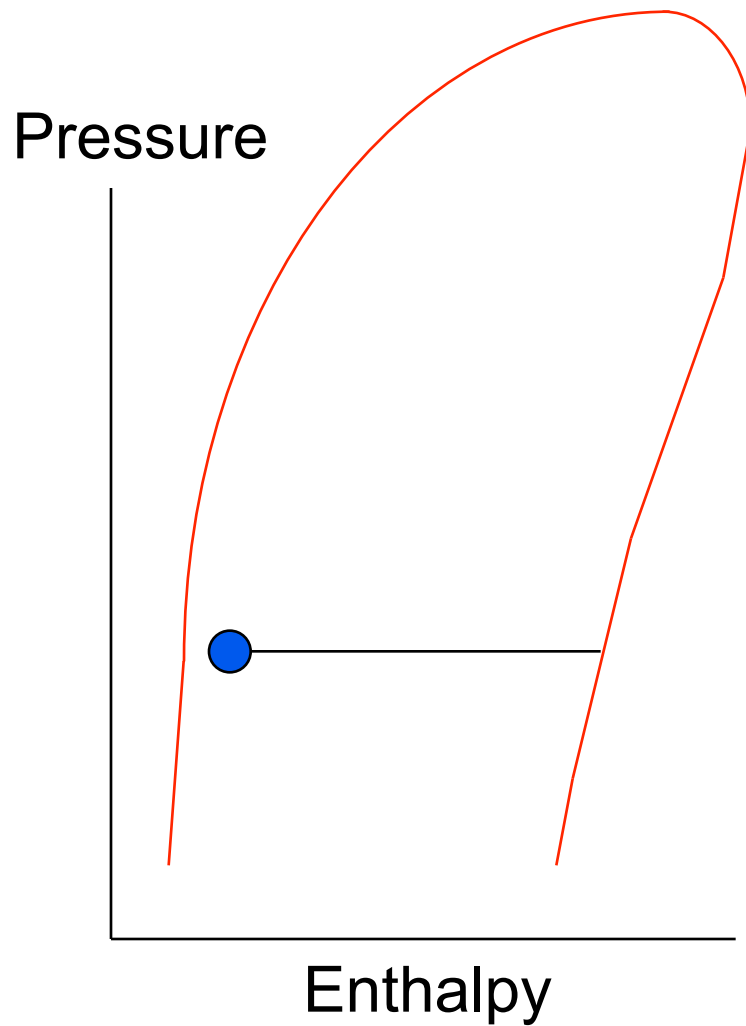
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**Evaporator**

# Pressure - Enthalpy Chart

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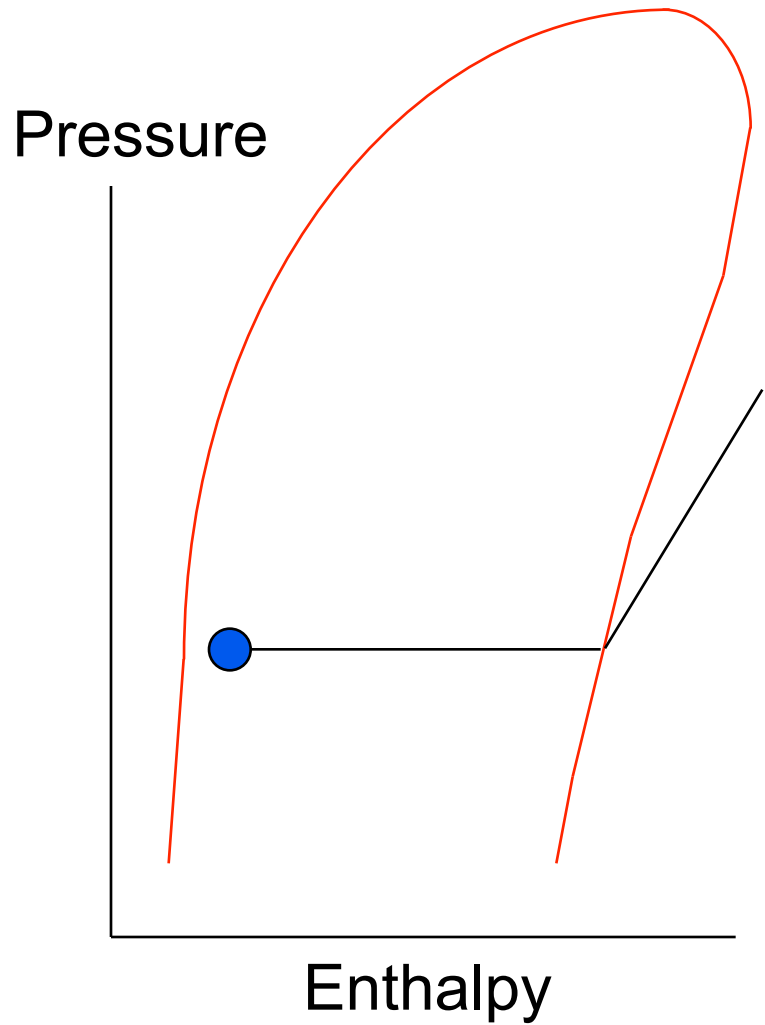


**Compressor**

Or Heat Input

# Pressure - Enthalpy Chart

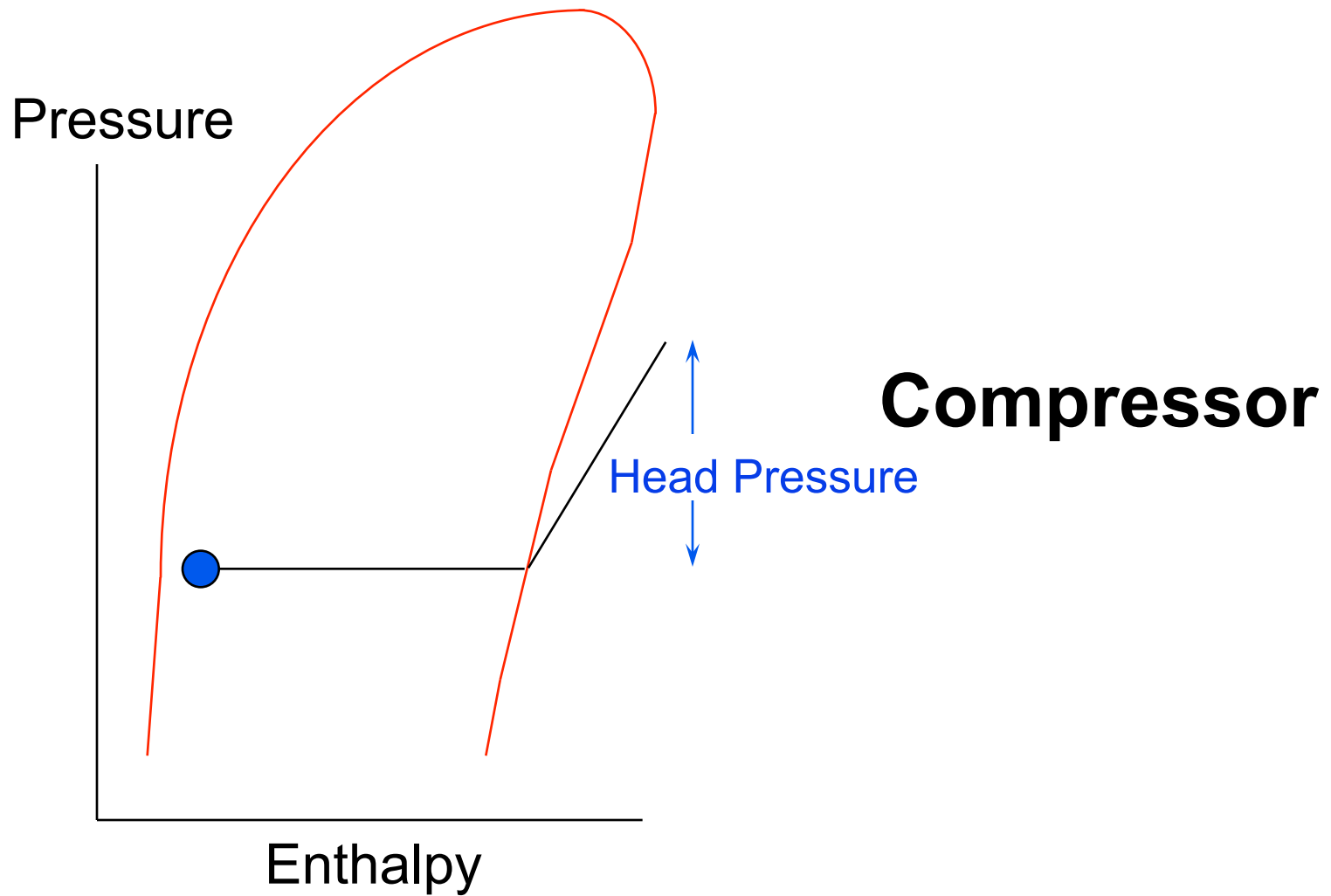
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**Compressor**

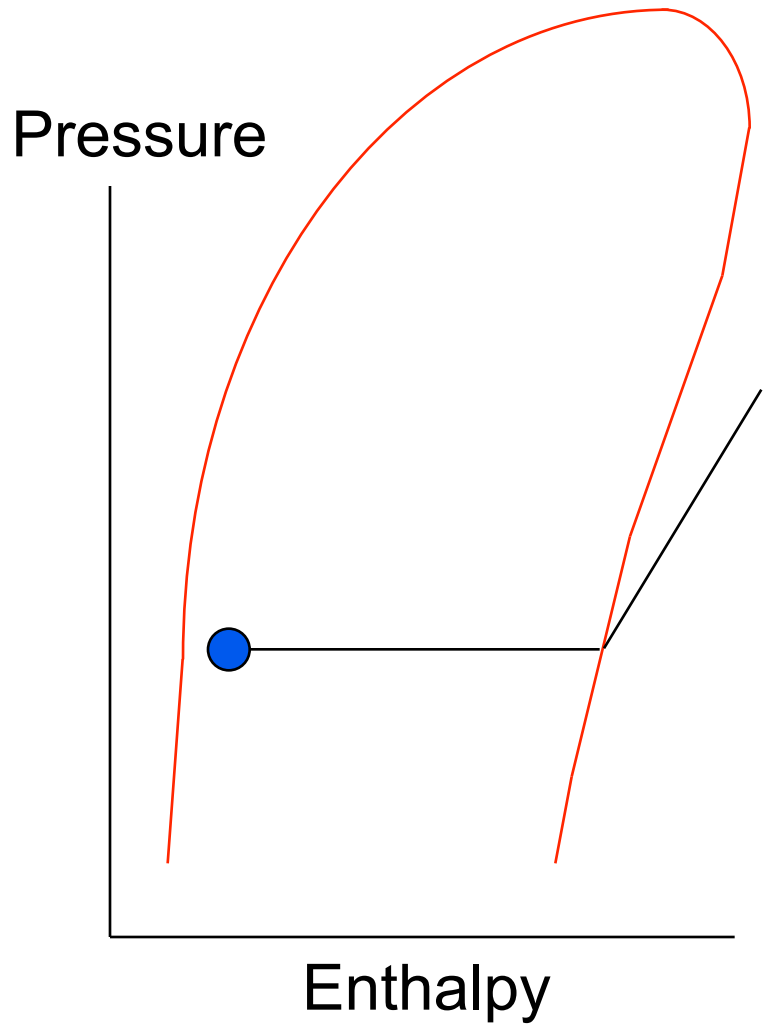
# Pressure - Enthalpy Chart

---



# Pressure - Enthalpy Chart

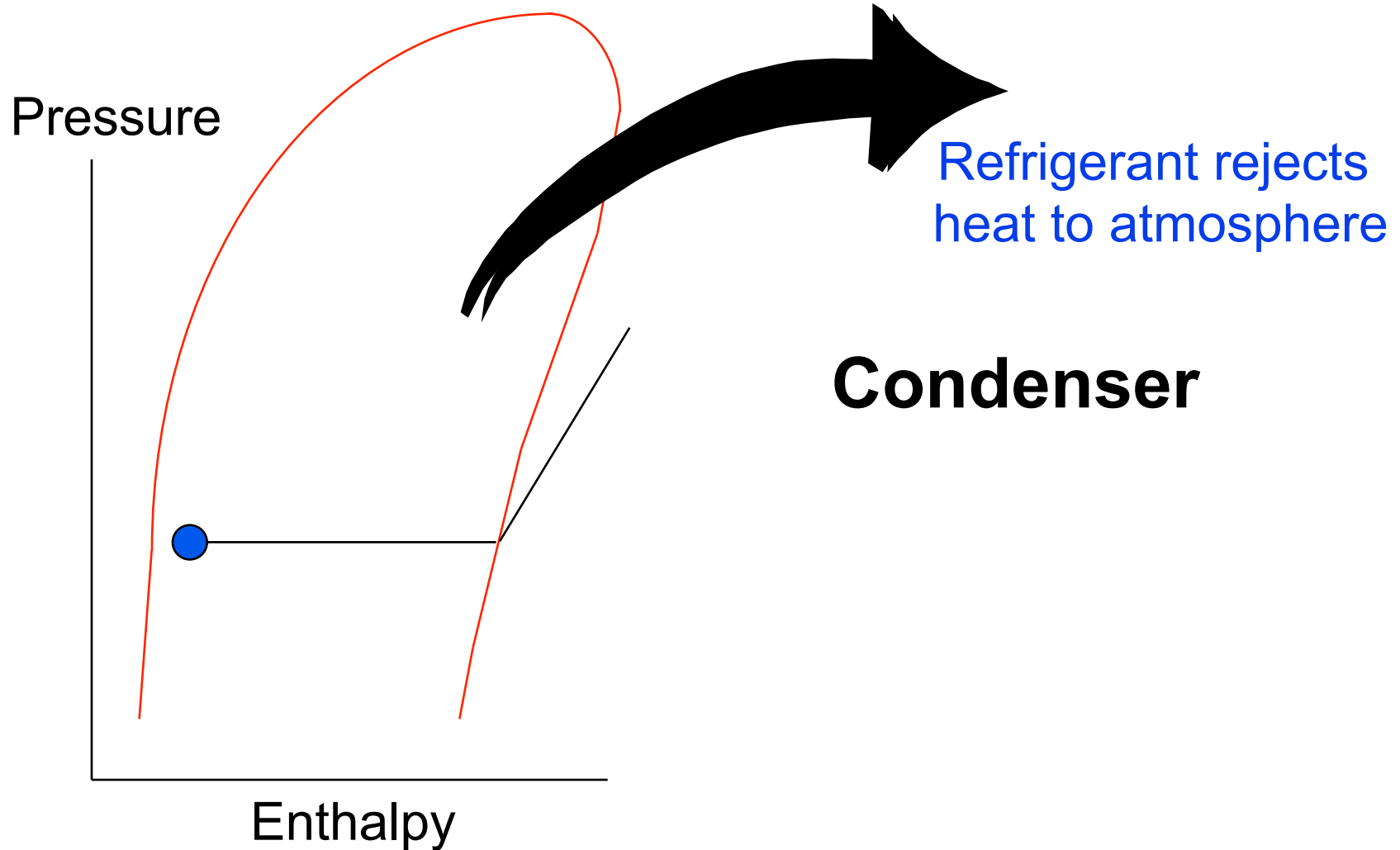
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**Condenser**

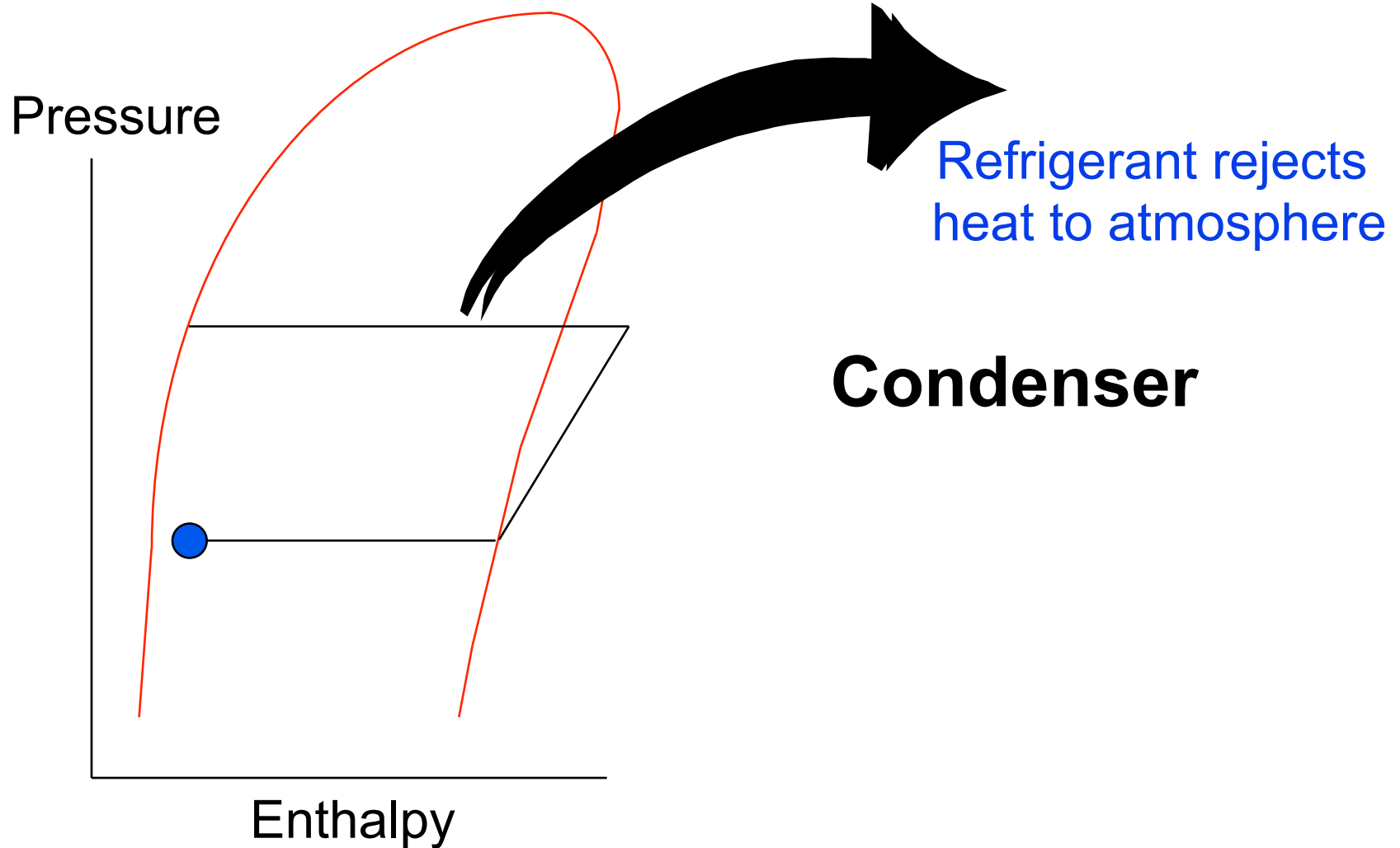
# Pressure - Enthalpy Chart

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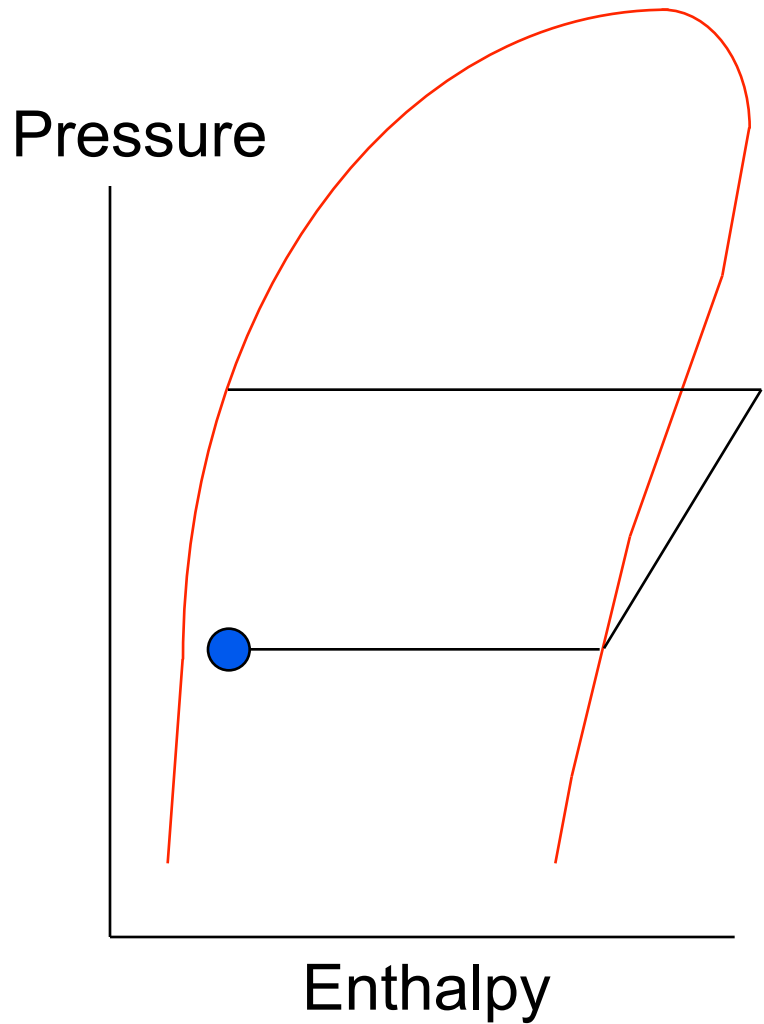
# Pressure - Enthalpy Chart

---



# Pressure - Enthalpy Chart

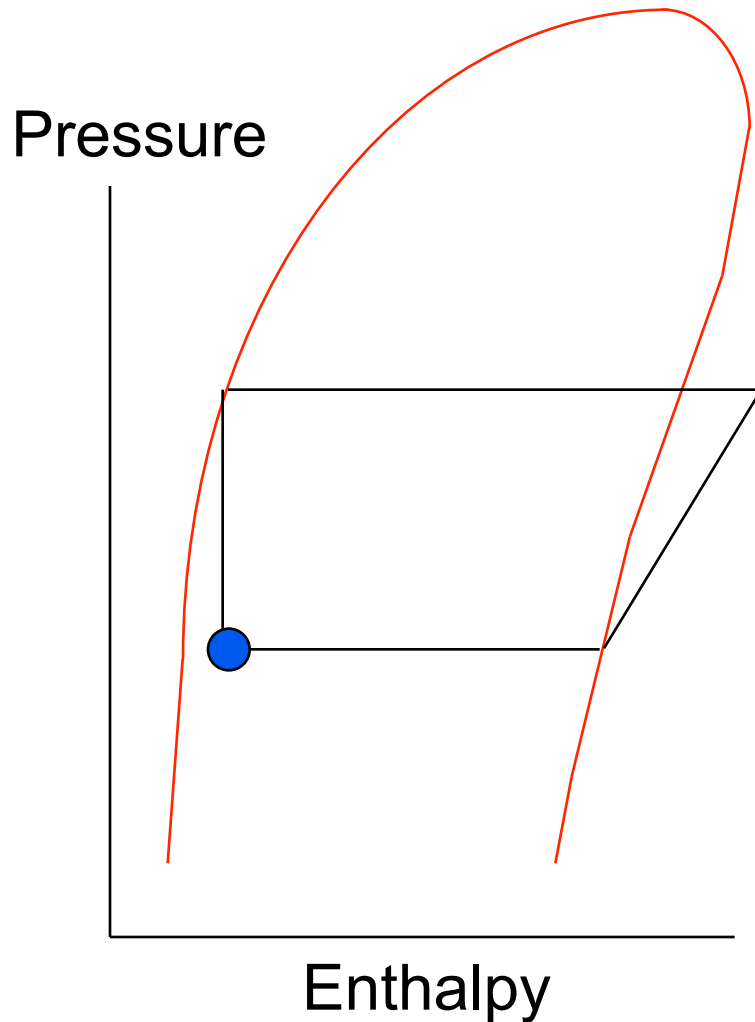
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**Metering Device**

# Pressure - Enthalpy Chart

---



## Metering Device

- Thermal expansion valve

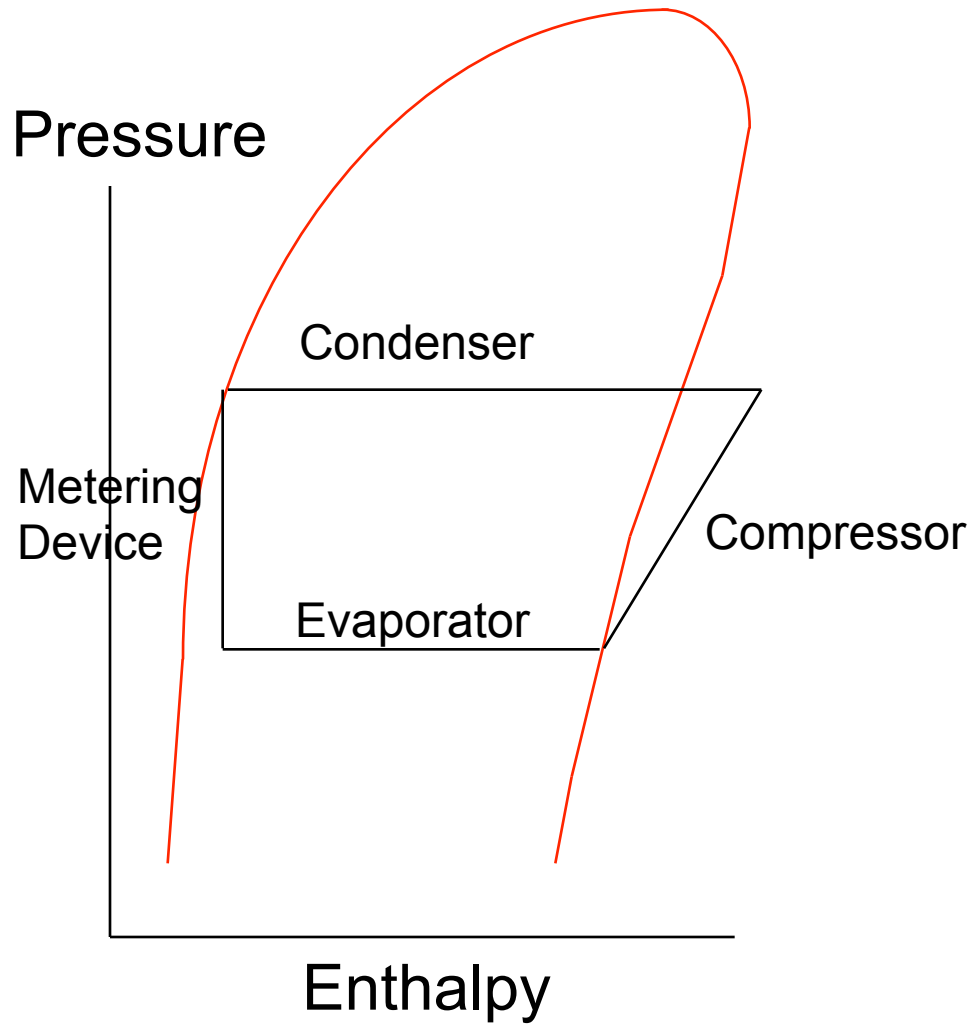
In refrigeration gas units

- Orifice Device

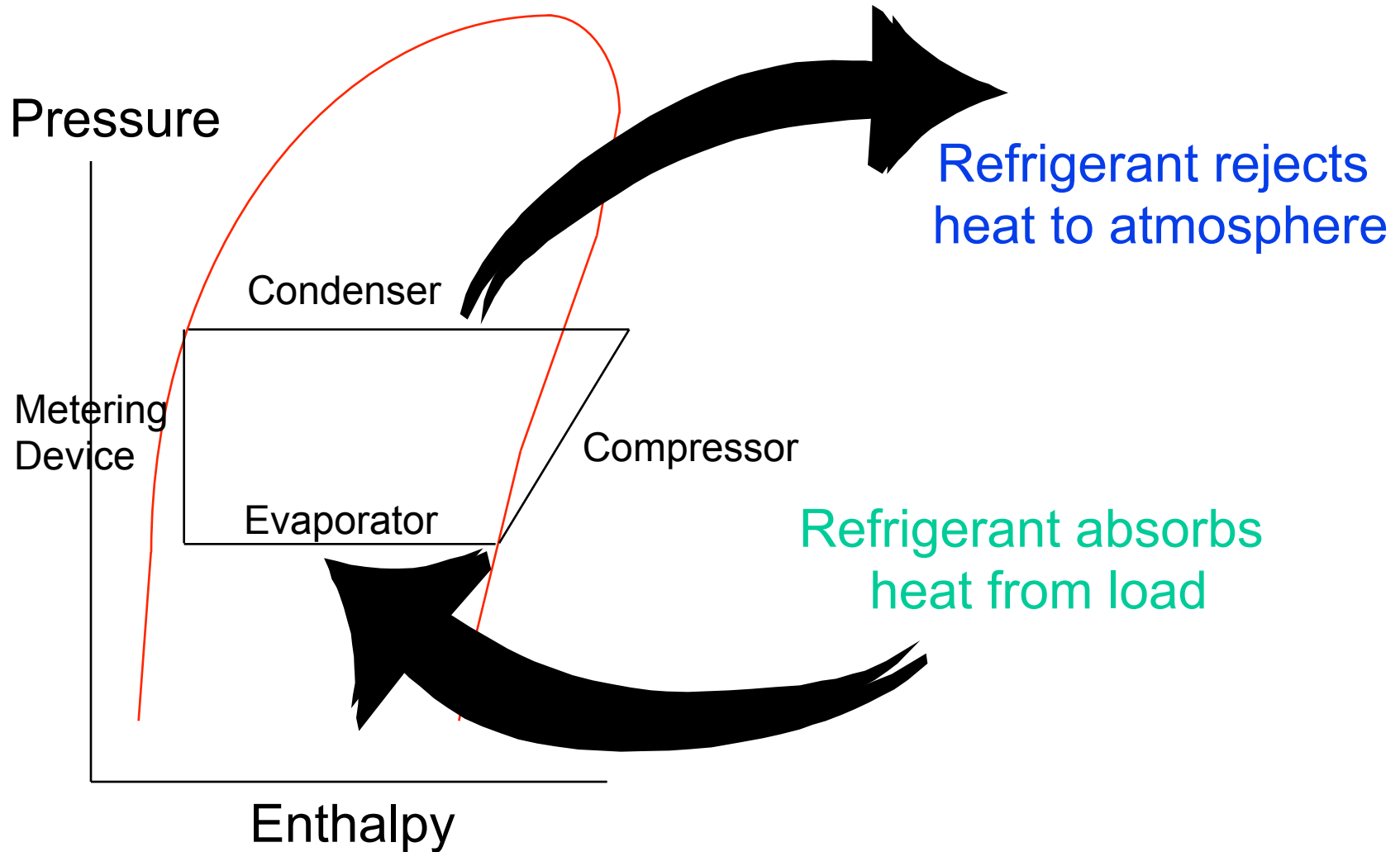
In Absorption Units

# Pressure - Enthalpy Chart

---

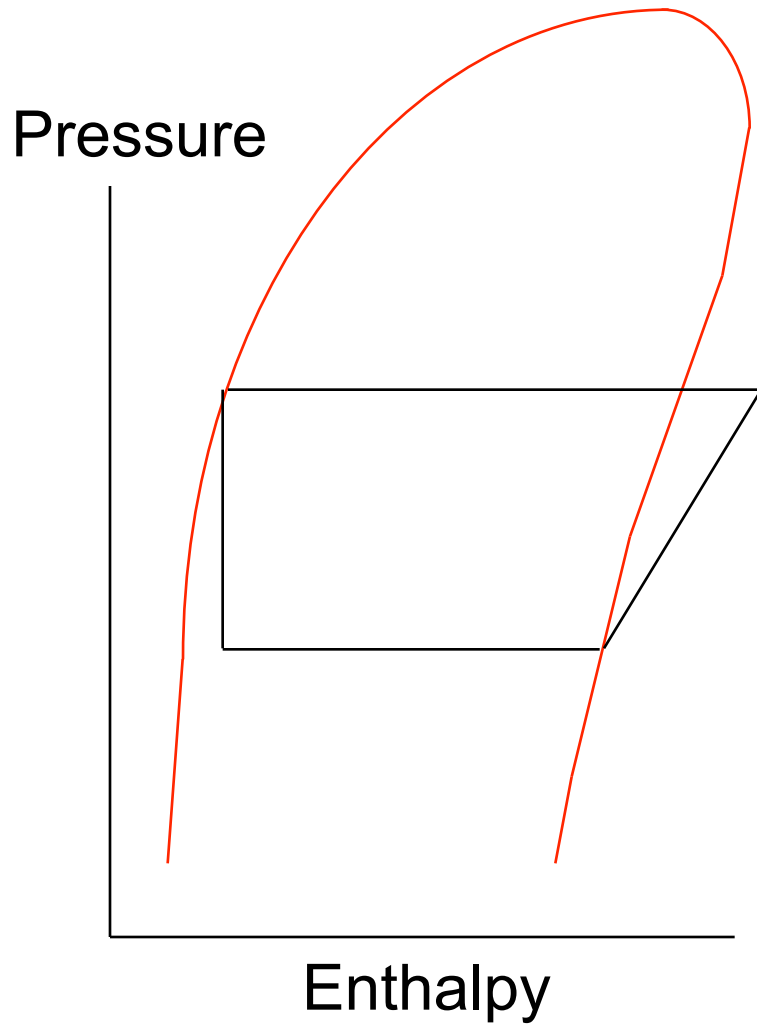


# Pressure - Enthalpy Chart



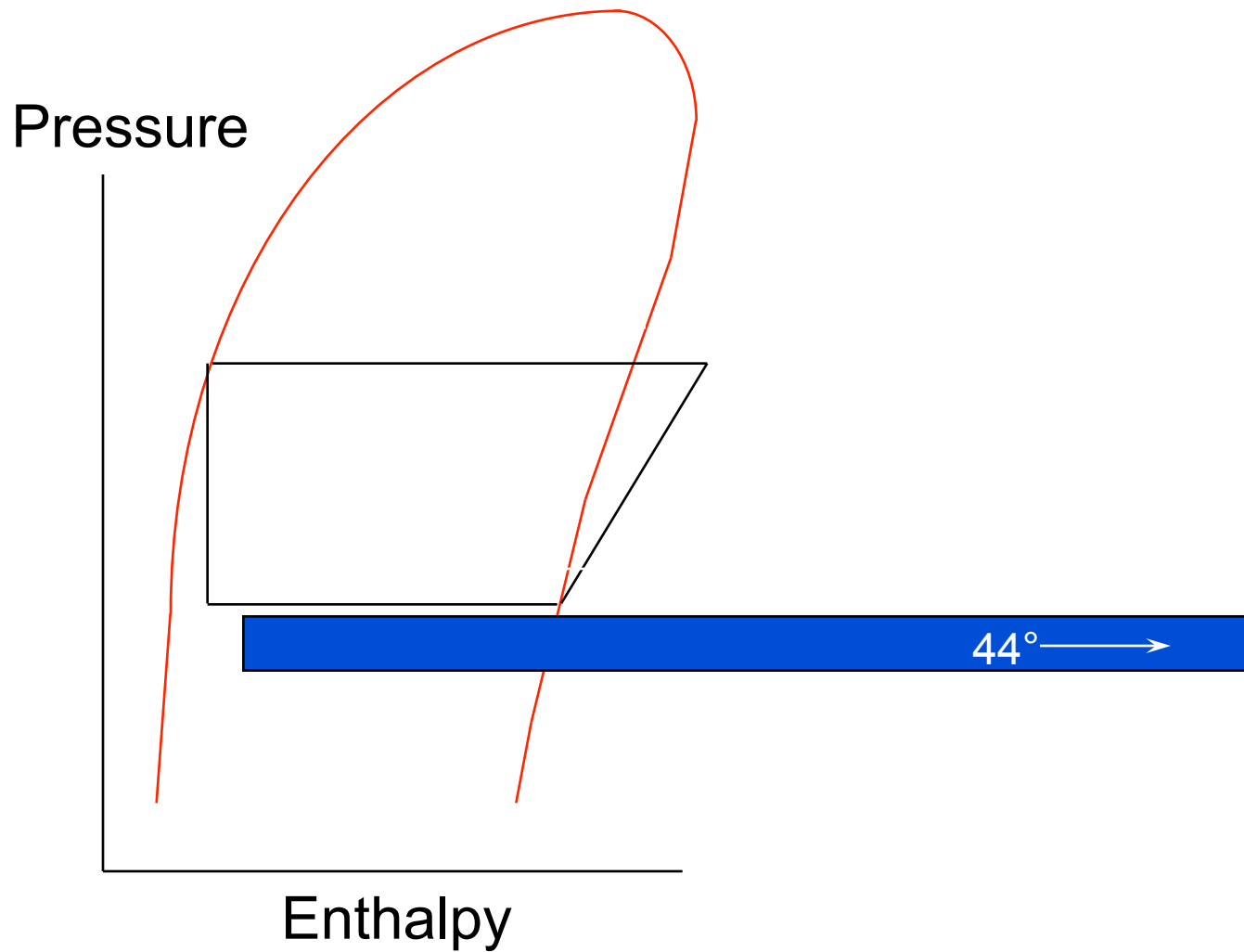
# Pressure - Enthalpy Chart

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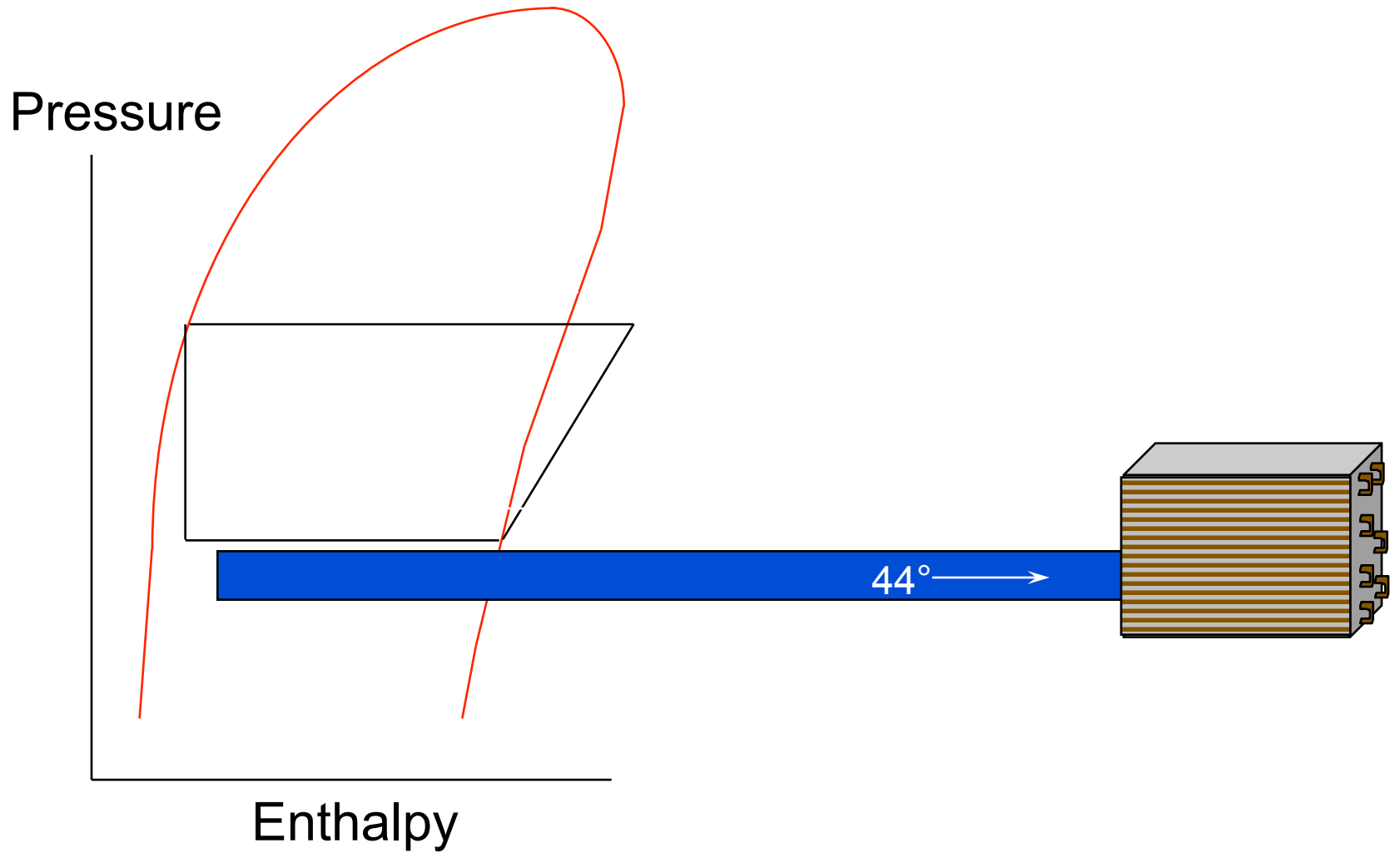


# Pressure - Enthalpy Chart

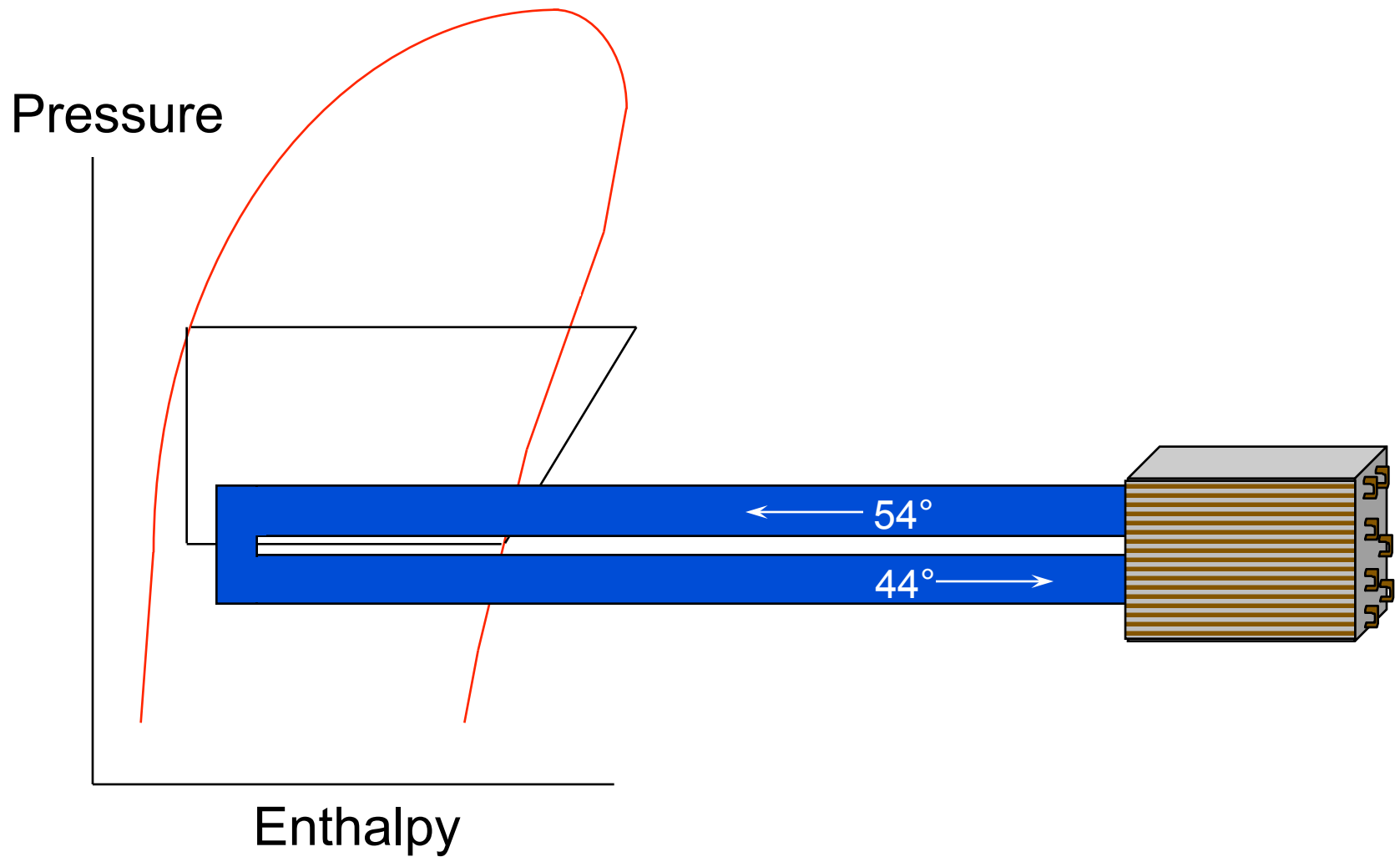
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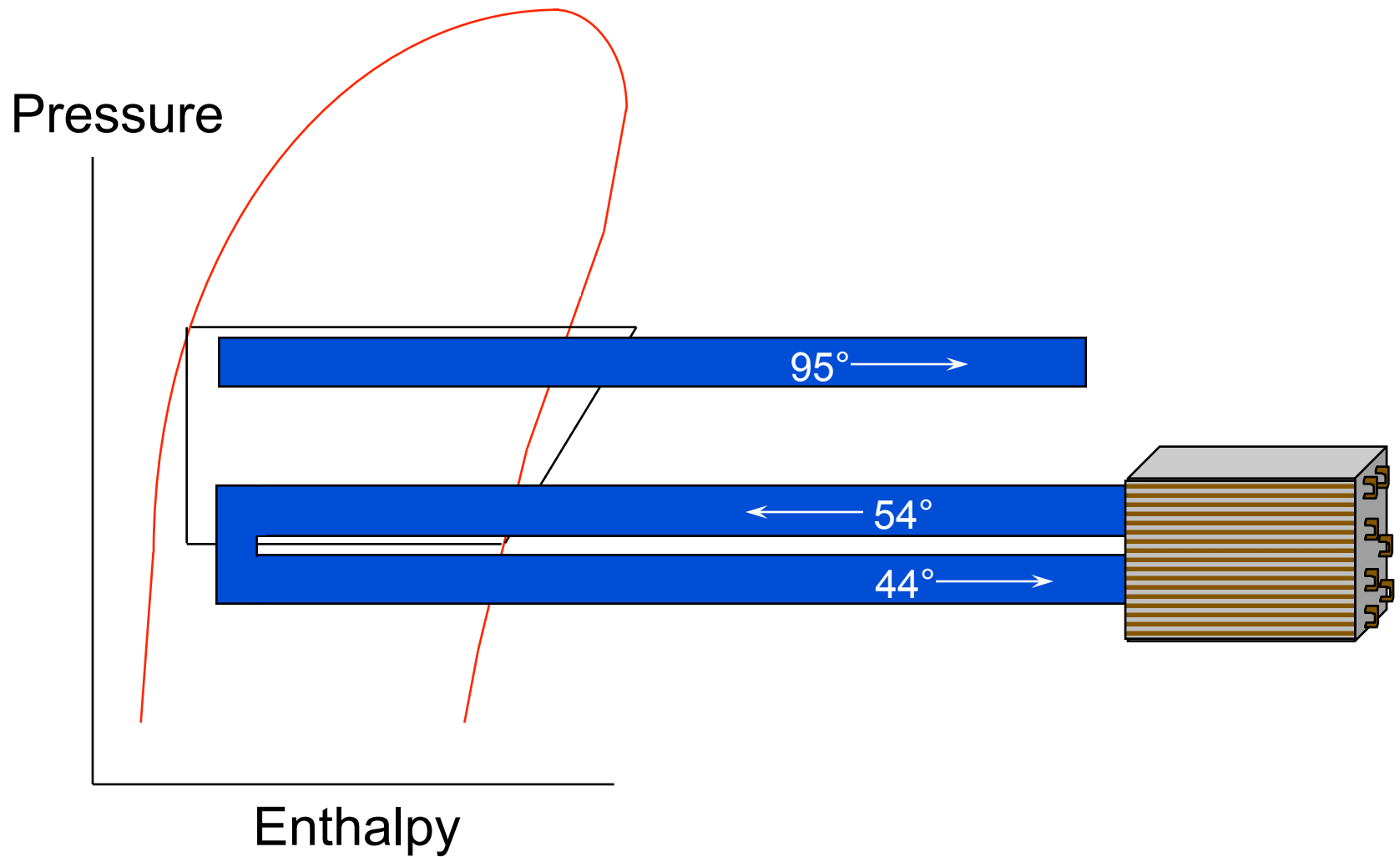
# Pressure - Enthalpy Chart



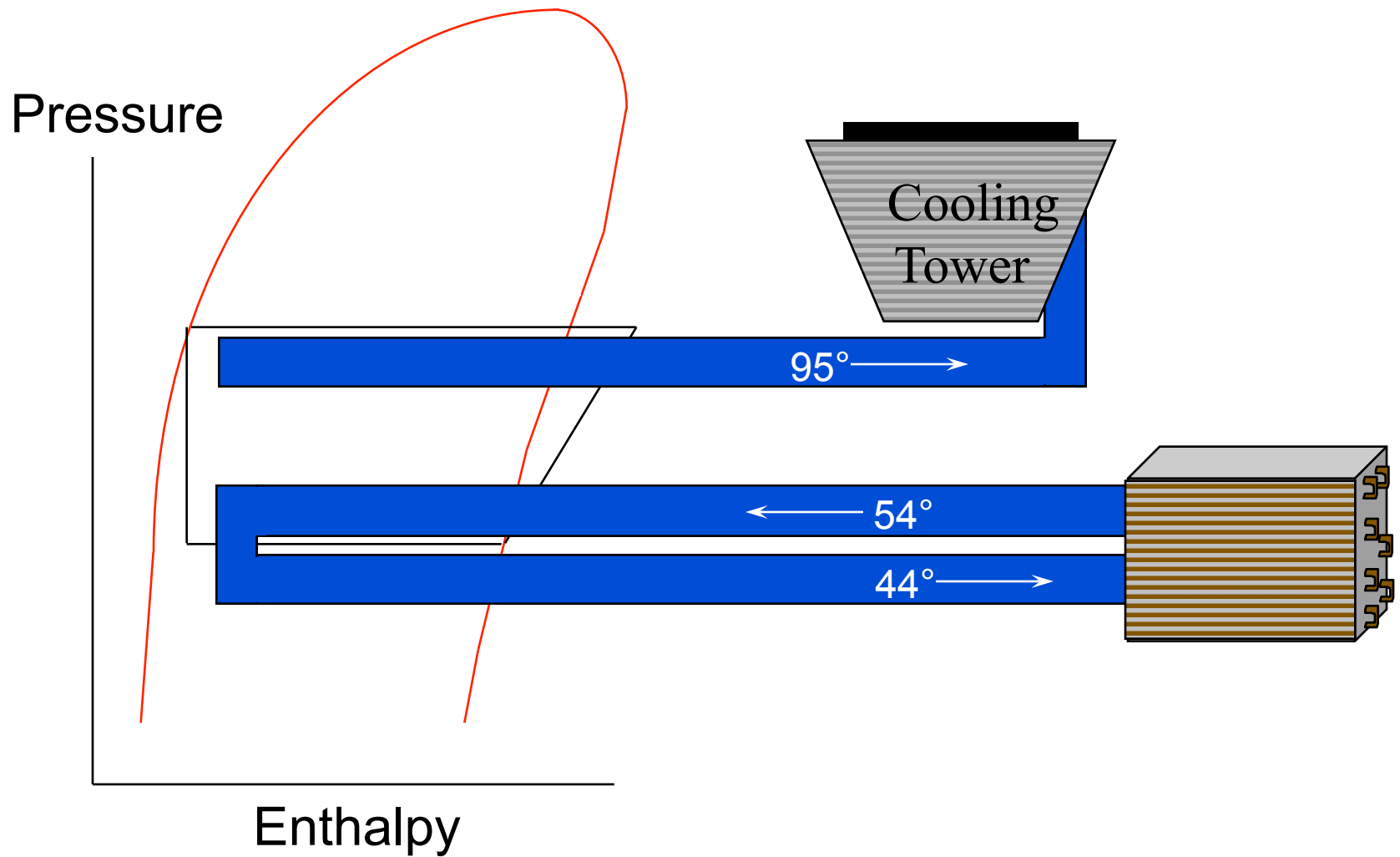
# Pressure - Enthalpy Chart



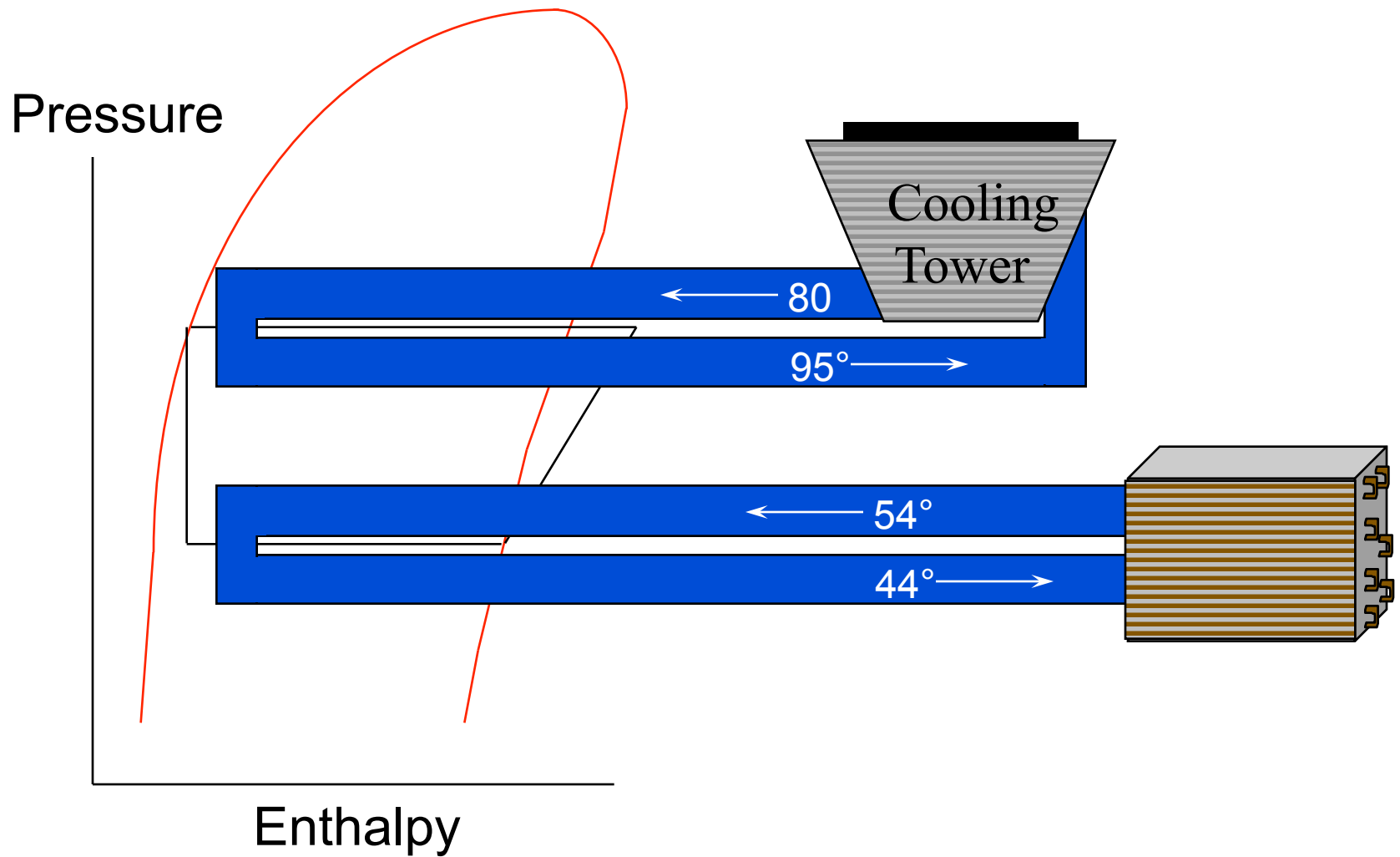
# Pressure - Enthalpy Chart



# Pressure - Enthalpy Chart

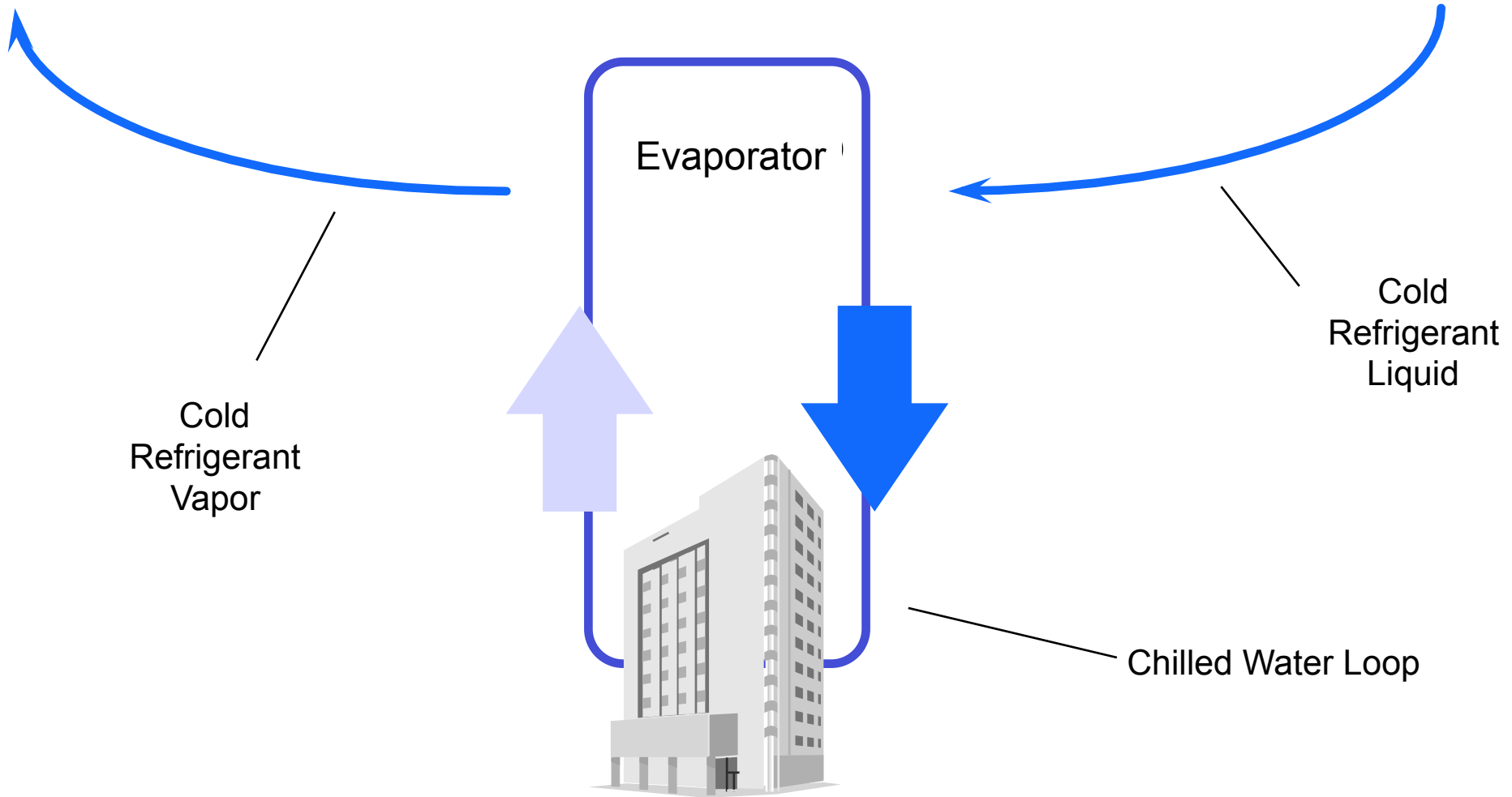


# Pressure - Enthalpy Chart

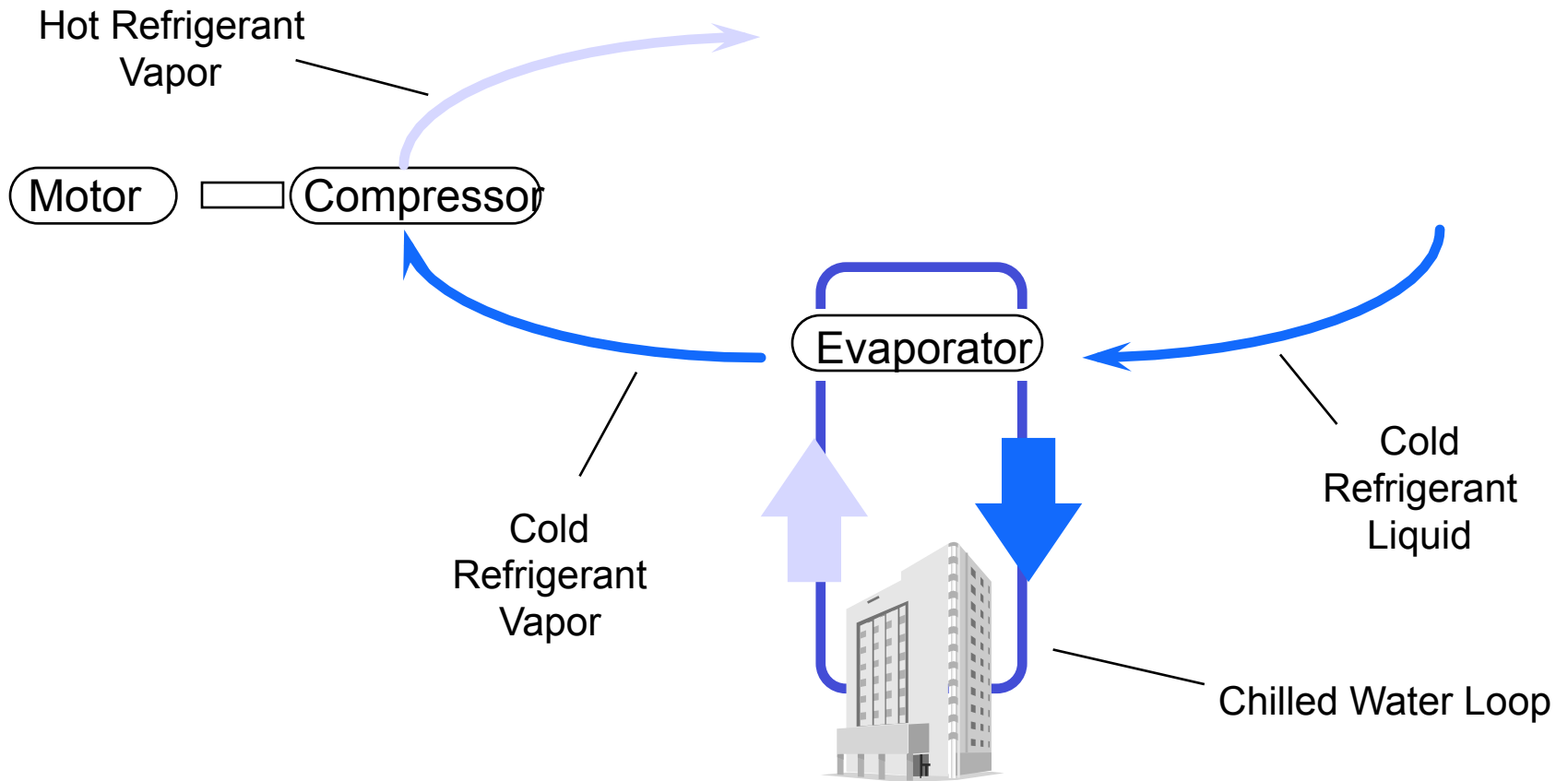


Now for something  
completely different

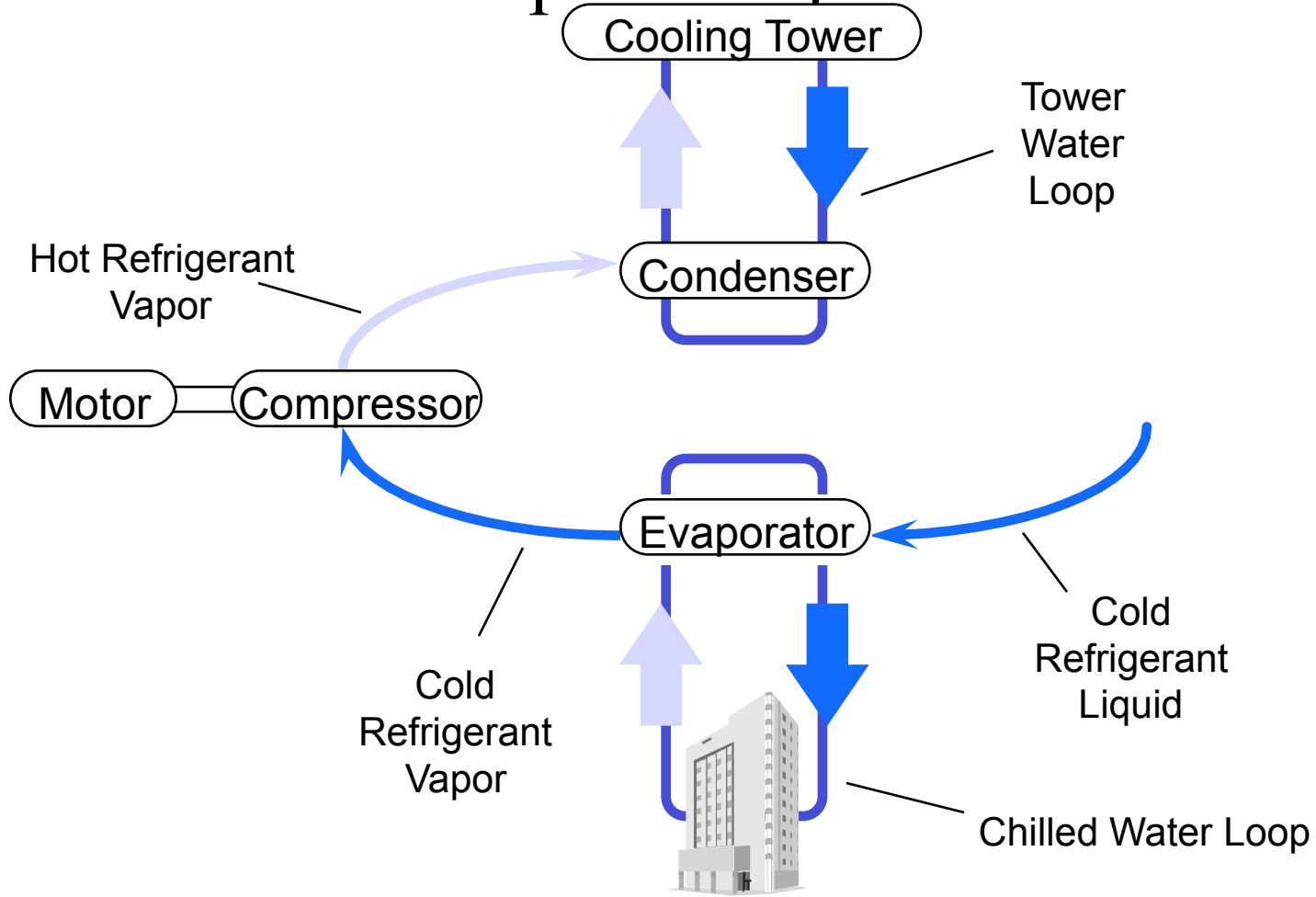
# Vapor Compression



# Vapor Compression



# Vapor Compression



# Pressure Conversions

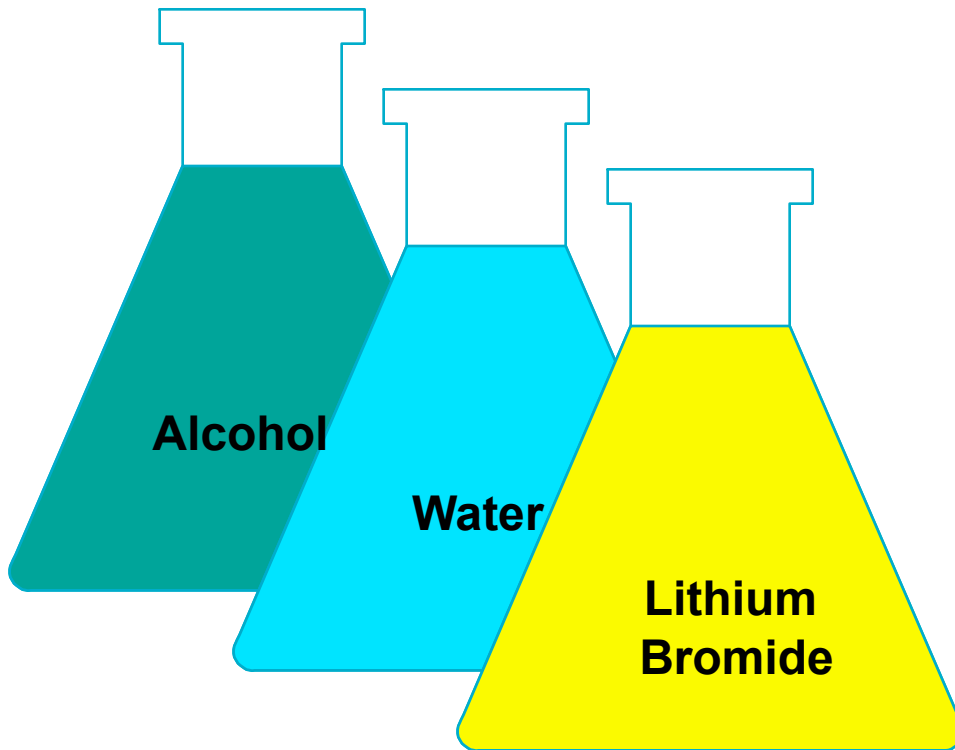
- Absolute Pressure = Gauge Pressure + Atmospheric Pressure
- Gauge Pressure = Absolute Pressure - Atmospheric Pressure
- Example:

Convert 8 PSIG to Absolute Pressure

$$P_{\text{Abs}} = P_{\text{gauge}} + P_{\text{atm}}$$

$$P_{\text{Abs}} = 8 \text{ PSIG} + 14.7 = 22.7 \text{ PSIA}$$

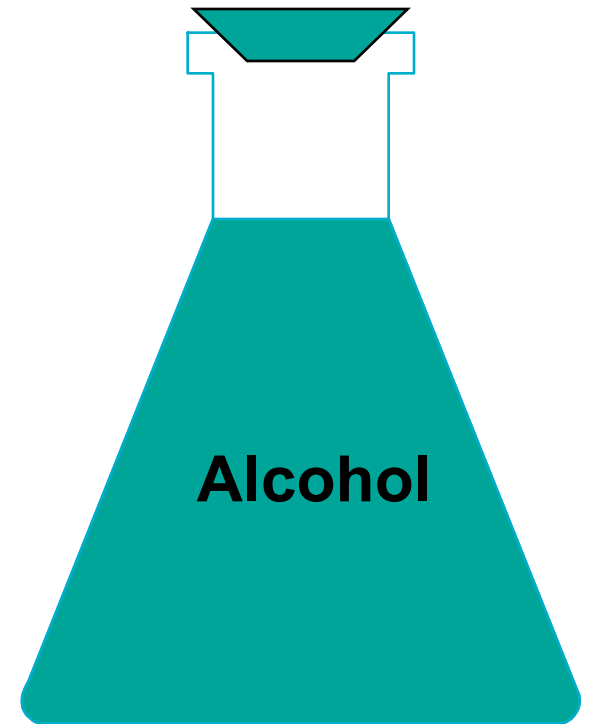
# Absorption System Fluids



- Alcohol
- Water
- Lithium Bromide

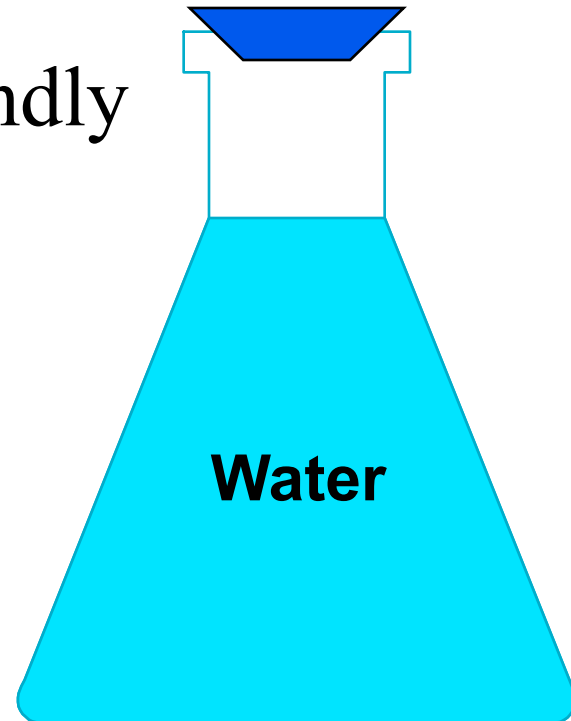
# Alcohol

- 2 - Ethyl Hexanol
- Alcohol is only 1% of the total charge in the system, but increases efficiency by 10%-15%
- Each model chiller has an optimum alcohol charge.
- Adding more alcohol will reduce efficiency



# Refrigerant is De-Ionized Water

- Stable
- Non-toxic
- Environmentally Friendly
- Low cost



# Latent Heat Values (Btu/Lb)

**Ammonia = 536**

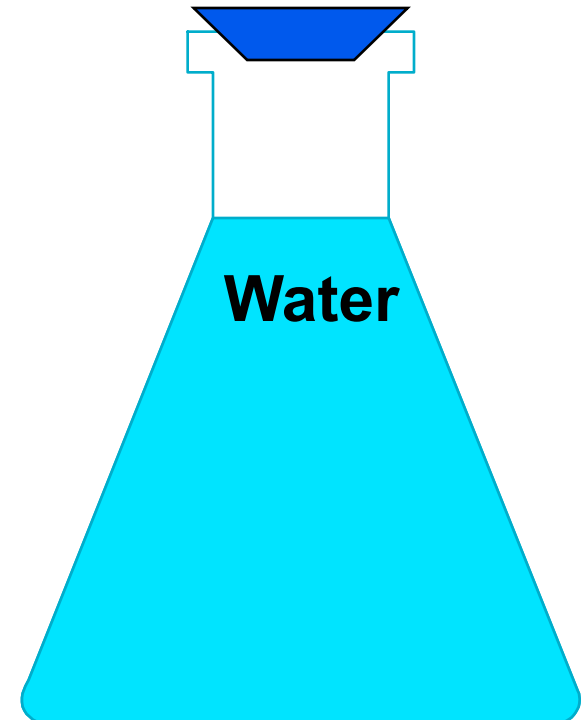
**R-22 = 86**

**R-123 = 77**

**Water = 1070**

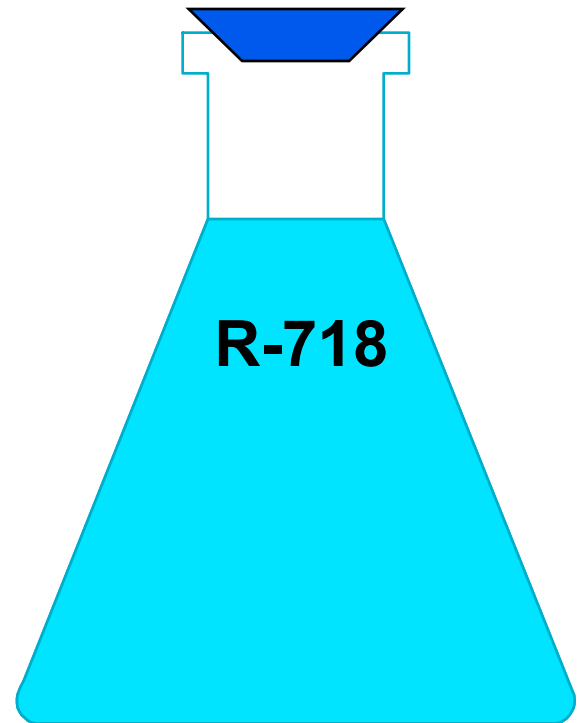
# Refrigerant is Water

- The refrigerant number for water is



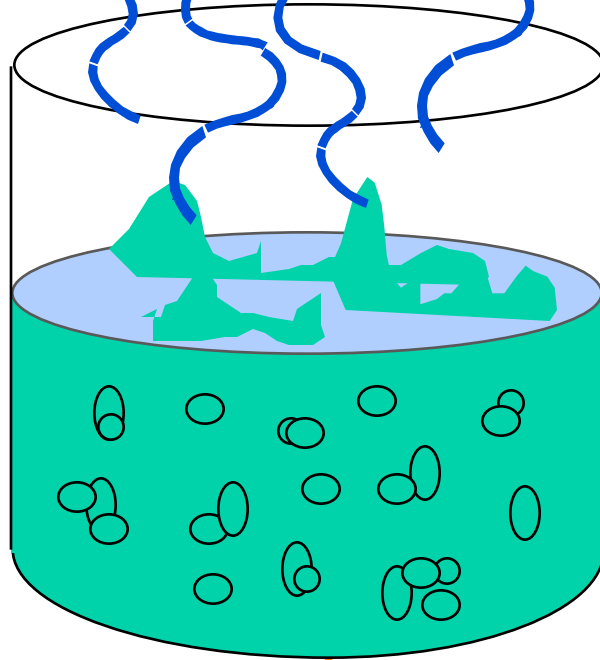
# Refrigerant is Water

- The refrigerant number for water is R- 718.



# Boiling Point of Water

**Atmospheric  
Pressure  
(14.696 psia)**



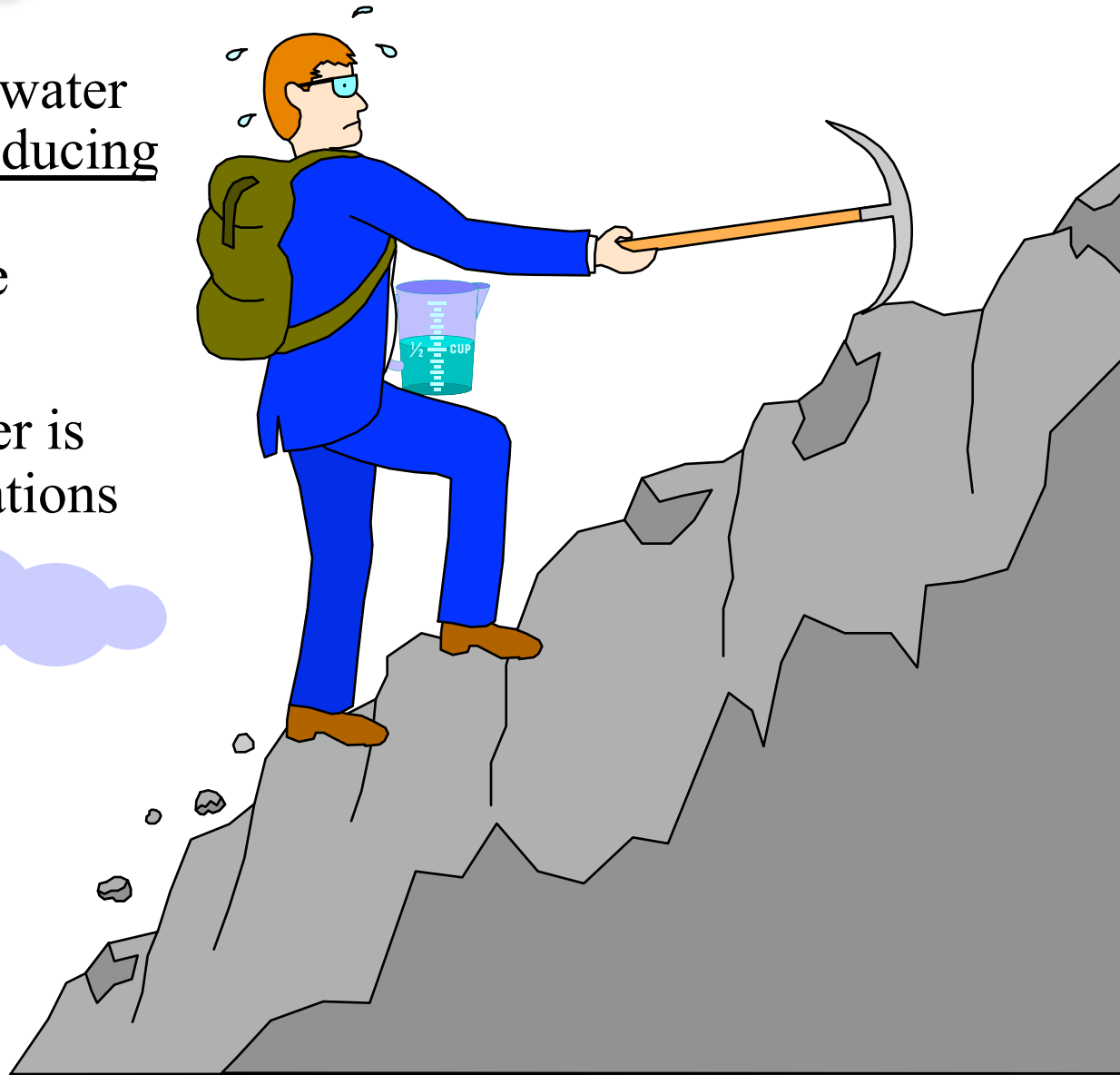
**Liquid turns to  
vapor**

**212°**

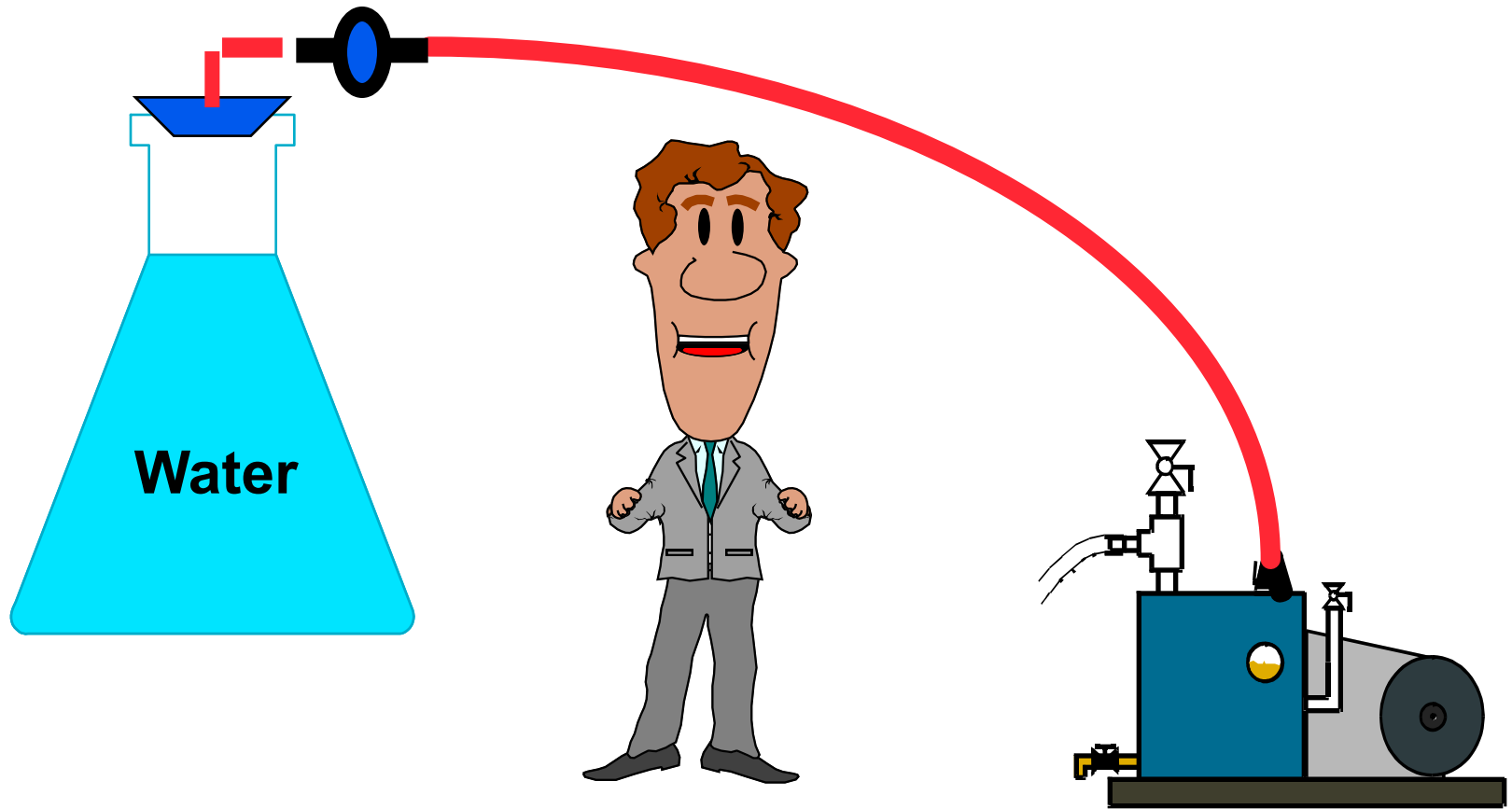
# Boiling Point of Water

The boiling point of water can be lowered by reducing the amount of air or atmospheric pressure

Boiling point of water is lower at higher elevations



By removing a portion of the air (atmospheric pressure) a P/T chart can be applied to water which indicates the boiling point of water at various pressures ...

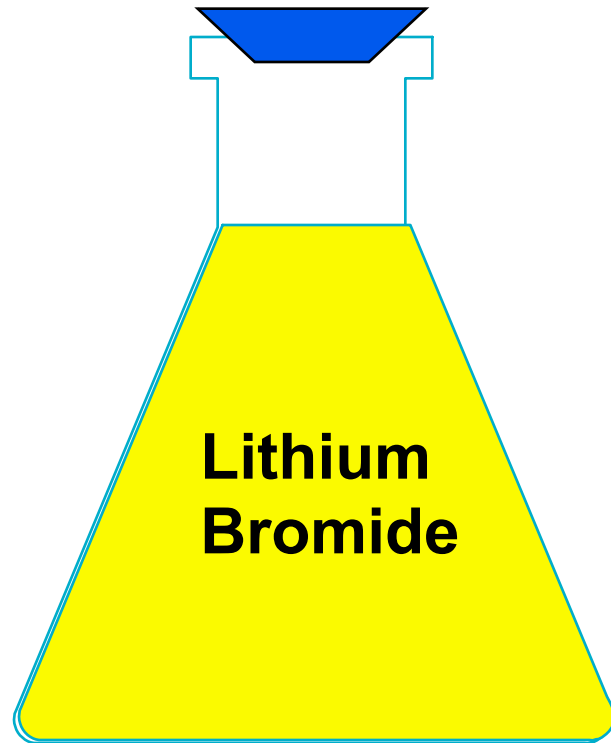


# Boiling Temperatures Of Water At Various Pressures

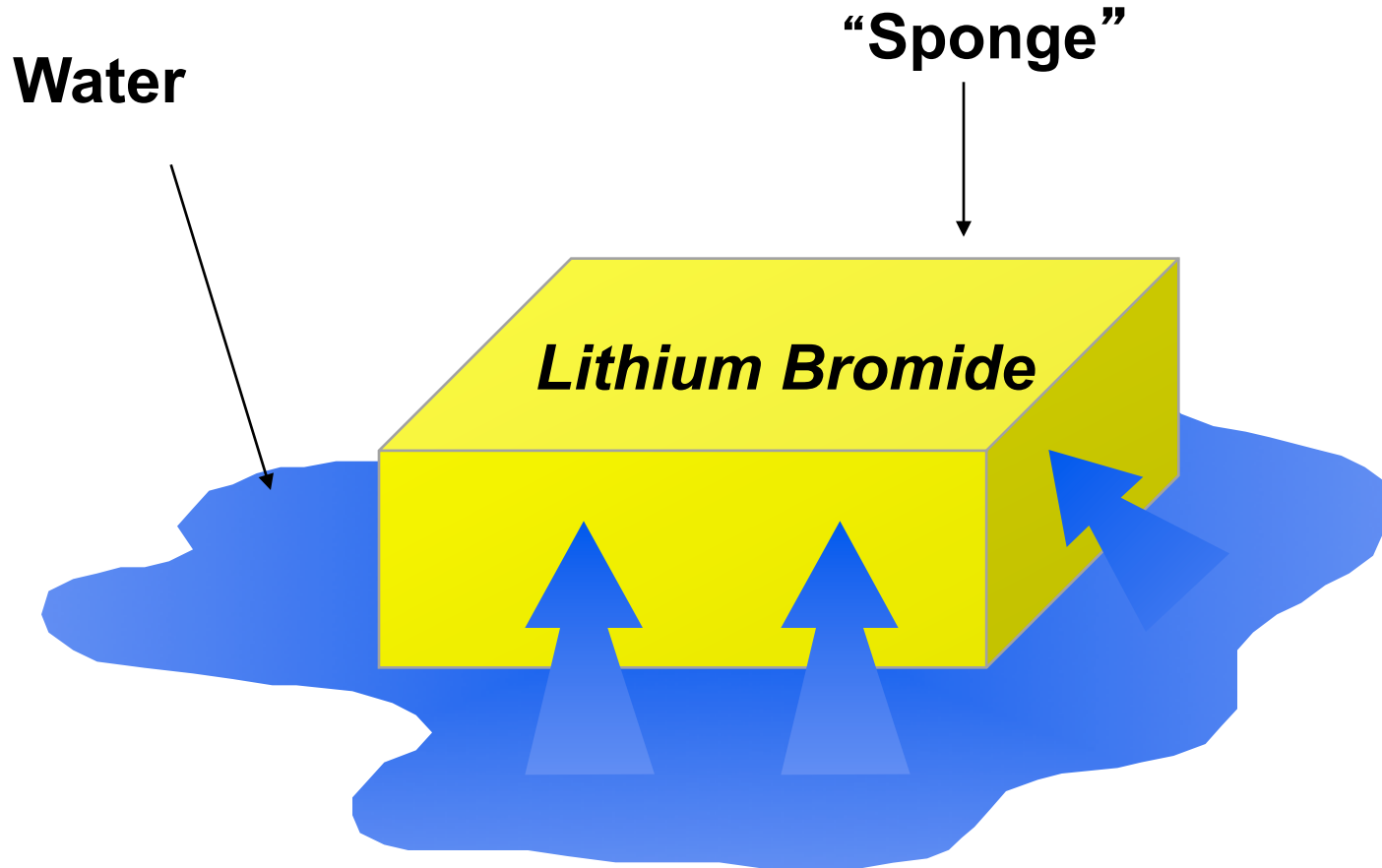
<i>Temps.</i> Deg. F	<i>Gauge Pressures</i> Pressure (ins. Hg.)	<i>Absolute Pressures</i>		
		Psia Lb/Sq. In.	Ins. Hg.	Mm. Hg.
<b>212</b>	<b>0</b>	<b>14.696</b>	<b>29.921</b>	<b>759.99</b>
<b>160</b>	<b>20.265</b>	<b>4.742</b>	<b>9.656</b>	<b>245.25</b>
<b>100</b>	<b>27.988</b>	<b>.950</b>	<b>1.933</b>	<b>49.11</b>
<b>80</b>	<b>28.888</b>	<b>.507</b>	<b>1.032</b>	<b>26.22</b>
<b>70</b>	<b>29.182</b>	<b>.363</b>	<b>.739</b>	<b>18.77</b>
<b>60</b>	<b>29.399</b>	<b>.256</b>	<b>.522</b>	<b>13.25</b>
<b>50</b>	<b>29.599</b>	<b>.178</b>	<b>.362</b>	<b>9.20</b>
<b>45</b>	<b>29.621</b>	<b>.147</b>	<b>.300</b>	<b>7.63</b>
<b>40</b>	<b>29.673</b>	<b>.122</b>	<b>.248</b>	<b>6.29</b>
<b>35</b>	<b>29.718</b>	<b>.100</b>	<b>.203</b>	<b>5.17</b>
<b>30</b>	<b>29.757</b>	<b>.081</b>	<b>.165</b>	<b>4.18</b>
<b>20</b>	<b>29.818</b>	<b>.050</b>	<b>.103</b>	<b>2.61</b>

# Absorbent is Lithium Bromide

- Water absorbs readily into it
- High boiling point
- Non-toxic
- Relatively low cost

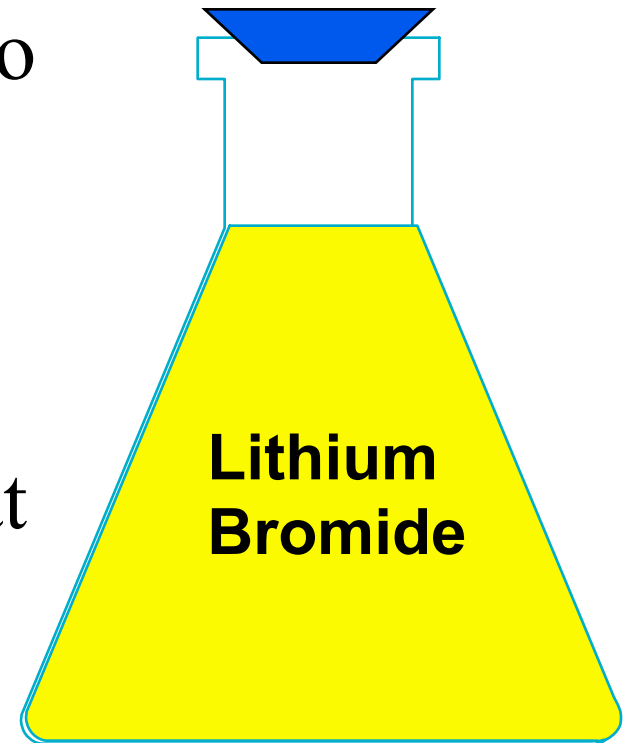


# Properties of Lithium Bromide and Water



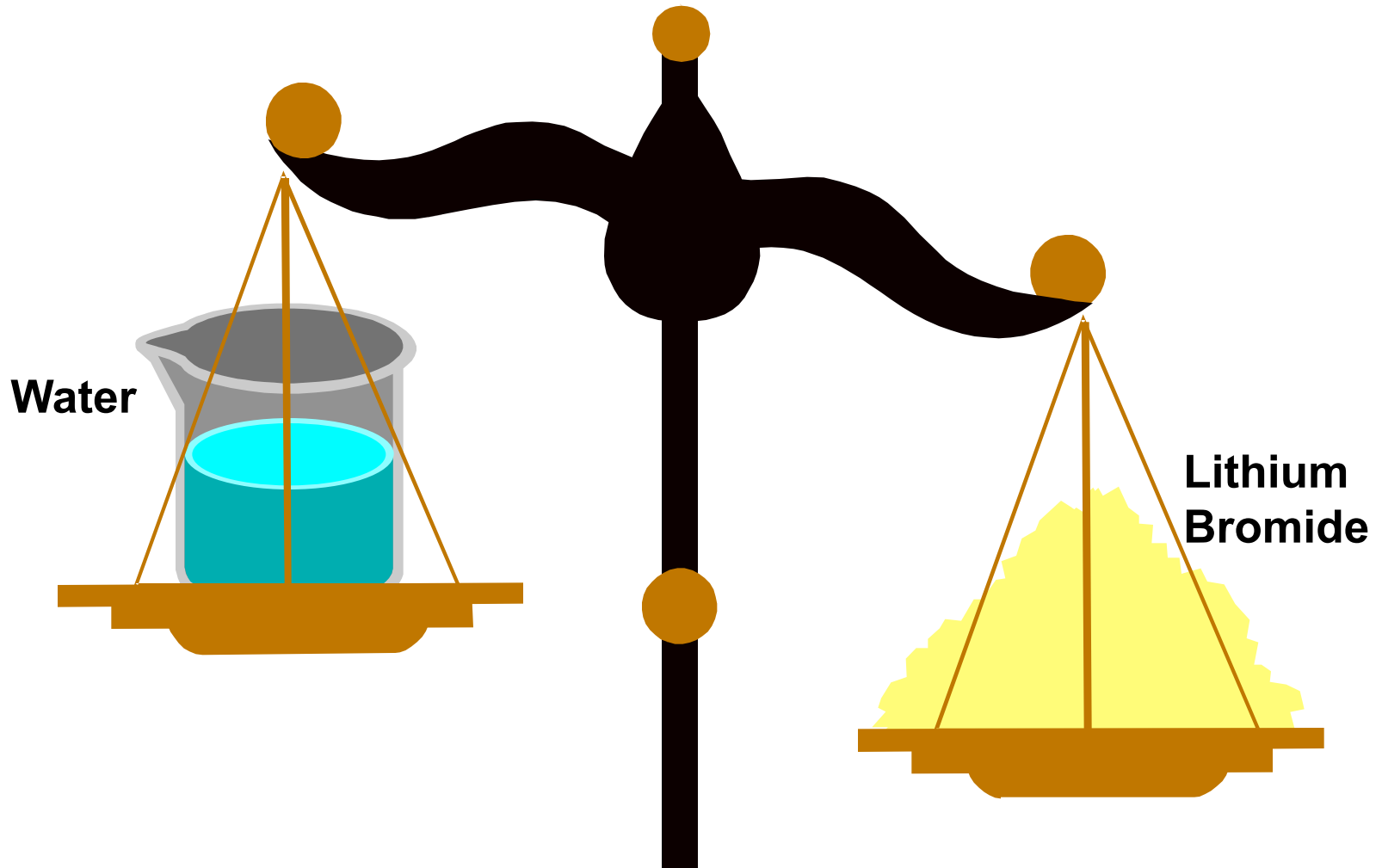
# Lithium Bromide

- Lithium Bromide is referred to as “solution”.
- This solution is expressed in “% concentration”.
- Lithium Bromide is shipped at a concentration of 55%.



# Concentrations

**Solution concentrations are determined by weight ...**



# Concentrations



+



= ???

# Concentrations



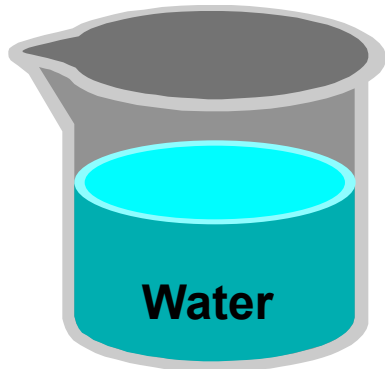
+



=

**2Lbs. of Solution @ 50% Concentration**

# Concentrations



Water

0%

**Weak Solution  
(Diluted)**



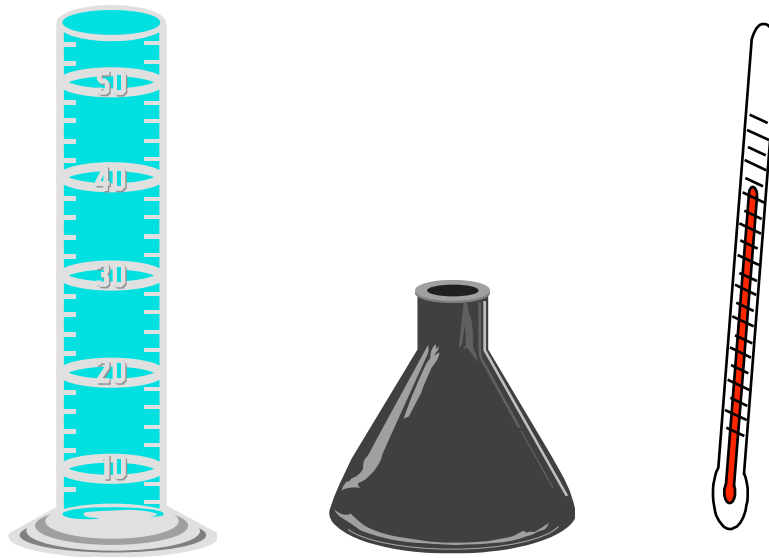
**1 Lb. Lithium  
Bromide**

100%

**Strong Solution  
(Concentrated)**

# Concentrations

**Solution concentrations can be determined by using instruments known as hydrometers to determine the specific gravity of the solution and by also measuring the temperature of the solution. This information can then be applied to a chart to determine the concentration of the sample.**

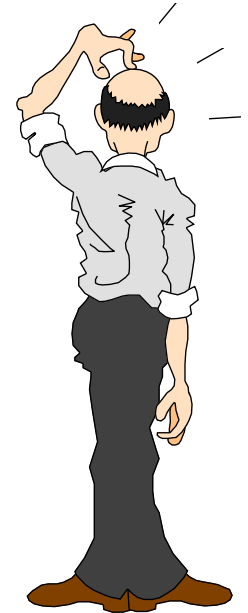


# Hydrometers



# Basic Refrigeration Cycle

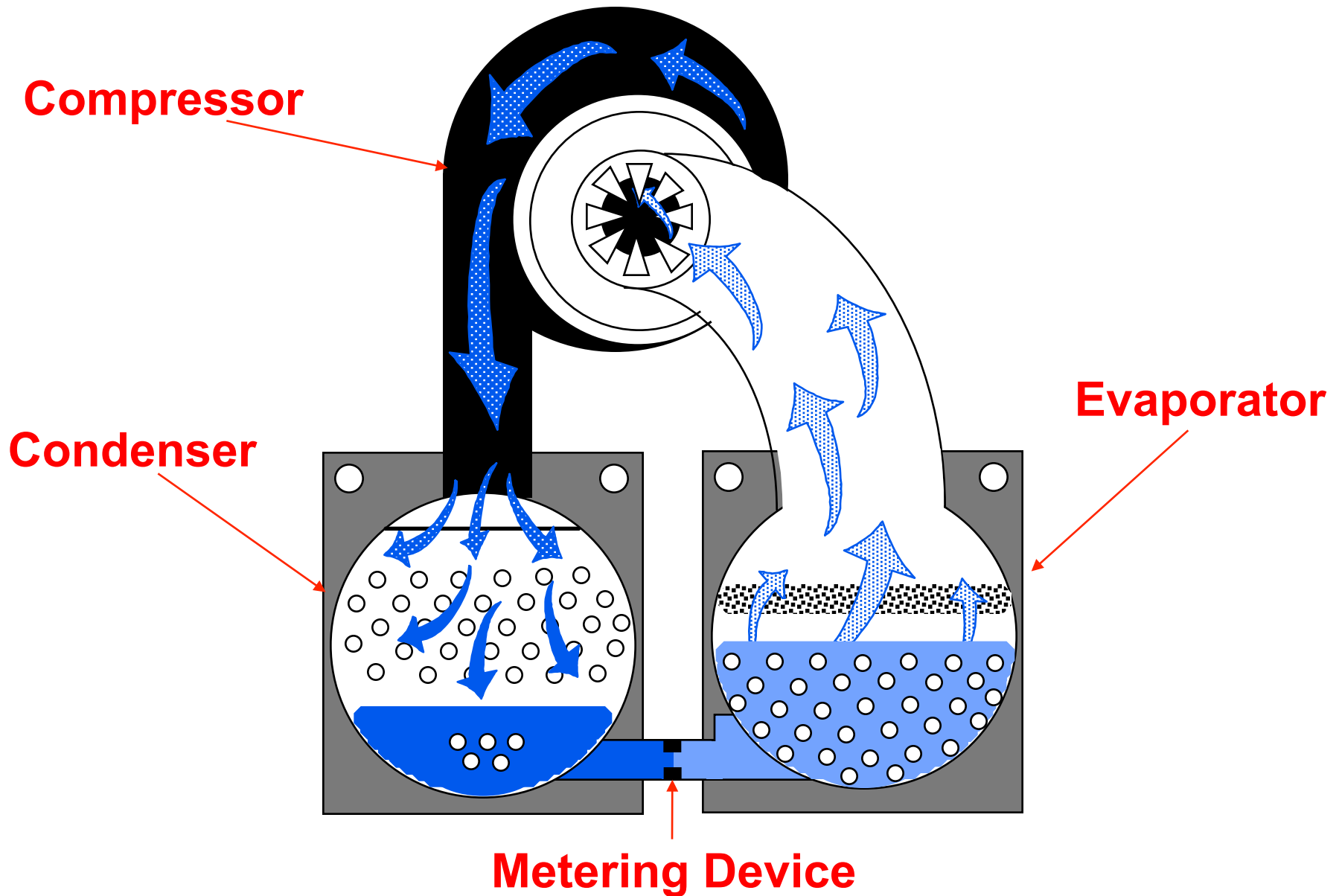
What are the four basic components of the vapor compression refrigeration cycle?



# Refrigeration Cycle Components

- Evaporator
- Compressor
- Condenser
- Metering Device

# Centrifugal Chiller Components



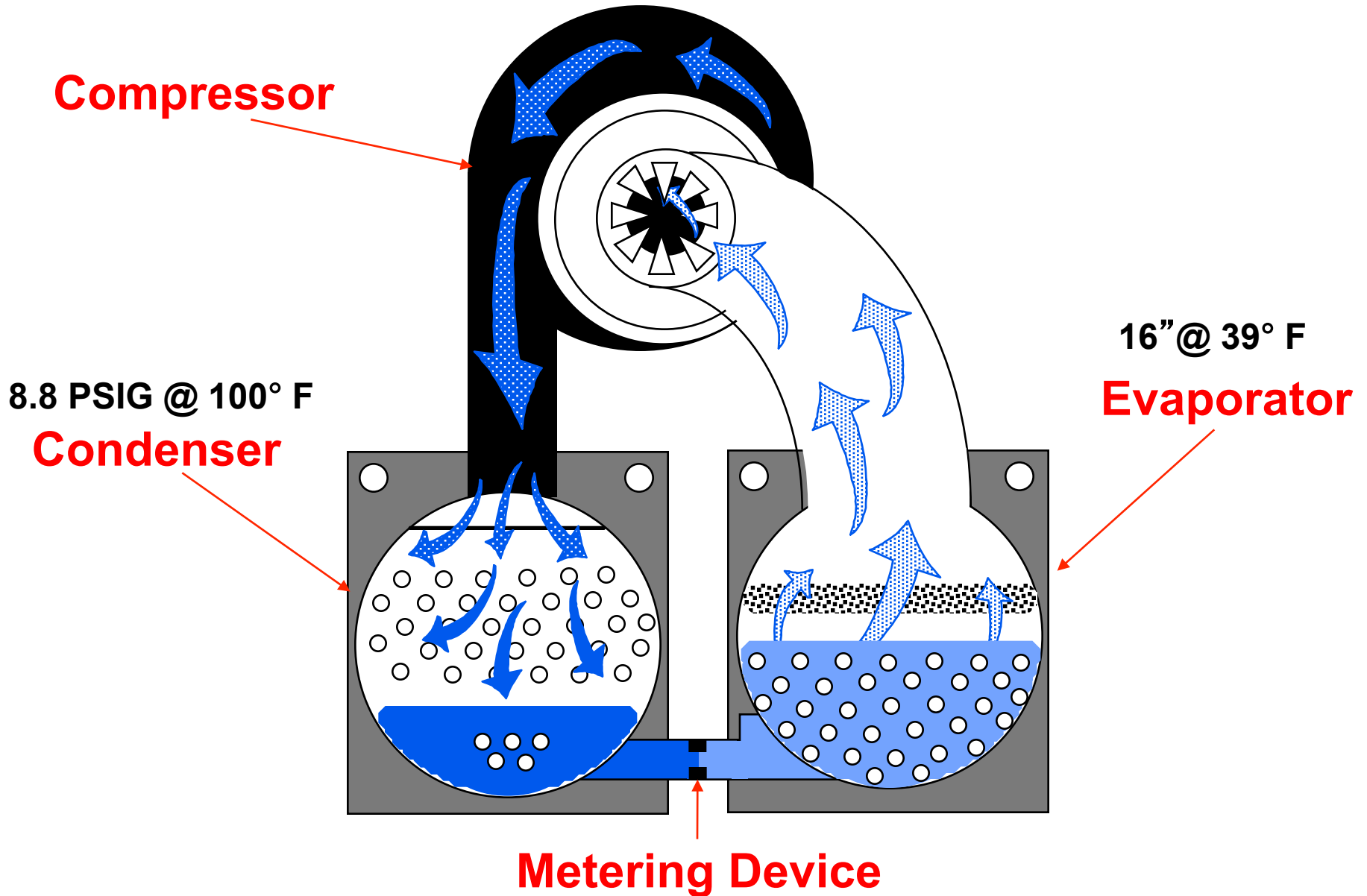
# Centrifugal Chiller

- What is the vapor pressure in an idle R-11 chiller if all the refrigerant is at 80° F?
- 1.5 PSIG

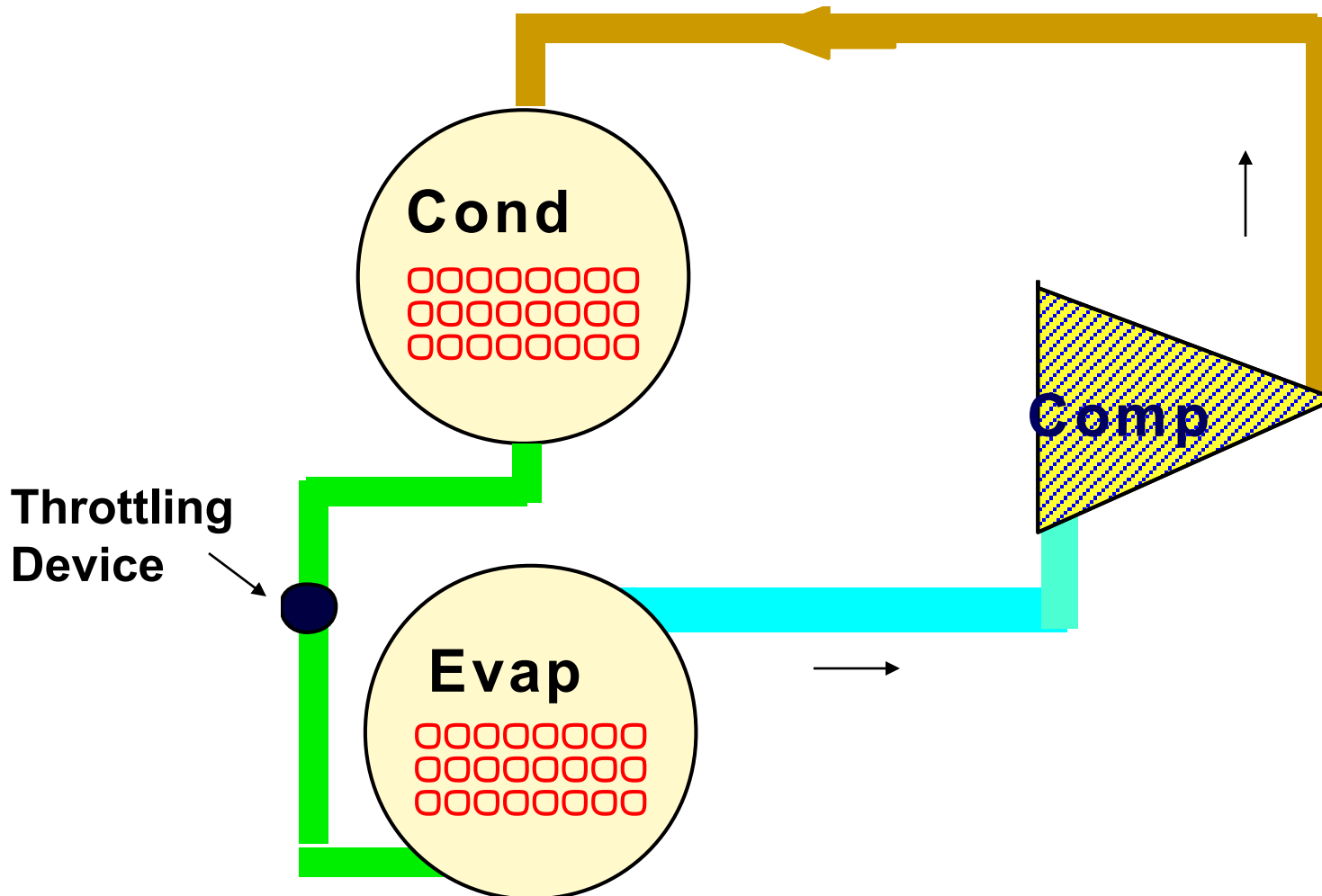
# Centrifugal Chiller

- What is the vapor pressure in an idle R-11 chiller if all the refrigerant is at 70° F?
- 2.8 in. Hg.

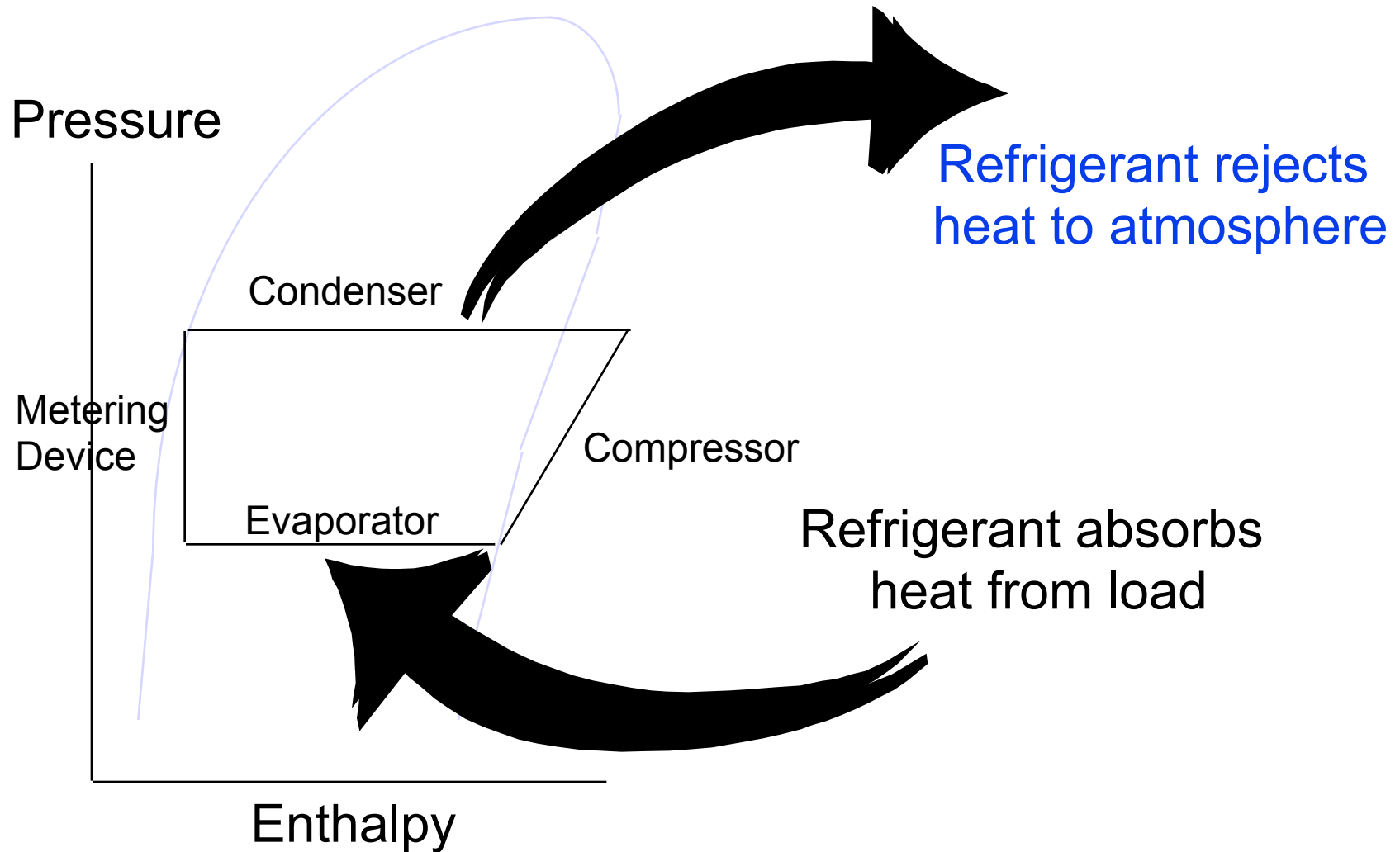
# Centrifugal Chiller Cycle



# Vapor Compression Cycle

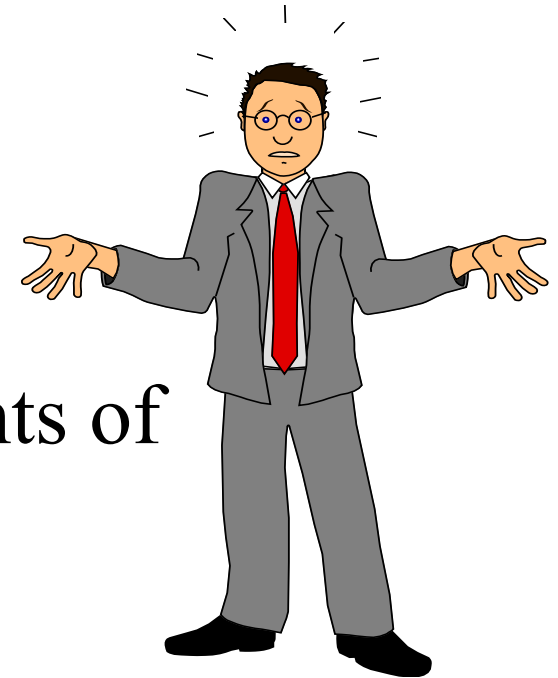


# Pressure - Enthalpy Diagram



# Basic Absorption Cycle

What are the four main components of the absorption cooling cycle?



# Absorber Components

- Condenser
- Generator
- Evaporator
- Absorber

# Absorption Vs. Vapor Compression

## Vapor Compression

## Absorption

Evaporator.....Evaporator

Condenser.....Condenser

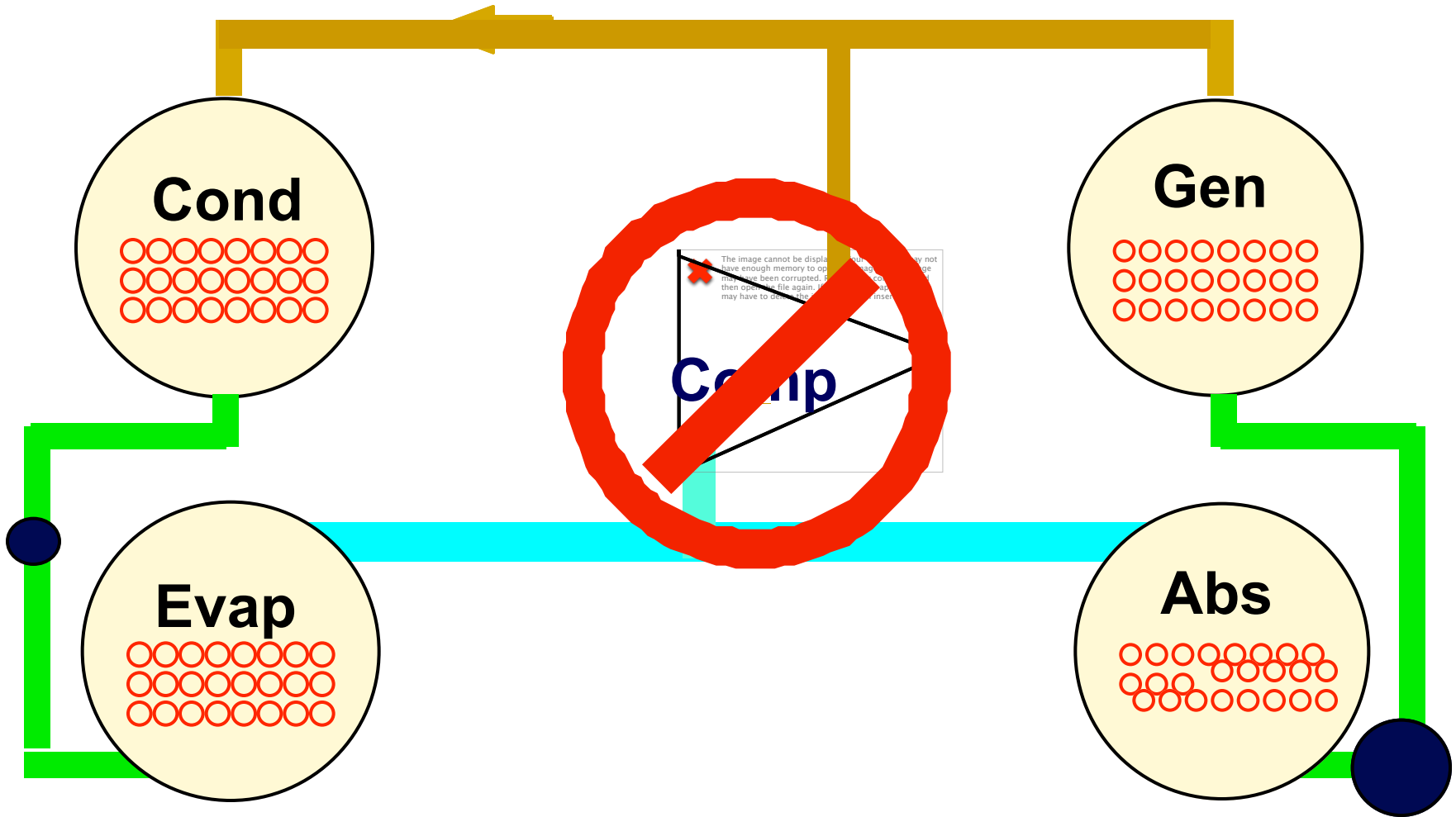
Throttling Device..... Throttling  
Device

Compressor .....Absorber

Pump

Generator

# Absorption Unit Conversion



# Absorption Vs. Vapor Compression

- Evaporator, condenser and throttling device are present on both technologies
- Compressor is replaced with the absorber and generator sections
- A pump is used to transport the solution from the absorber to the generator

# Absorption Vs. Vapor Compression

- Absorber is analogous to the suction stroke of the compressor.

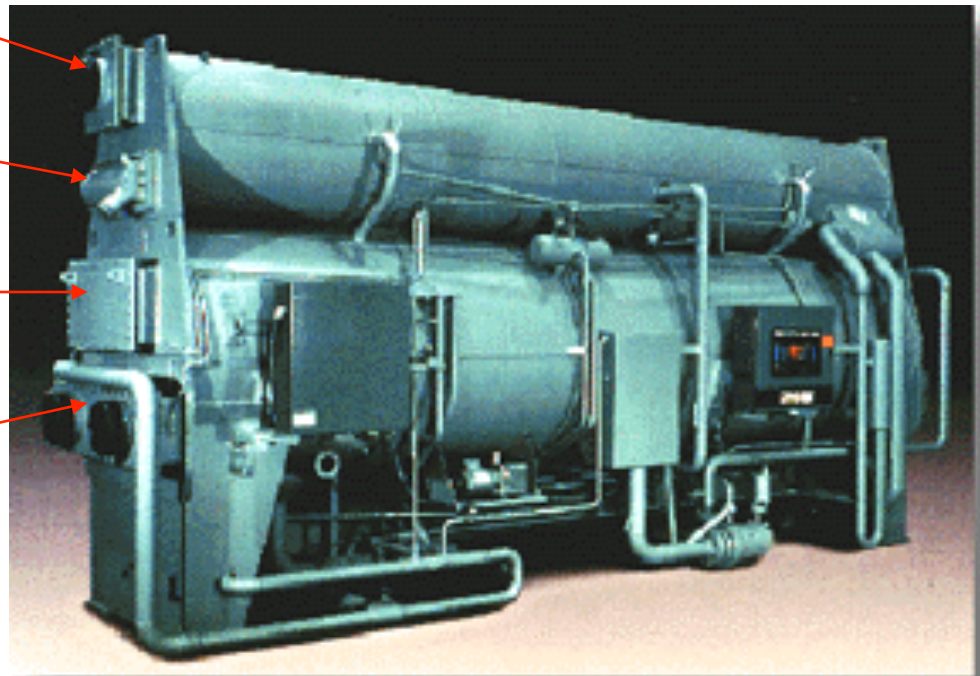
# Single Stage Absorber

**Condenser**

**Generator**

**Evaporator**

**Absorber**



# Generator

- Concentrates weak lithium bromide solution by boiling off refrigerant (water).
- Can be fired by either steam or hot water.

# Absorber

- Absorbs refrigerant vapor from the evaporator. In the process, the concentration of the solution is decreased (more water). The weight of the solution increases due to mass transfer.
- Three forms of heat must be removed from the absorber section. This heat is removed by the tower water.
  - **Sensible Heat** - Heat necessary to lower the temperature and vapor pressure of the solution below that of the refrigerant vapor vapor pressure.
  - **Latent Heat of Condensation** - Heat necessary to change refrigerant from a vapor to liquid.

# Absorber (Cont.)

- **Heat of Absorption (Dilution)** - Heat generated during the absorption process. Heat is given off as the solution is diluted.

Absorption process will cease if the above heat is not removed.

## Absorber (Cont.)

The absorption rate increases with an

- Increase in solution concentration
- Decrease in solution temperature

In both instances, the vapor pressure of the solution is lowered.

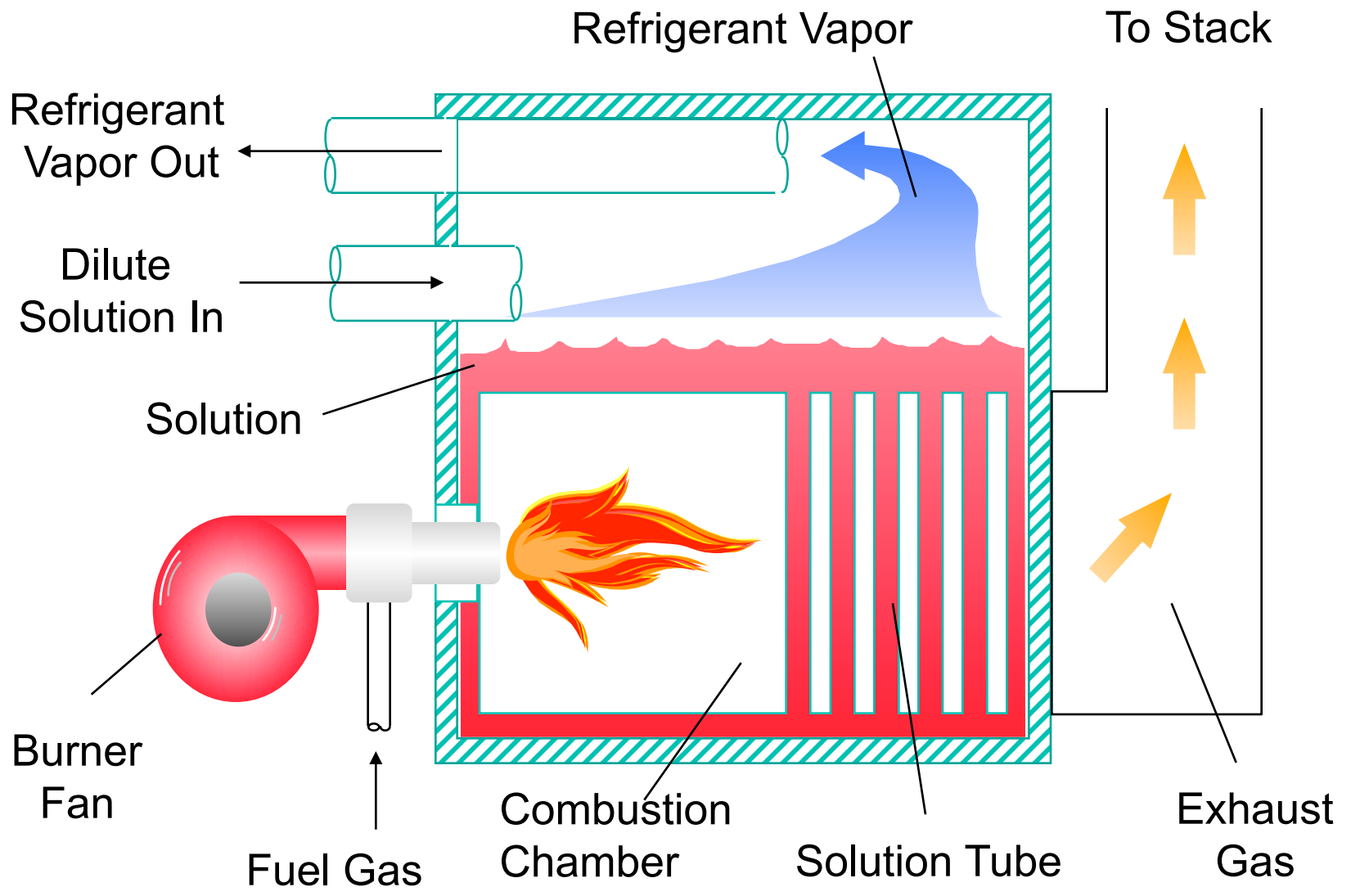
# Condenser

- Removes sufficient heat to condense refrigerant vapor coming from the generator.
- Heat is removed by the tower (cooling) water
- Tower water flows in series; 1st through the absorber tube bundle and then through the condenser

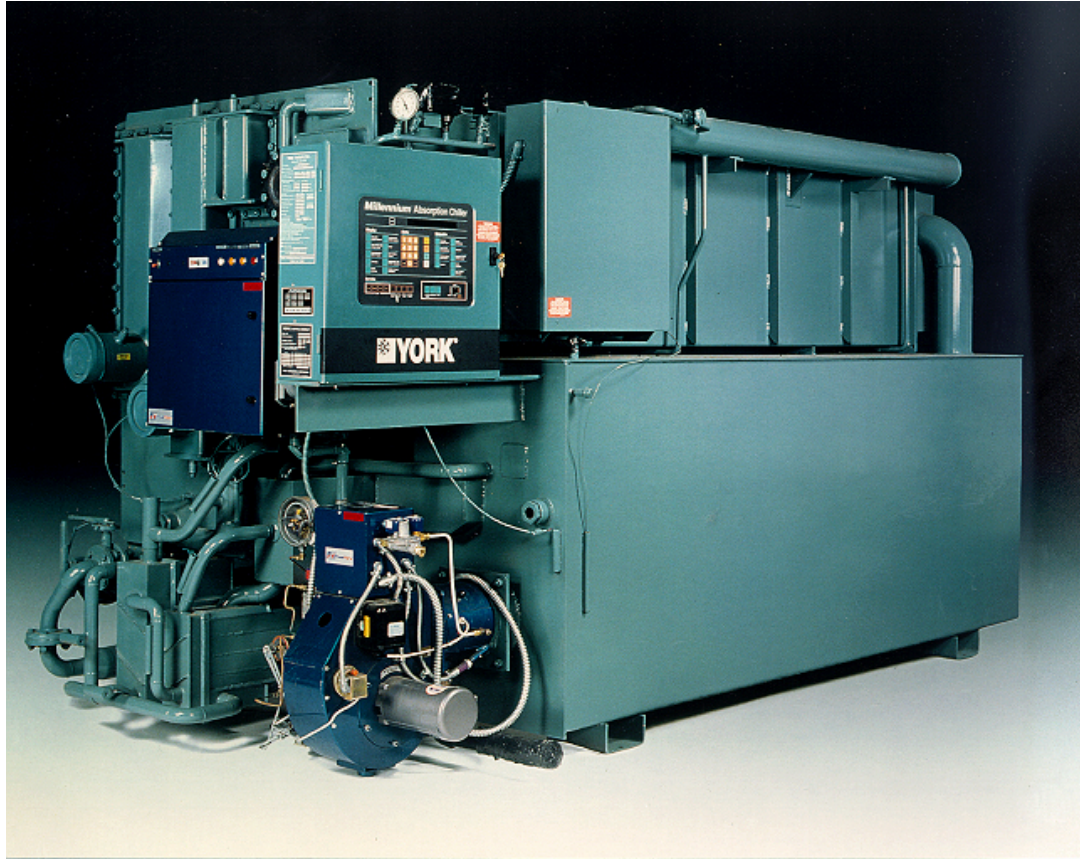
# Evaporator

- Absorbs heat from building chilled water loop
- Spray Type Evaporator

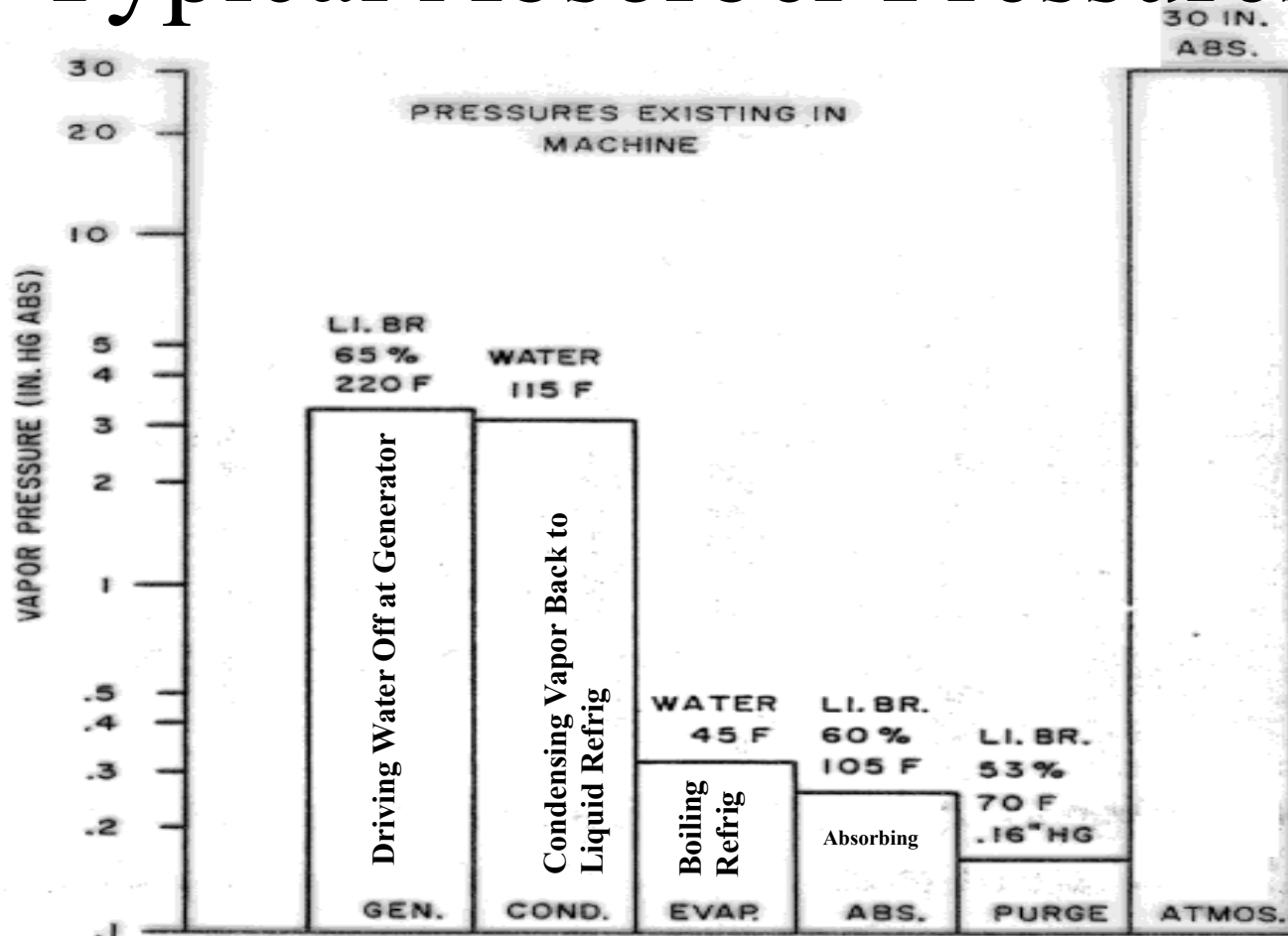
# Two-Stage Direct-Fired 1st Stage Generator



# Two-Stage Absorber



# Typical Absorber Pressures



**NOTE:** Absorption machine at full load, using lithium bromide as absorbent.

— PRESSURES AND TEMPERATURES OF A TYPICAL ABSORPTION MACHINE



Heat Source In

Generator

Out

Evaporator

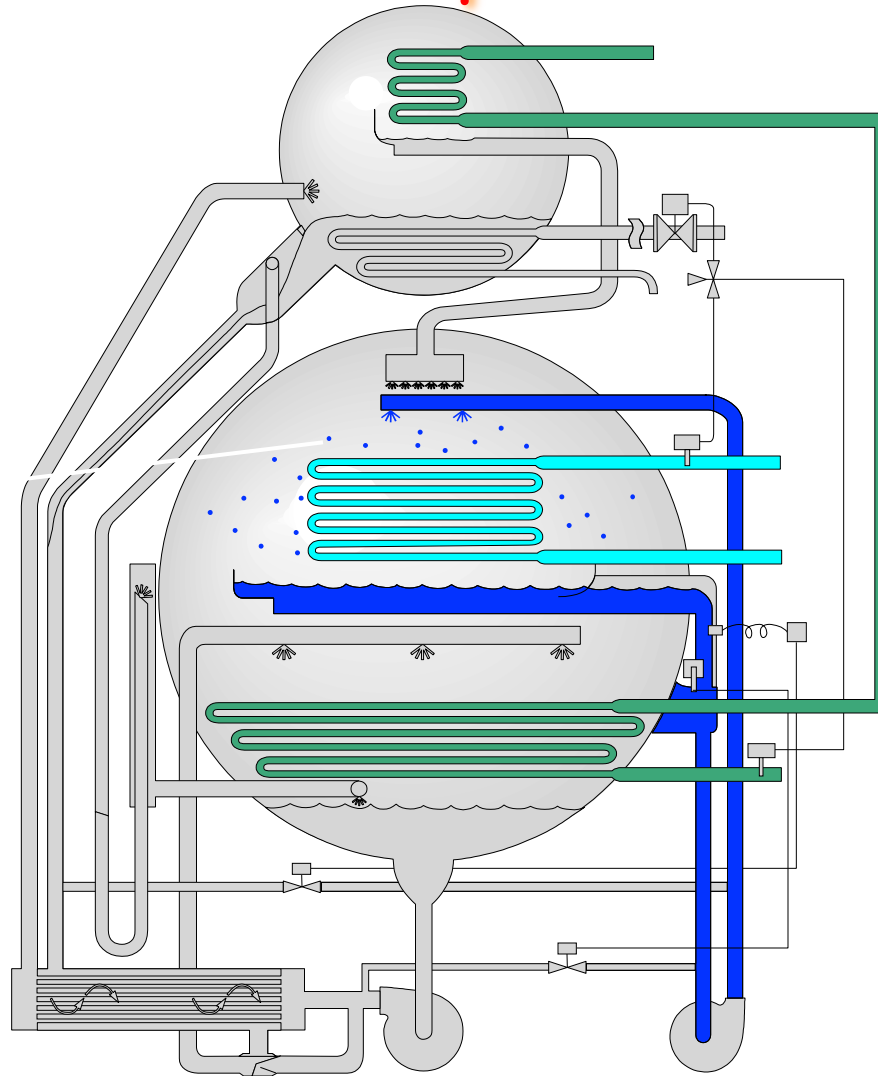
Chilled Water In

Absorber

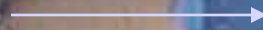
Tower Water In

# YIA: Evaporator

Evaporator



Hole for Spray header



Mist  
Eliminators





Spray Header Supports

Evaporator Tray

Refrigerant Outflow Box







Evaporator Tray showing  
outflow





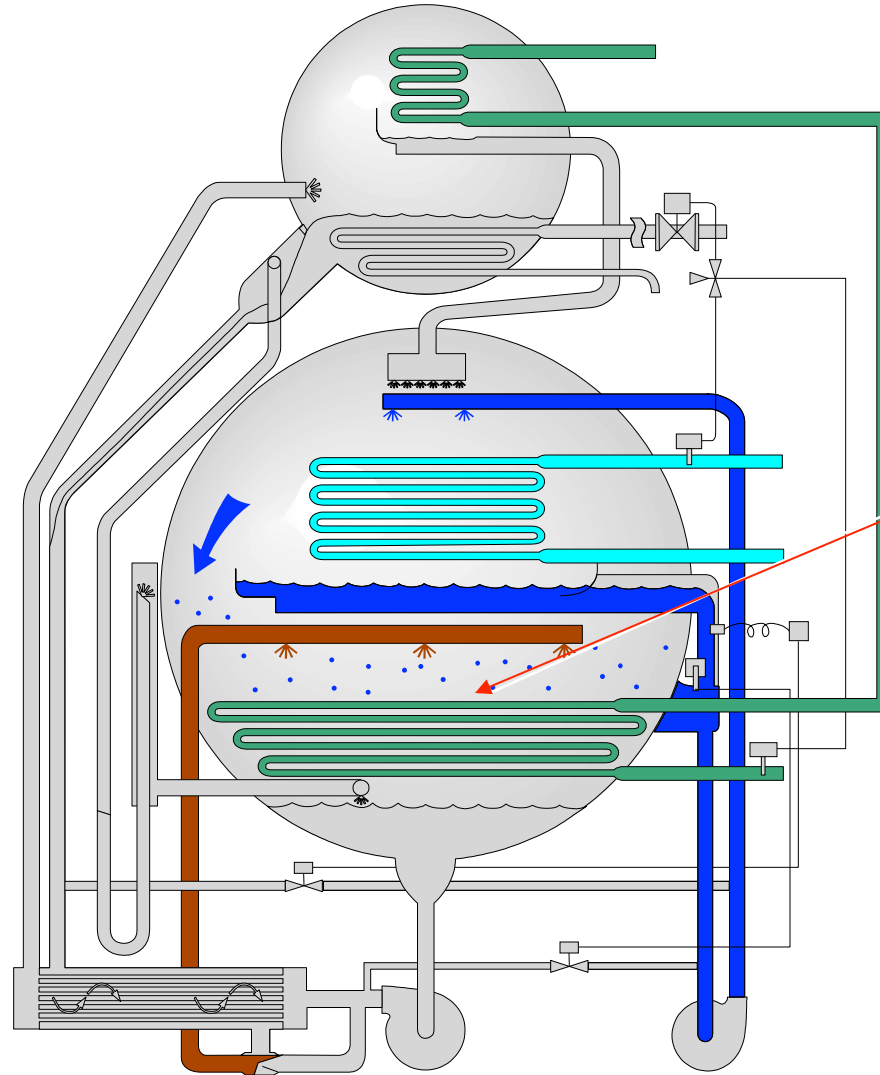
View of Refrigerant  
Spray Header (Upside  
Down)

Solution Flow





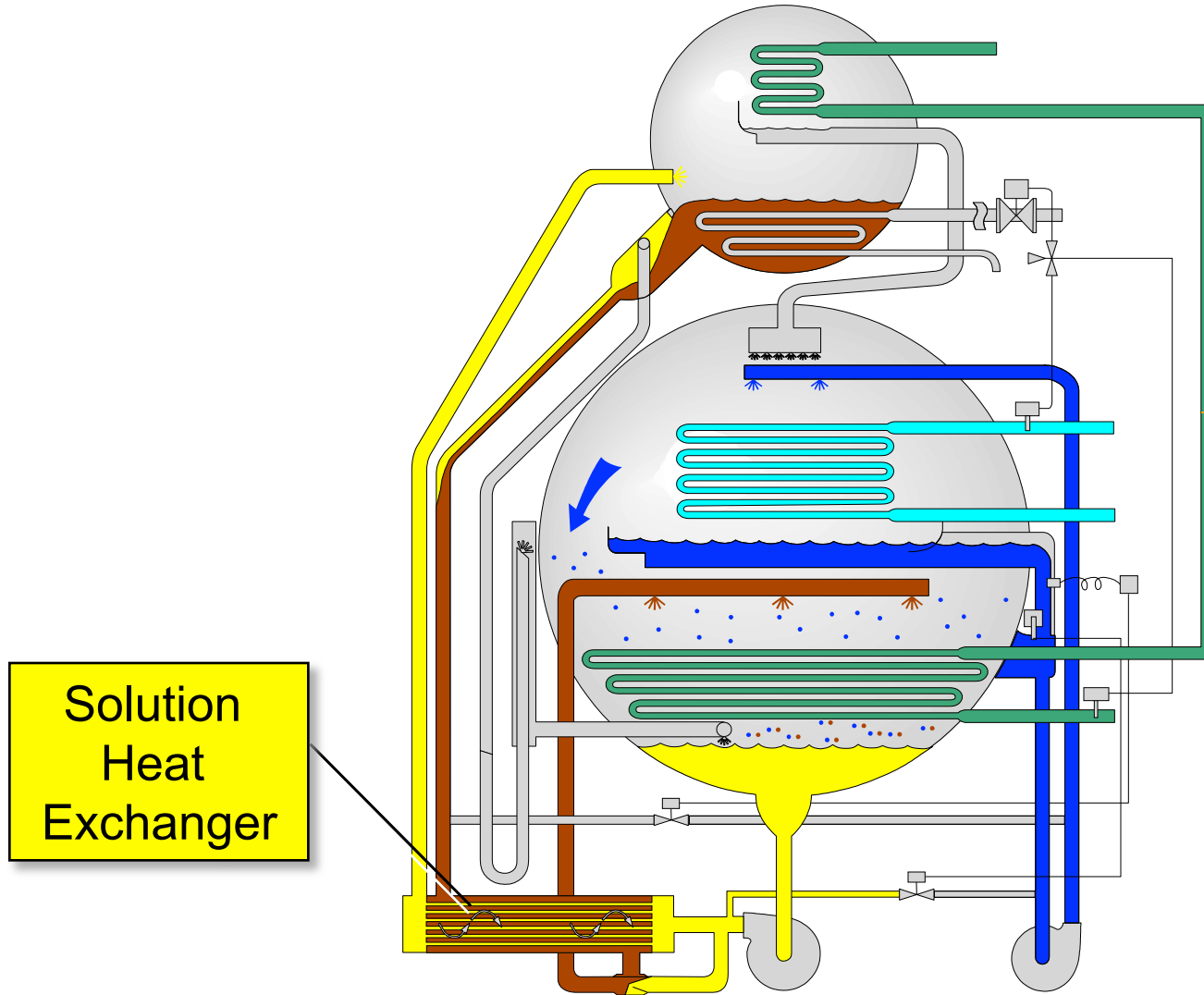
# YIA: Absorber



Absorber



# YIA: Solution Heat Exchanger



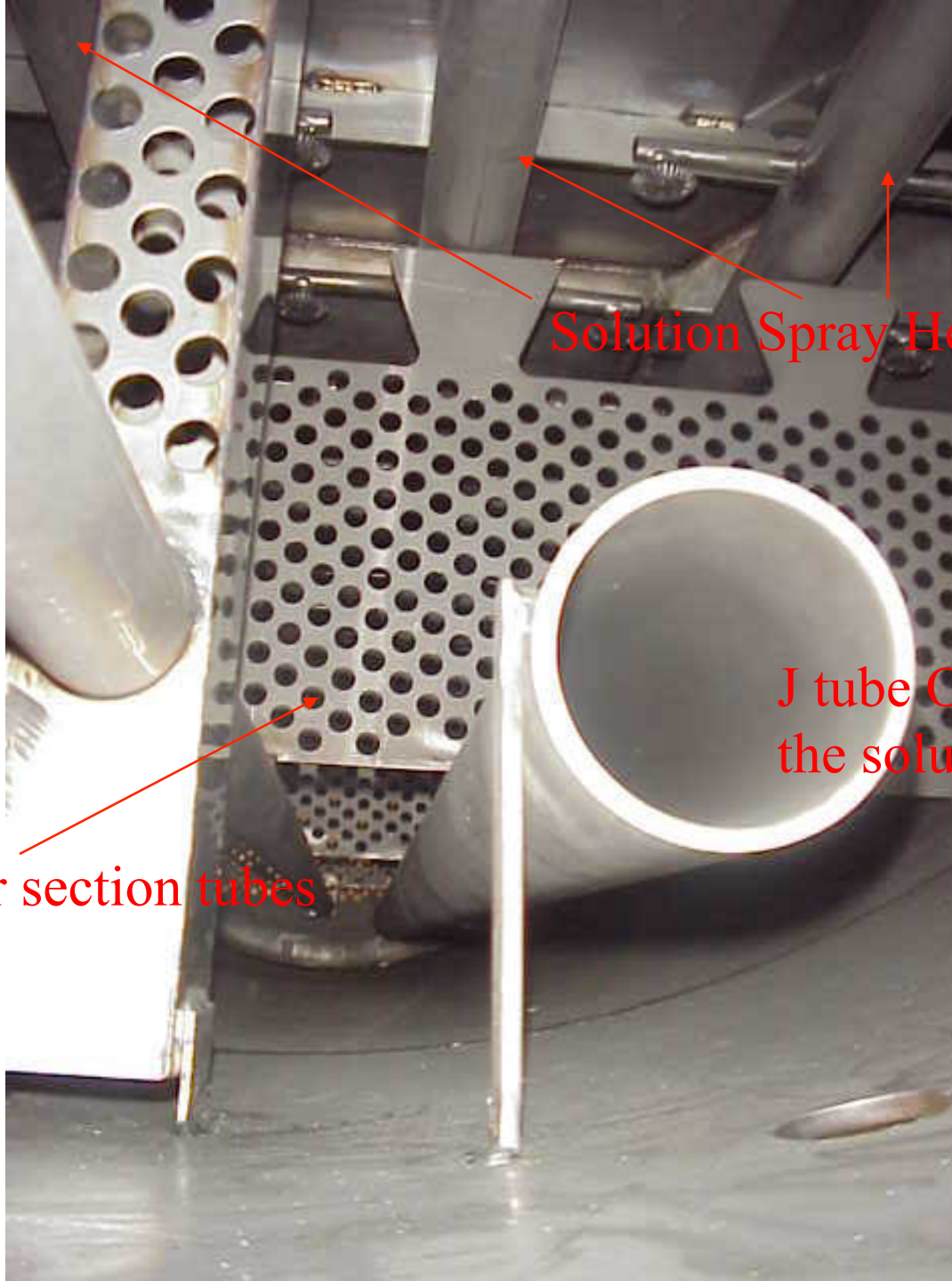


Purge Suction Header

Solution Spray Headers

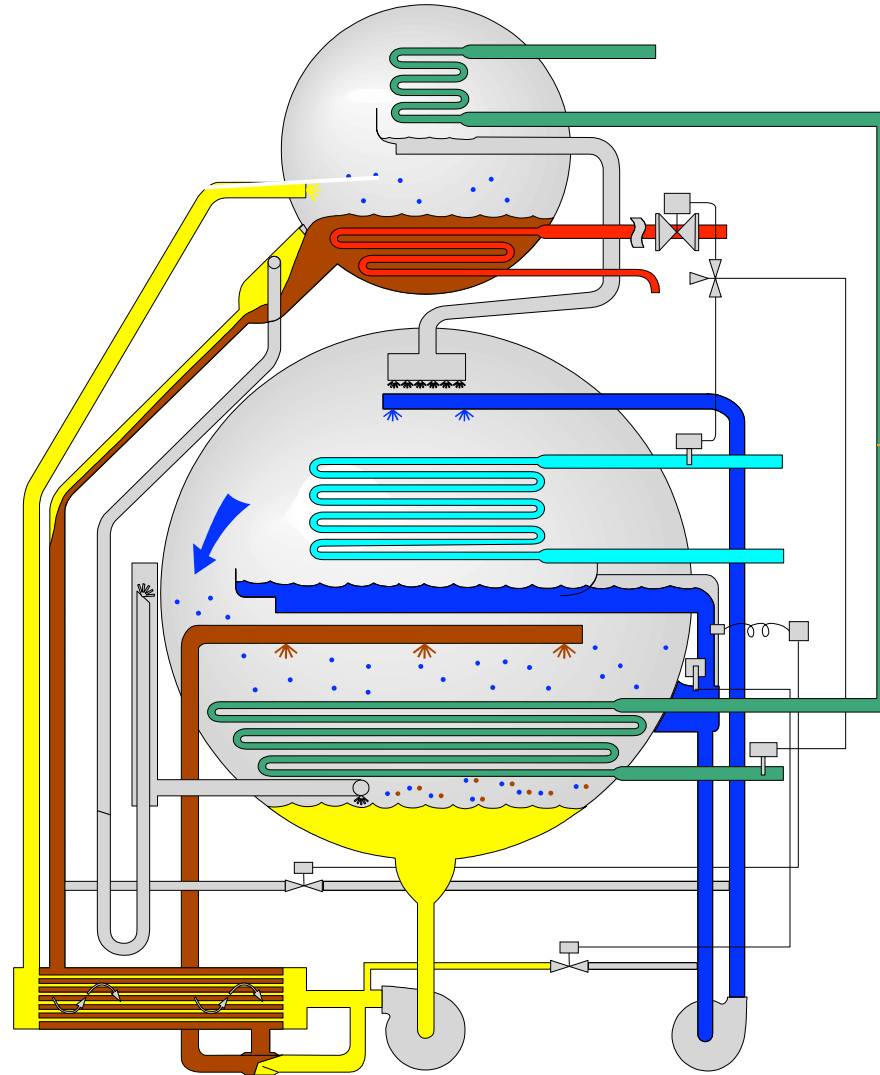
J tube Out flow into the solution suction

Absorber section tubes

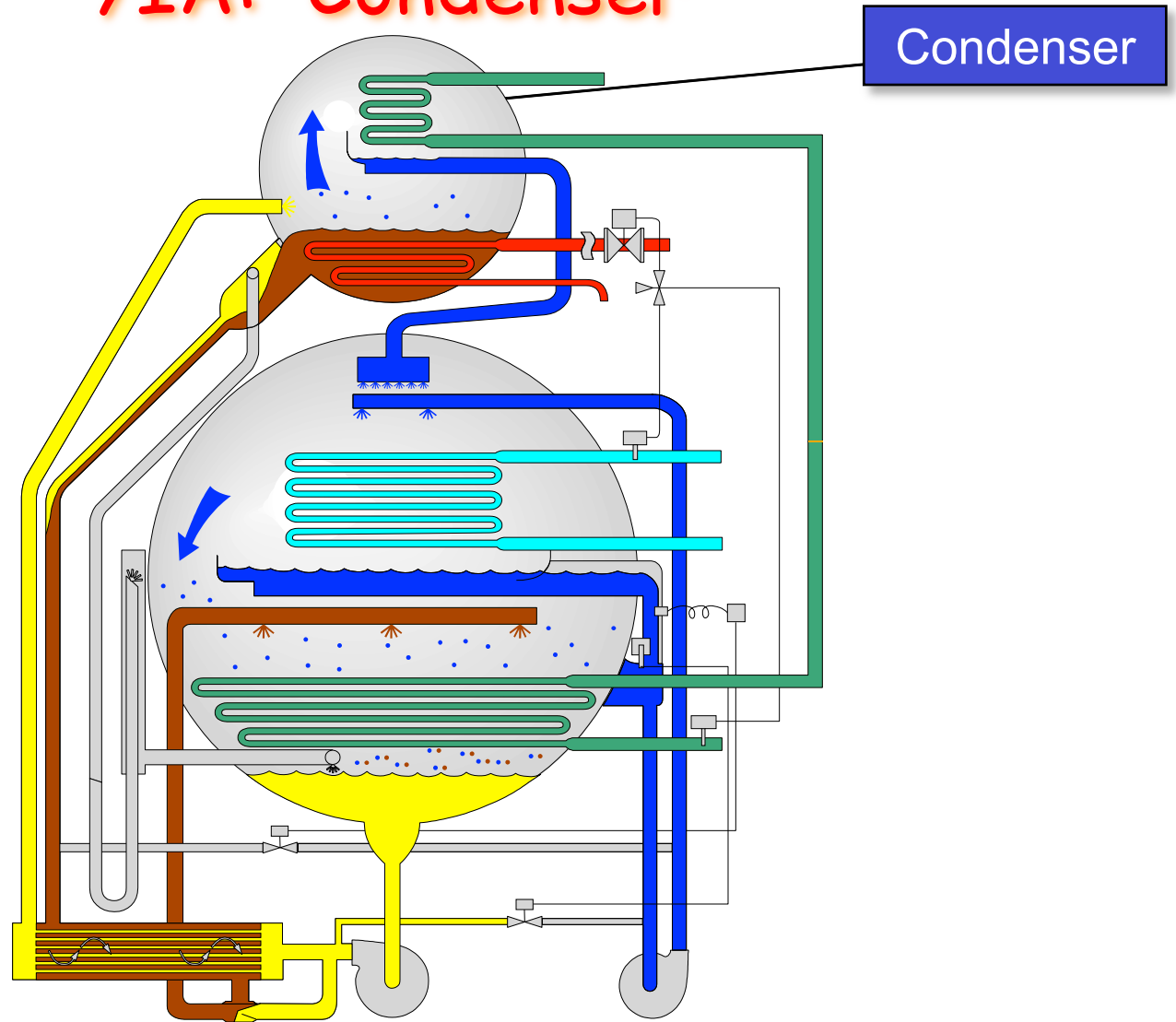


# YIA: Generator

Generator



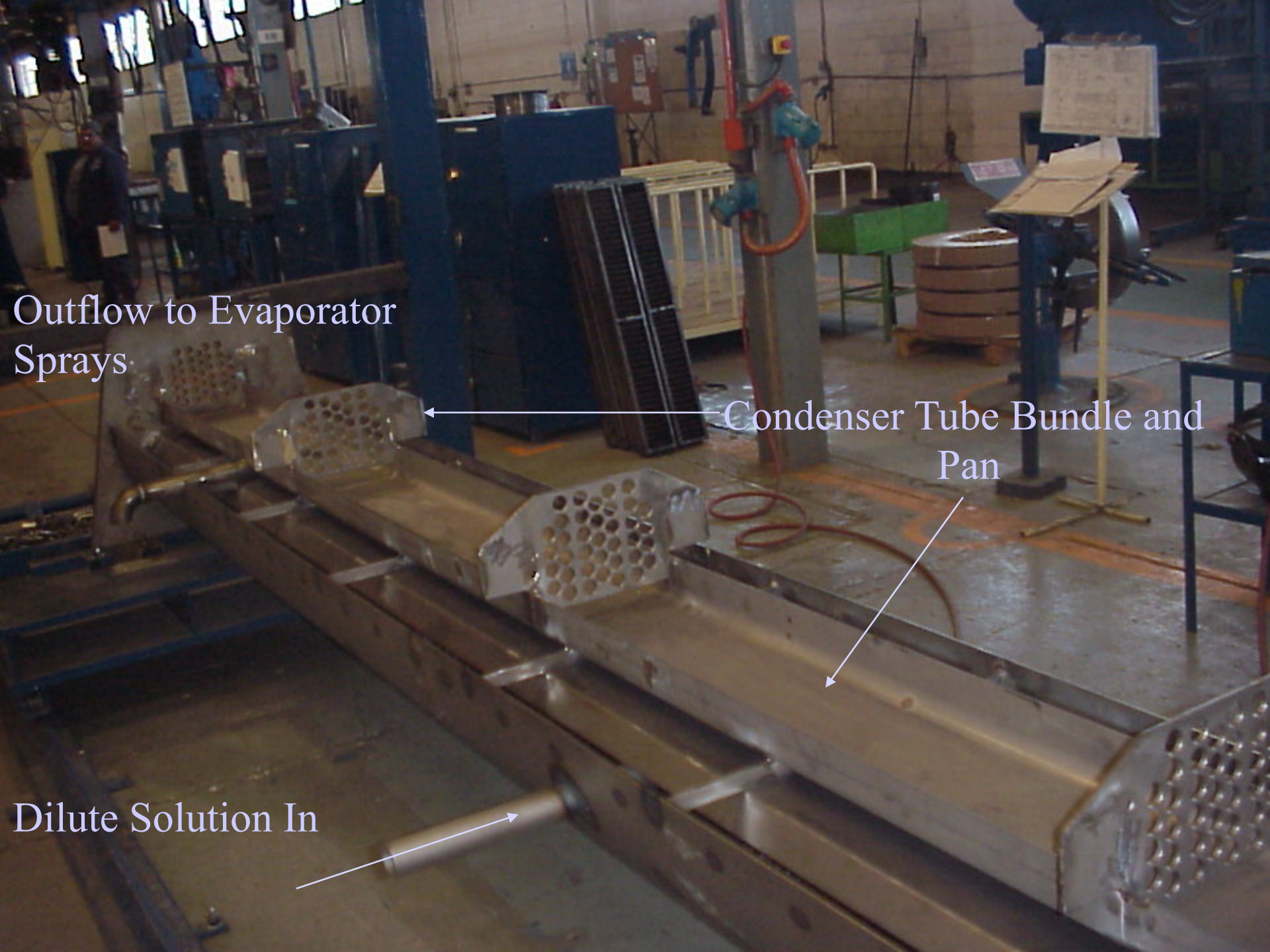
# YIA: Condenser

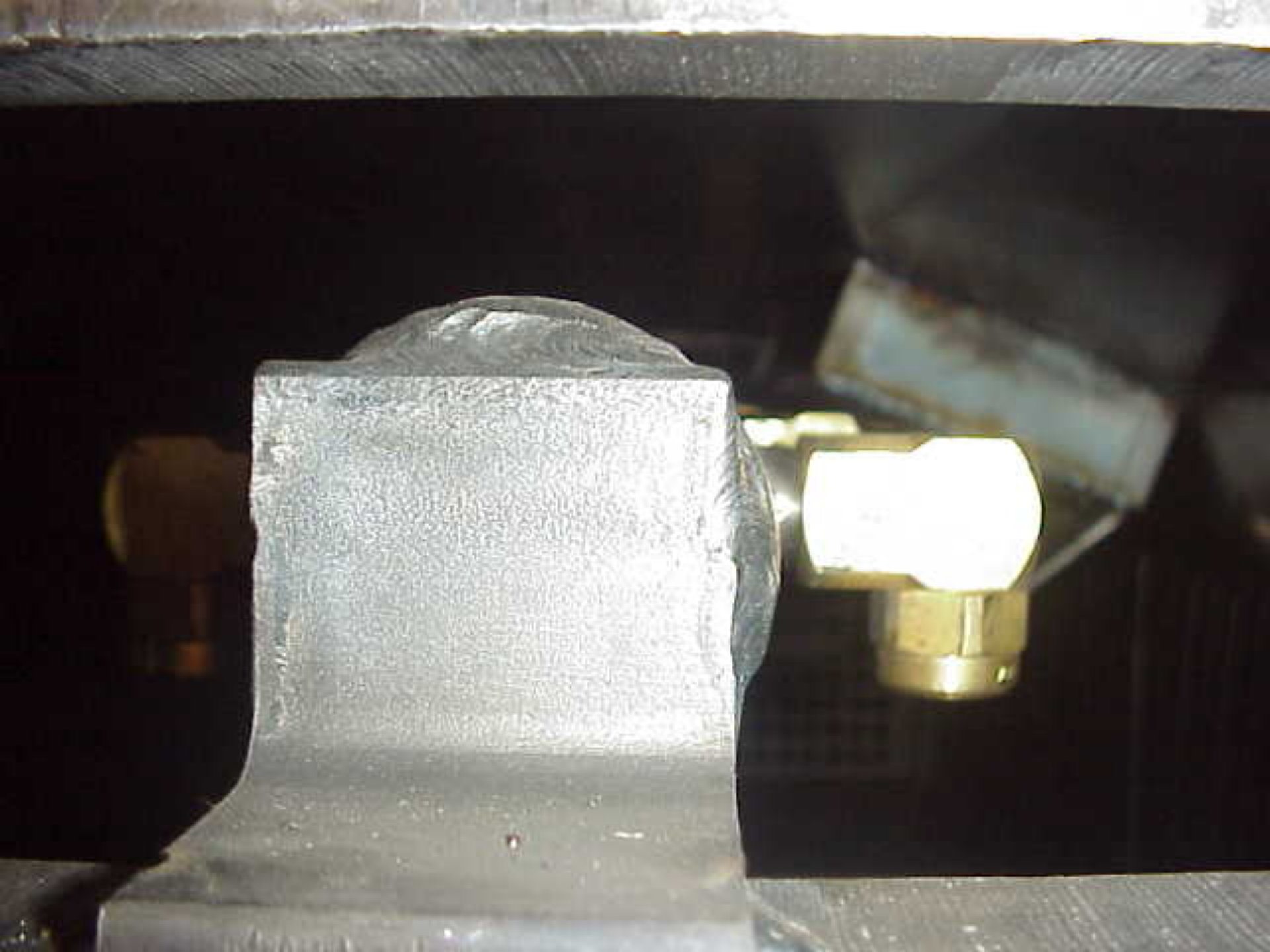


Outflow to Evaporator  
Sprays

Condenser Tube Bundle and  
Pan

Dilute Solution In





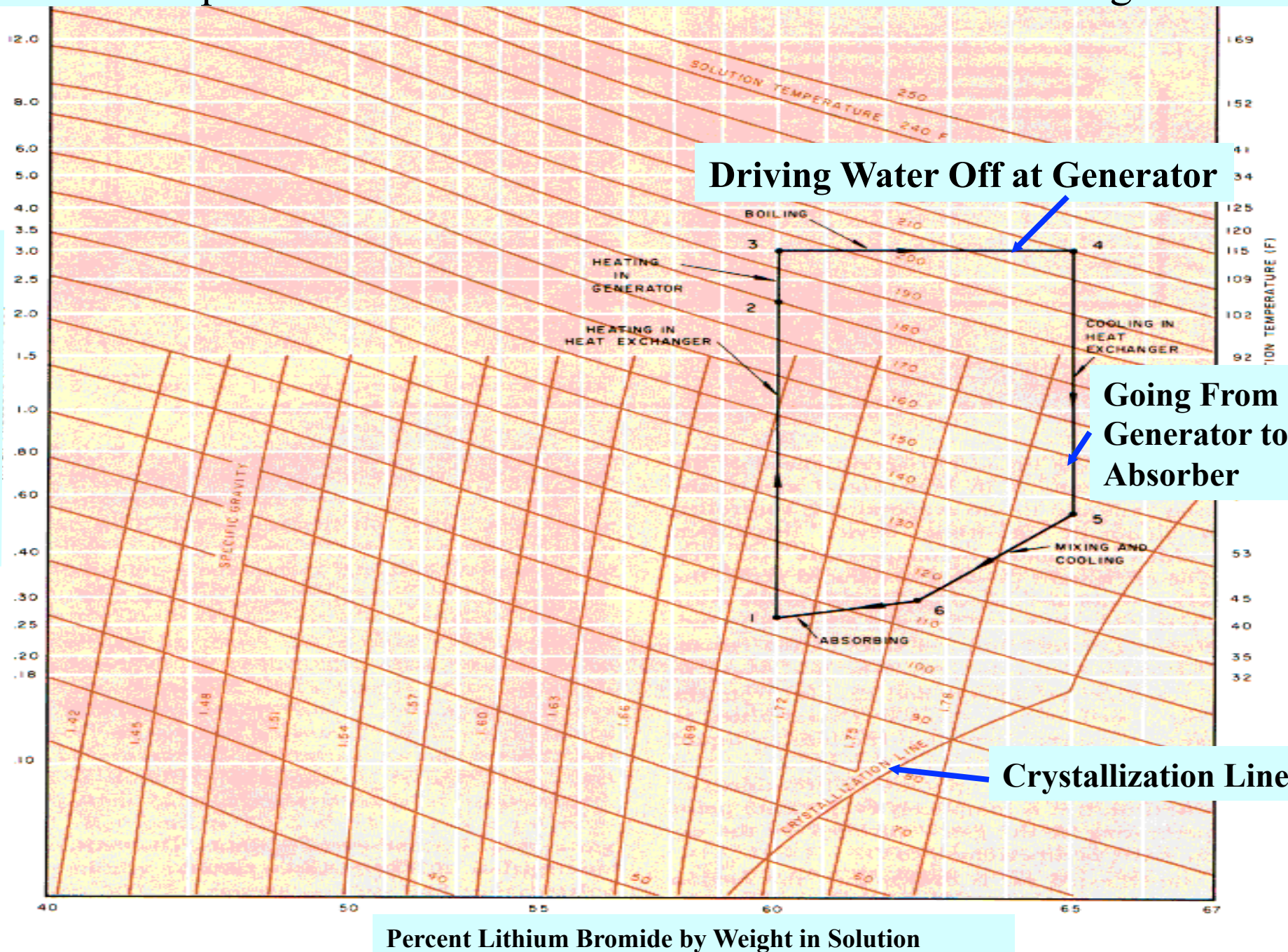
Sections of Condenser



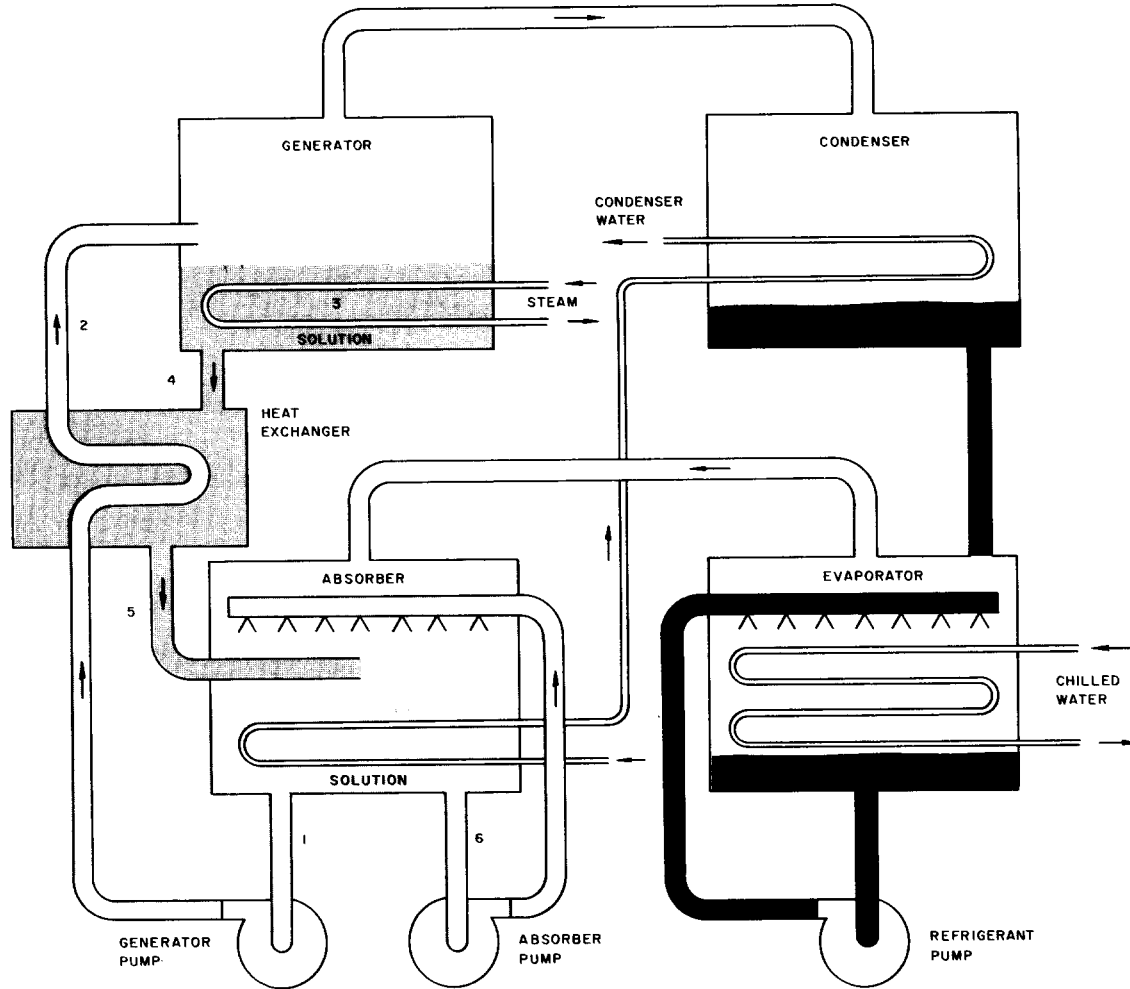
# Lithium Bromide

- A brine (salt solution)
- Reactive when combined with oxygen
- Can cause corrosion inside chiller
- Can crystallize when too concentrated

# Absorption Process and Lithium Bromide Solution Changes



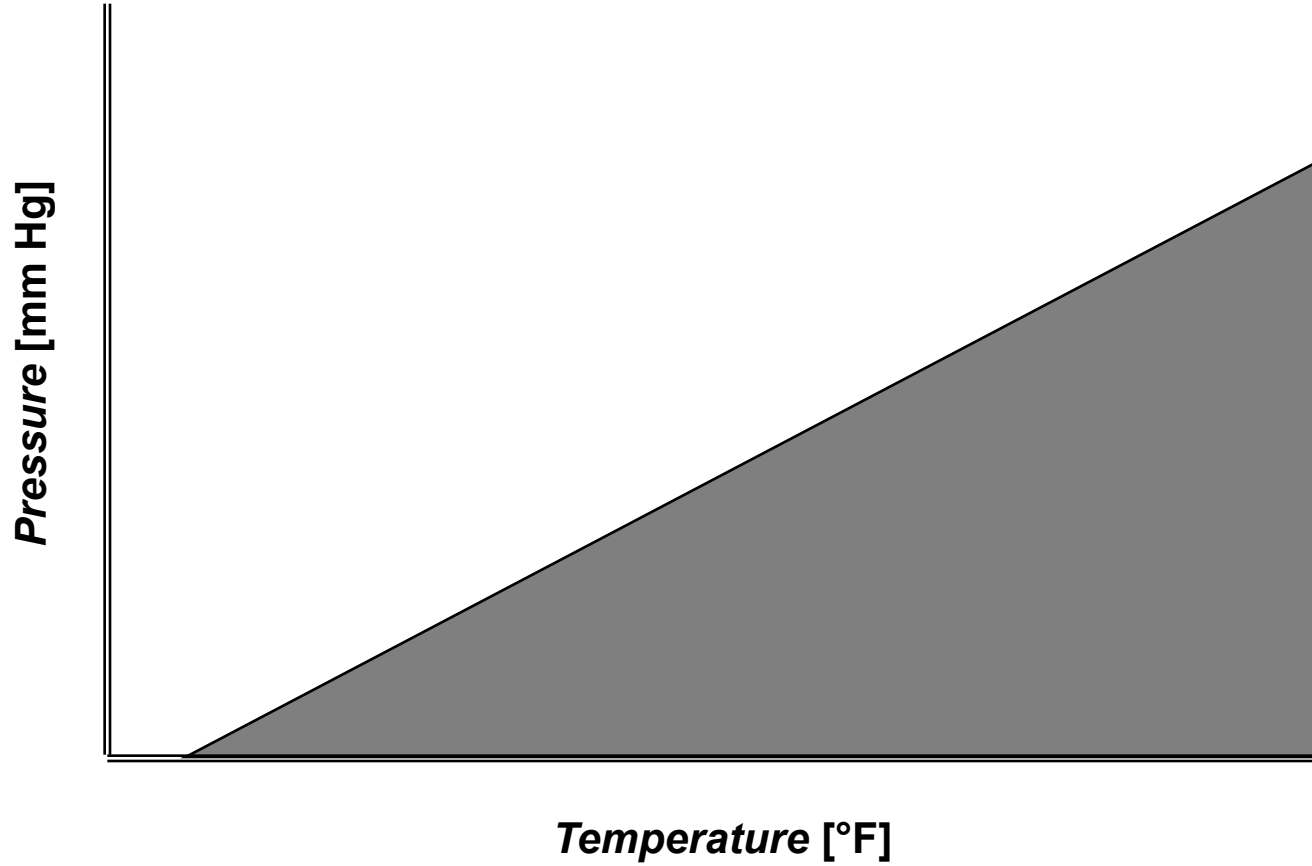
# Absorption Cycle



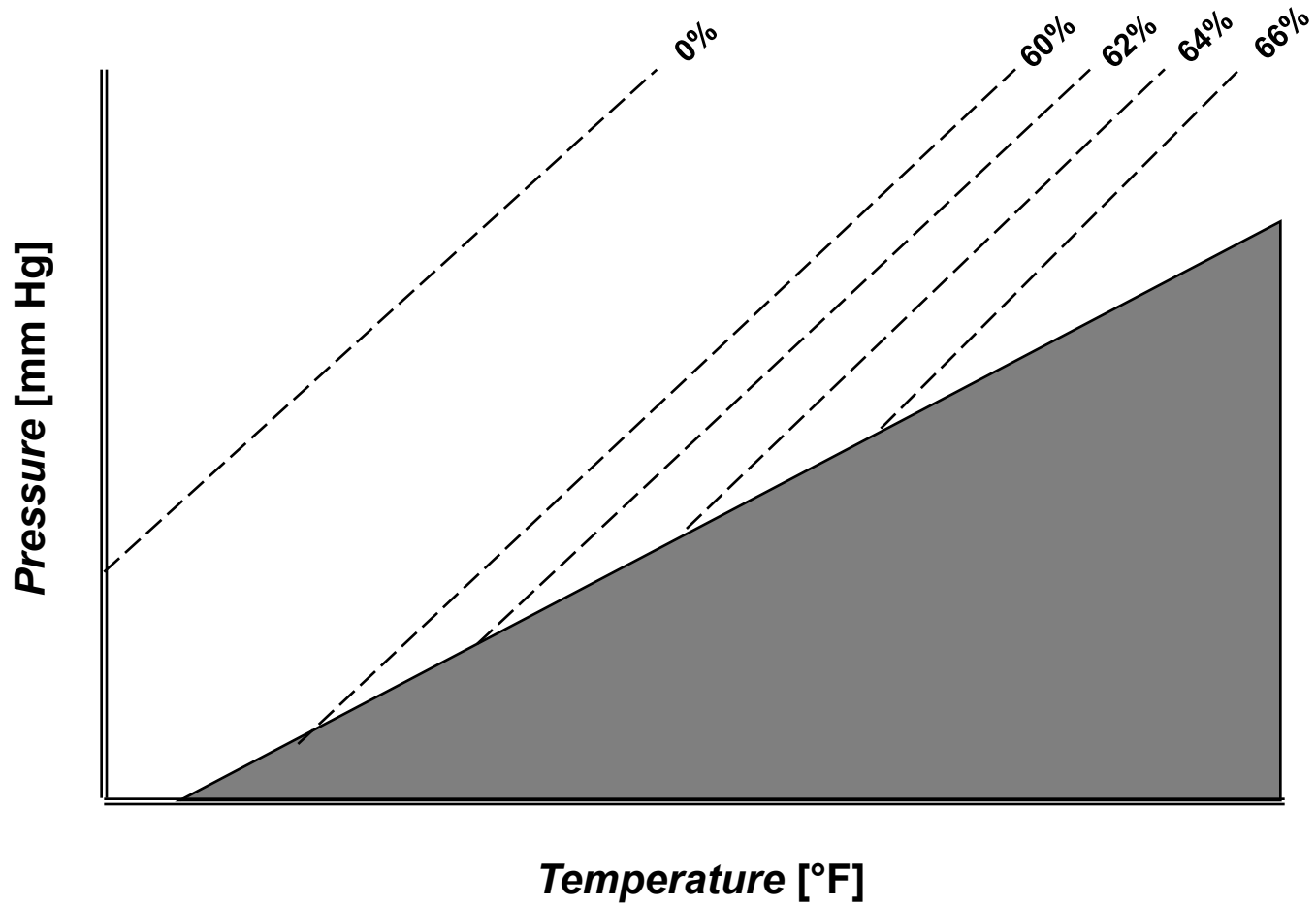
What is crystallization and  
how does it happen?



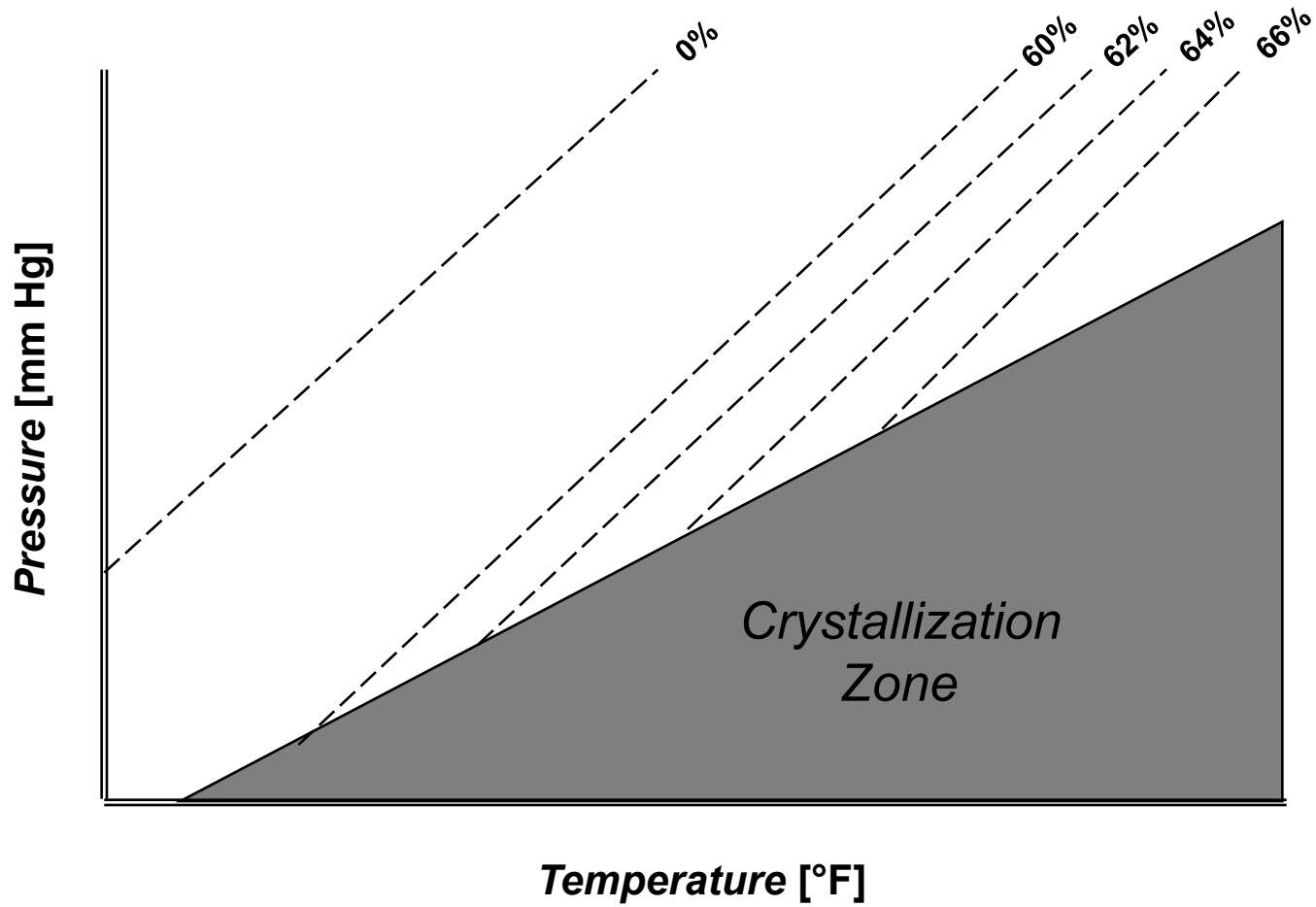
# PTX Chart

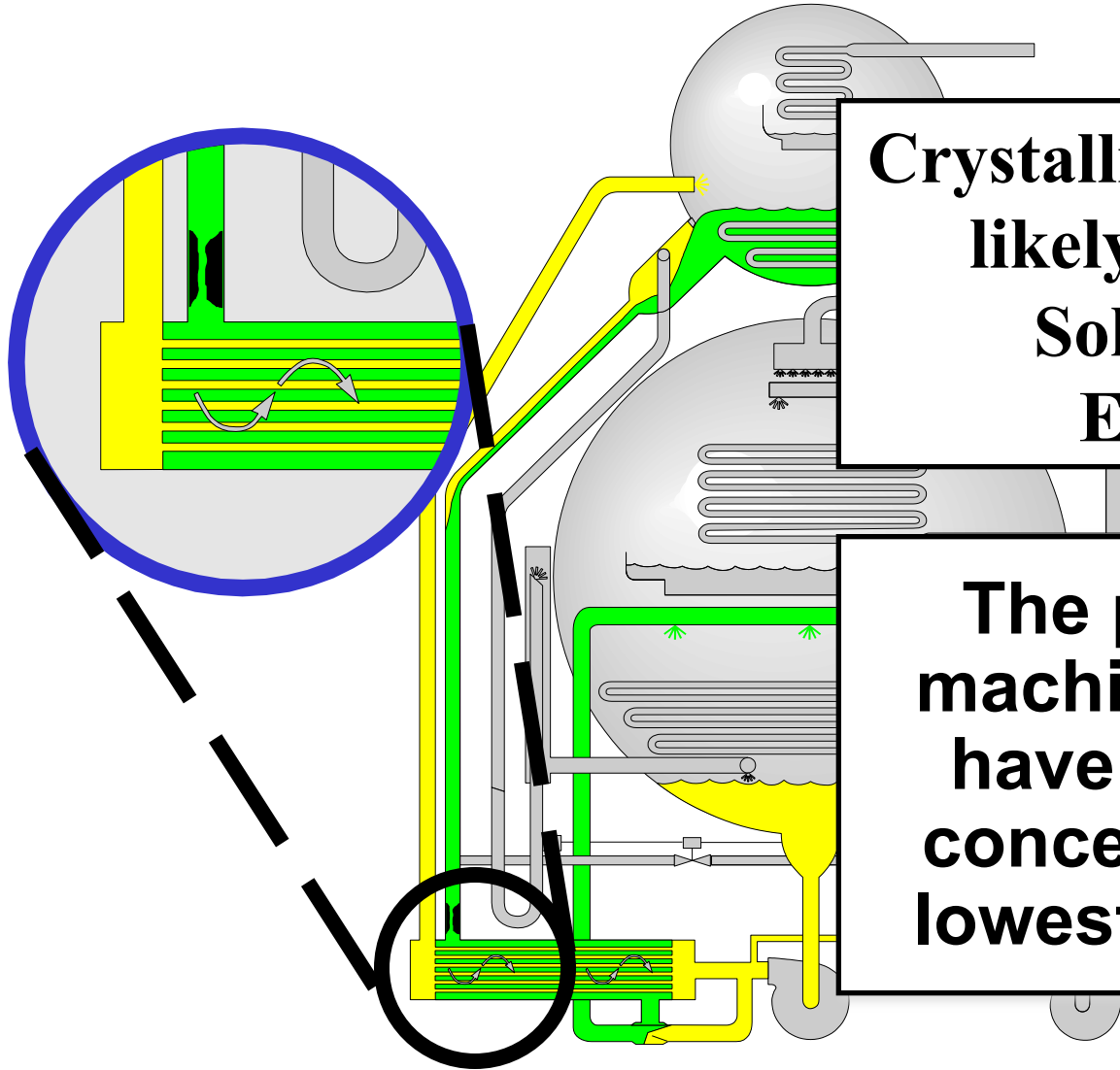


# PTX Chart



# PTX Chart

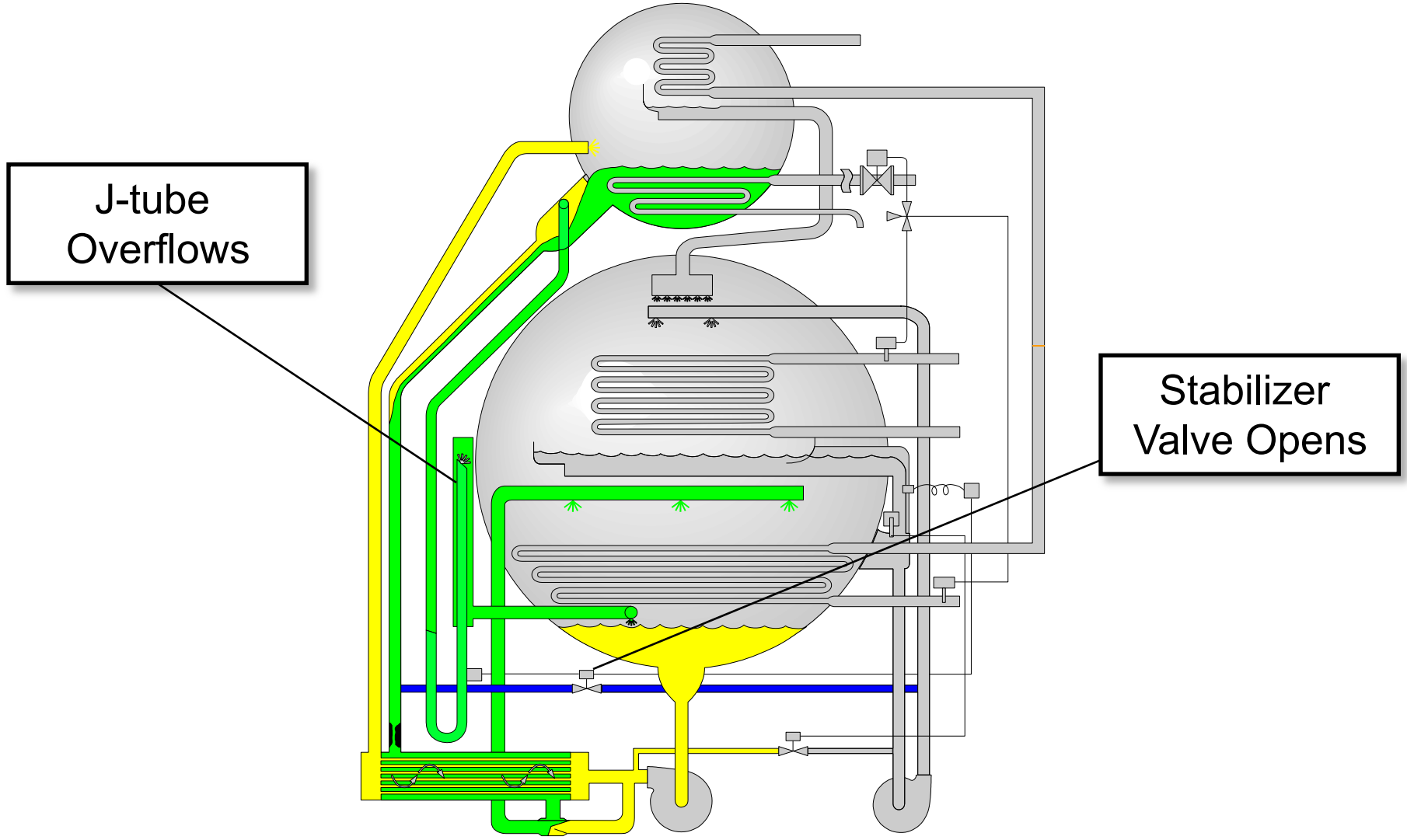




**Crystallization will most likely occur in the Solution Heat Exchanger**

**The place in the machine where we have the highest concentration and lowest temperature**

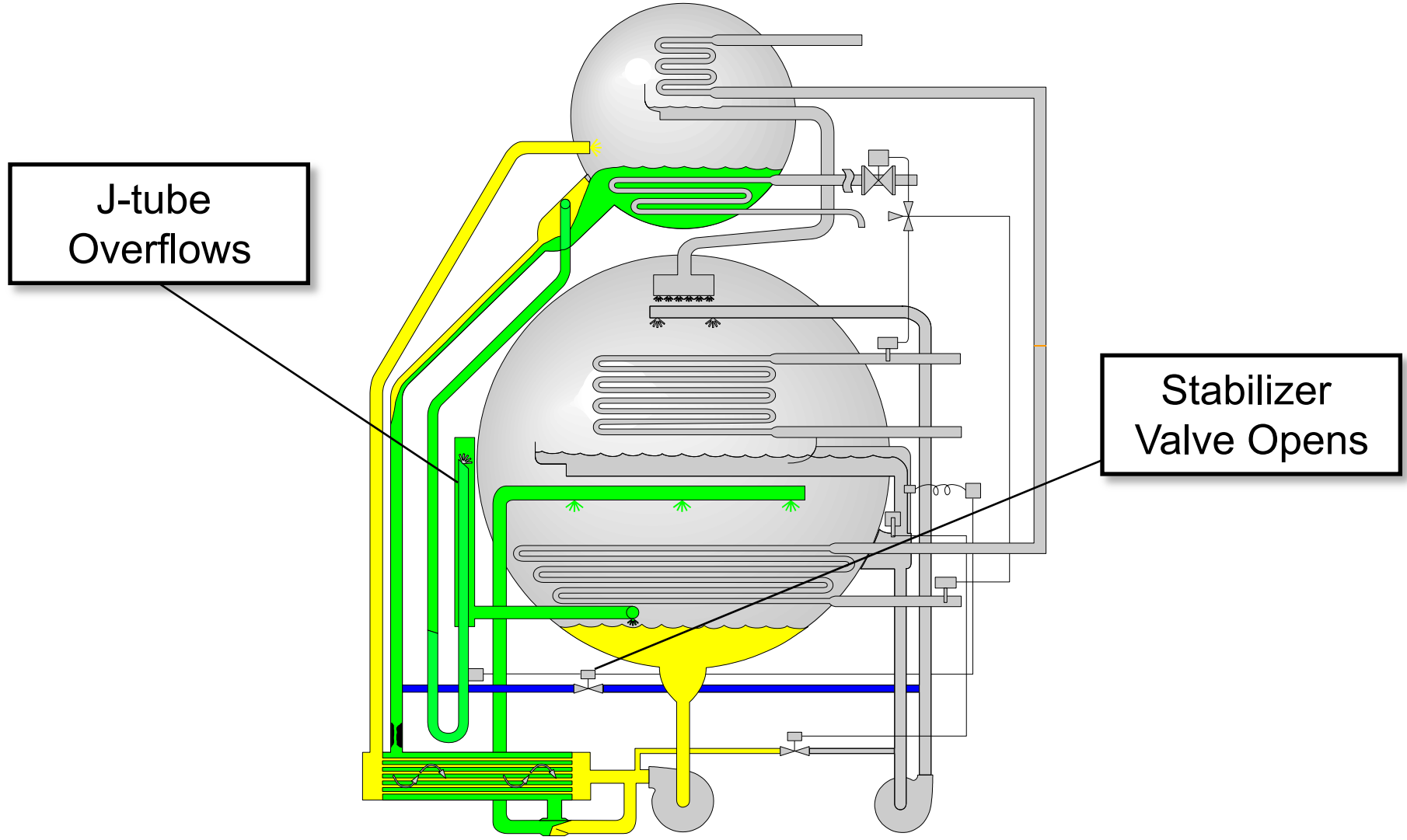
# Automatic Decrystallization



J-tube  
Overflows

Stabilizer  
Valve Opens

# Automatic Decrystallization

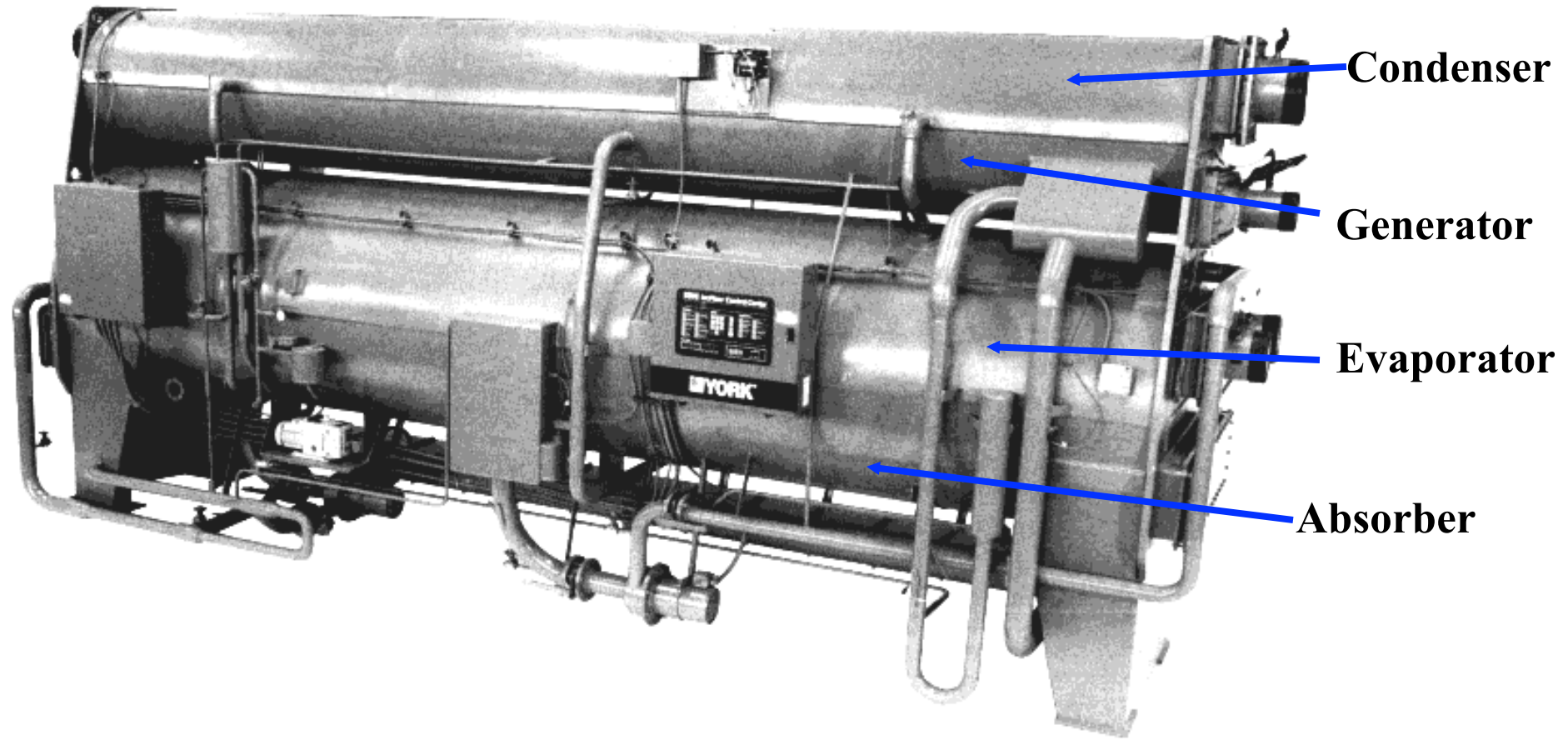


# Types of Absorption Chillers

- Single-stage, steam-fired
- Single-stage, hot water-fired
- Two-stage, steam-fired
- Two-stage, natural gas, direct-fired

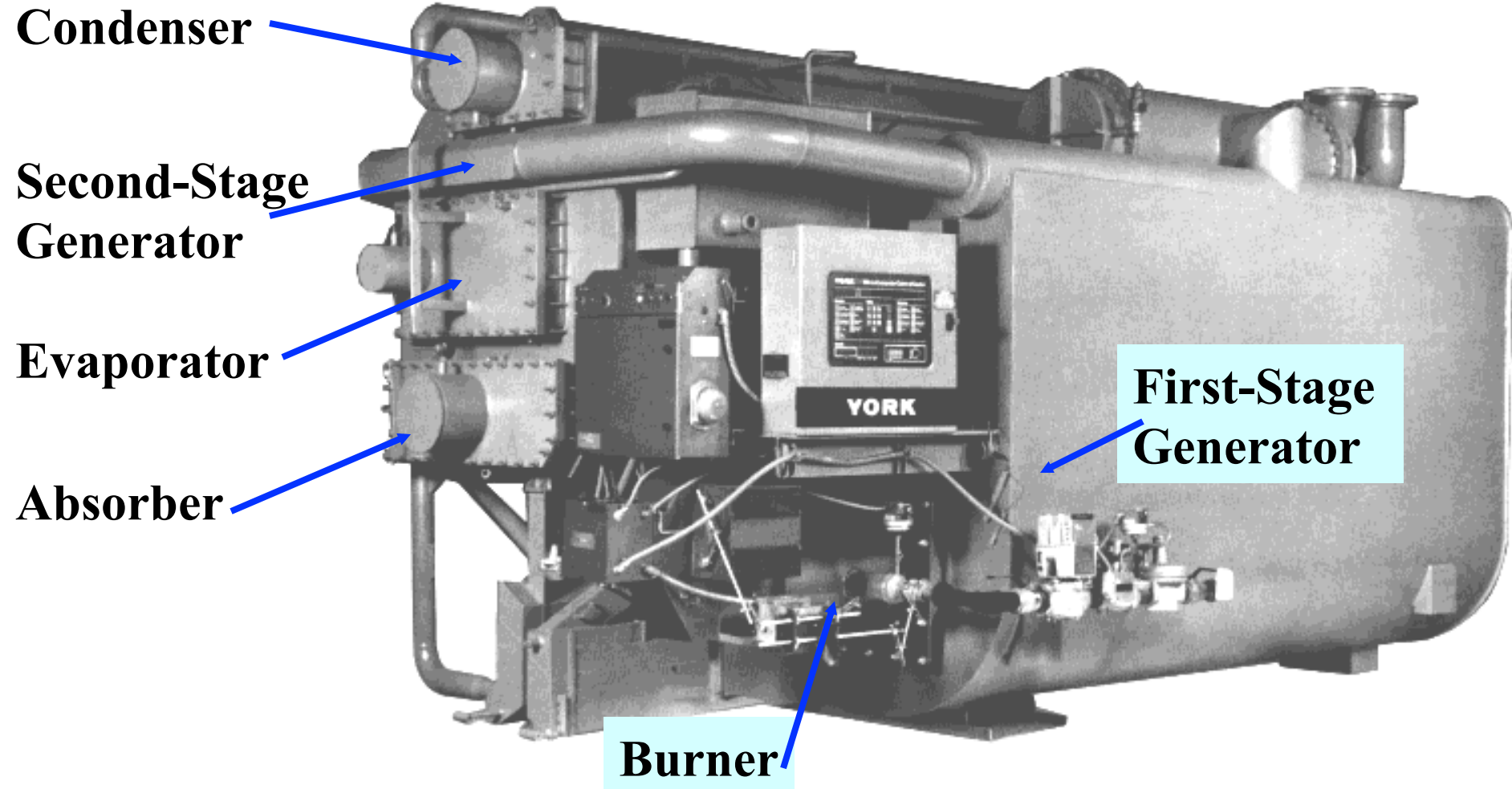


# Single-Stage, Steam-Fired Absorption Chiller (Previously Brand Named "IsoFlow")

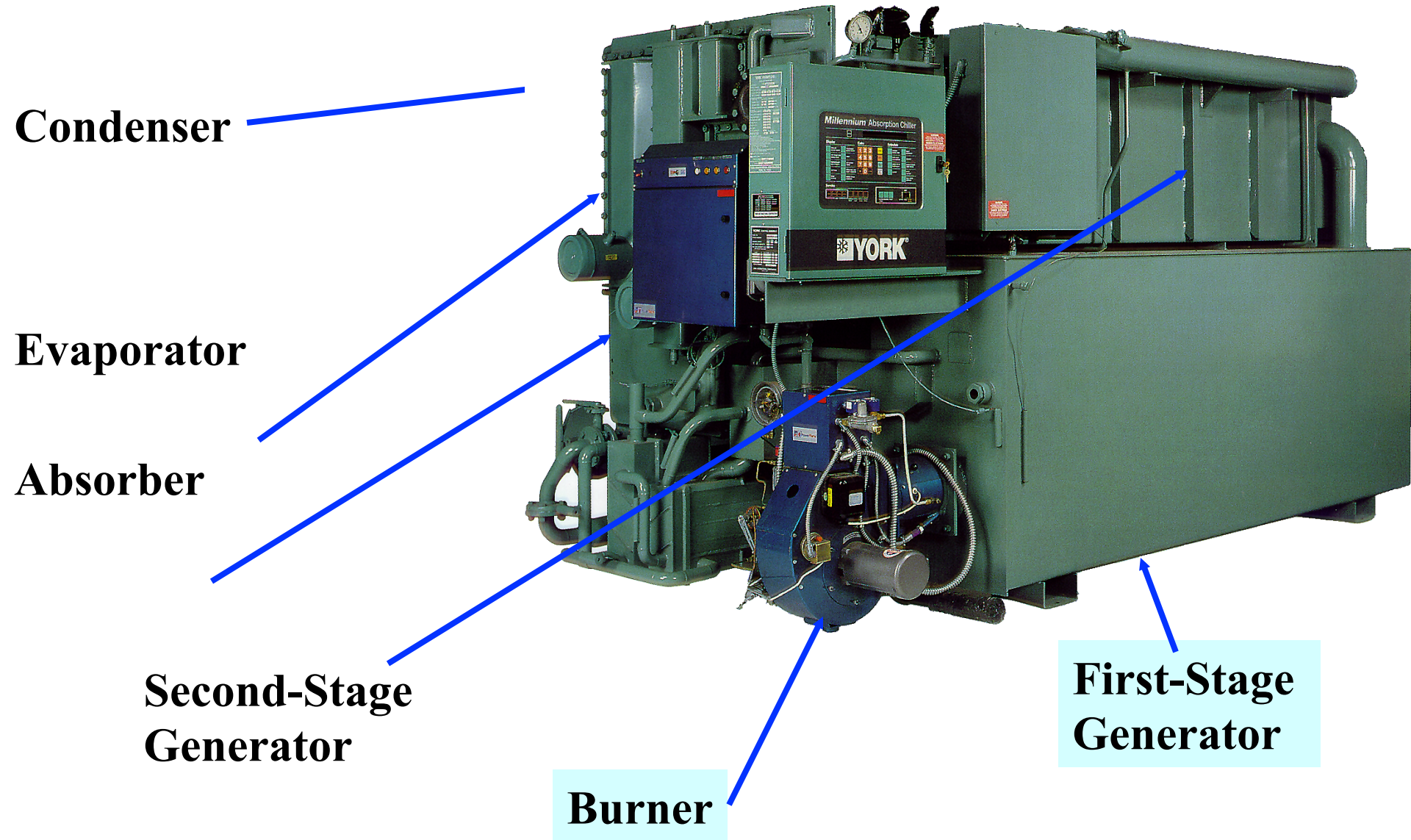


*Two-Stage  
Parallel Flow  
Absorption Cycle*

# Two-Stage, Natural Gas, Direct-Fired



# Two-Stage, Natural Gas, Direct-Fired (Previously Brand Named "ParaFlow")





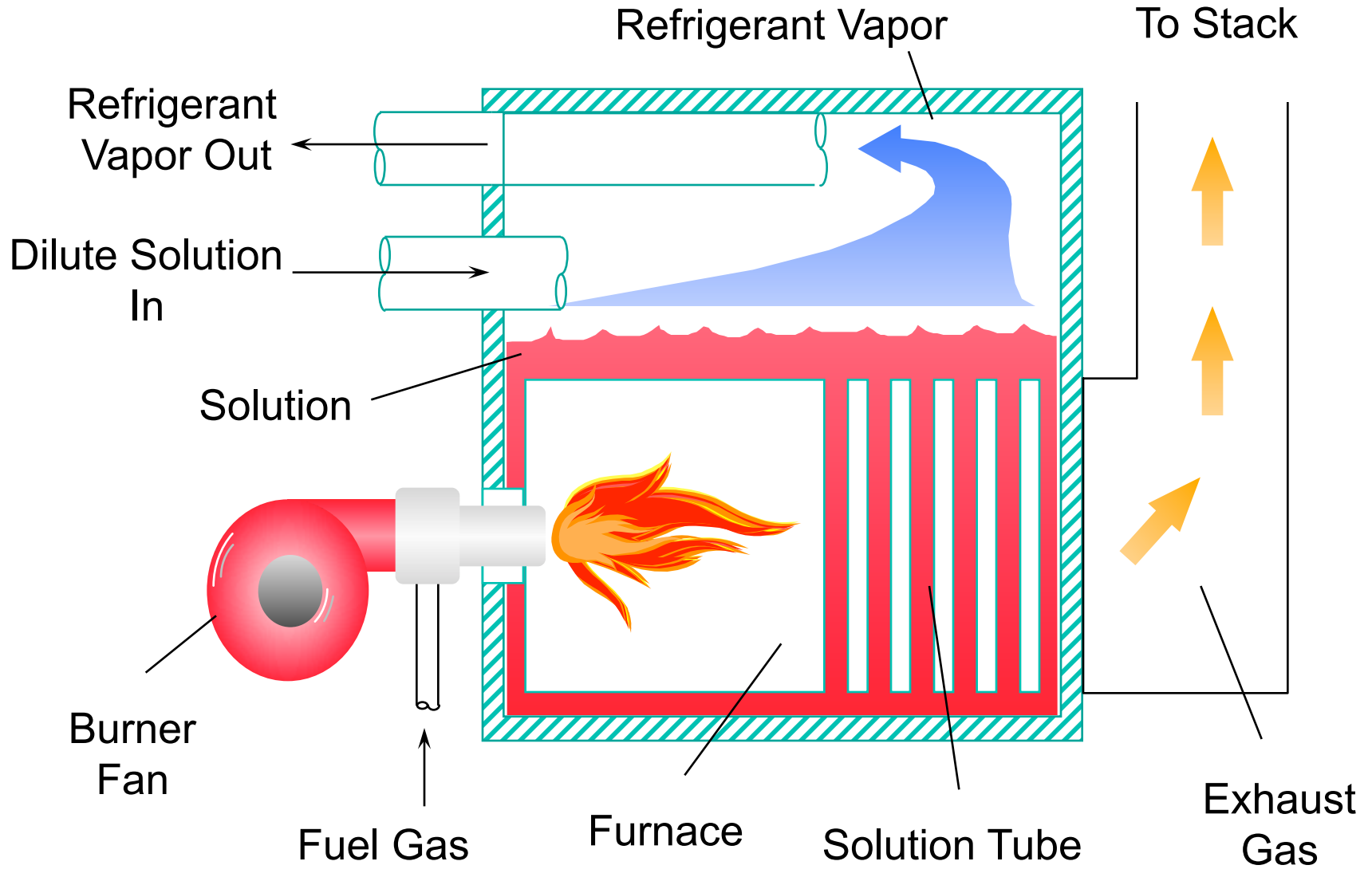




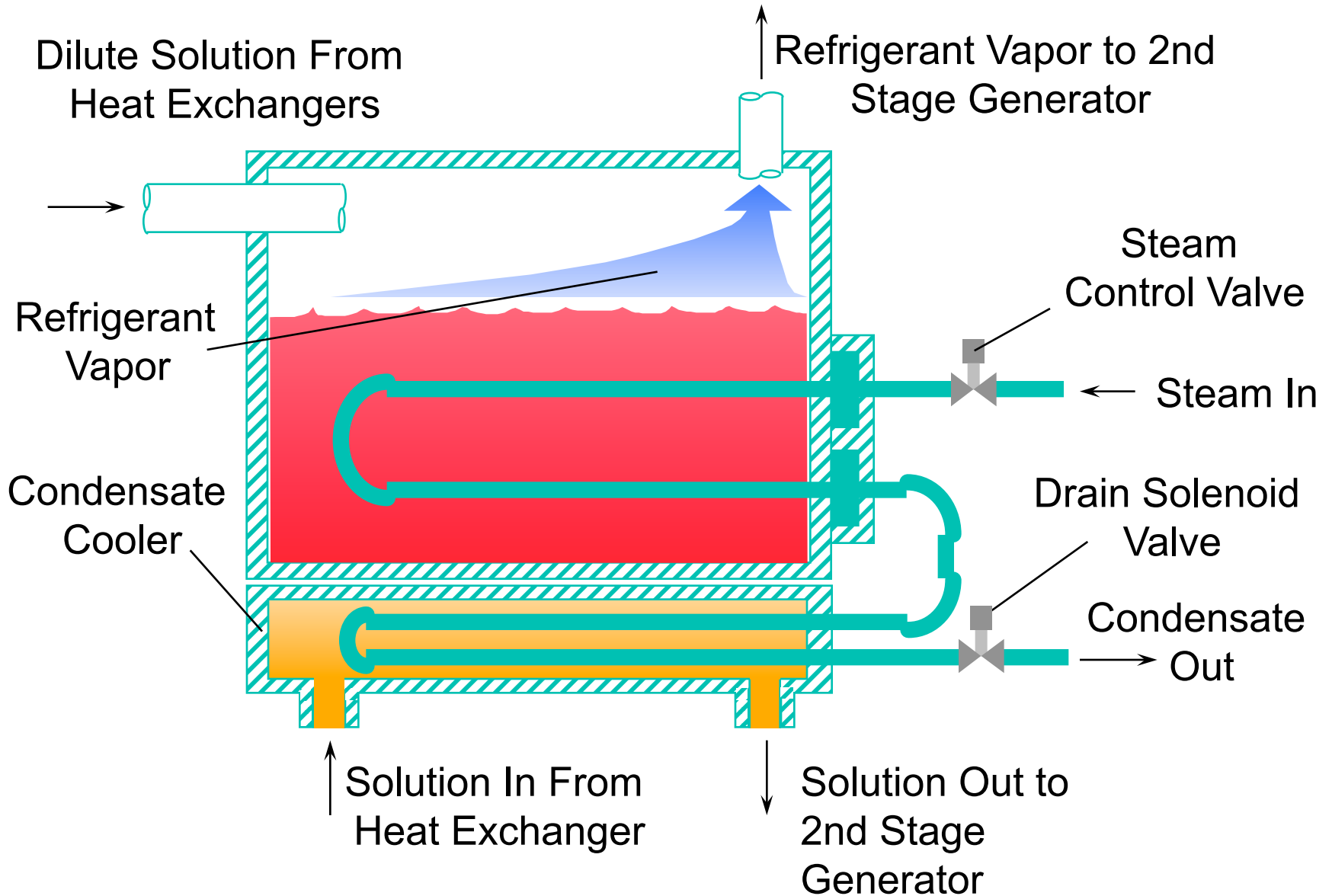




# Two Stage Direct Fired 1st Stage Generator



# Two Stage Steam Fired 1st Stage Generator



File: CHILLED WATER SYSTEMS EQUIPMENT MANUAL  
Absorption Water Cooling Systems Section  
Supersedes: Catalog EM-235 revised March 1959  
G-12, Form MS40 coded 459  
Equip. E & ER(L) - Int.

FEATURES  
DESCRIPTION  
SPECIFICATIONS

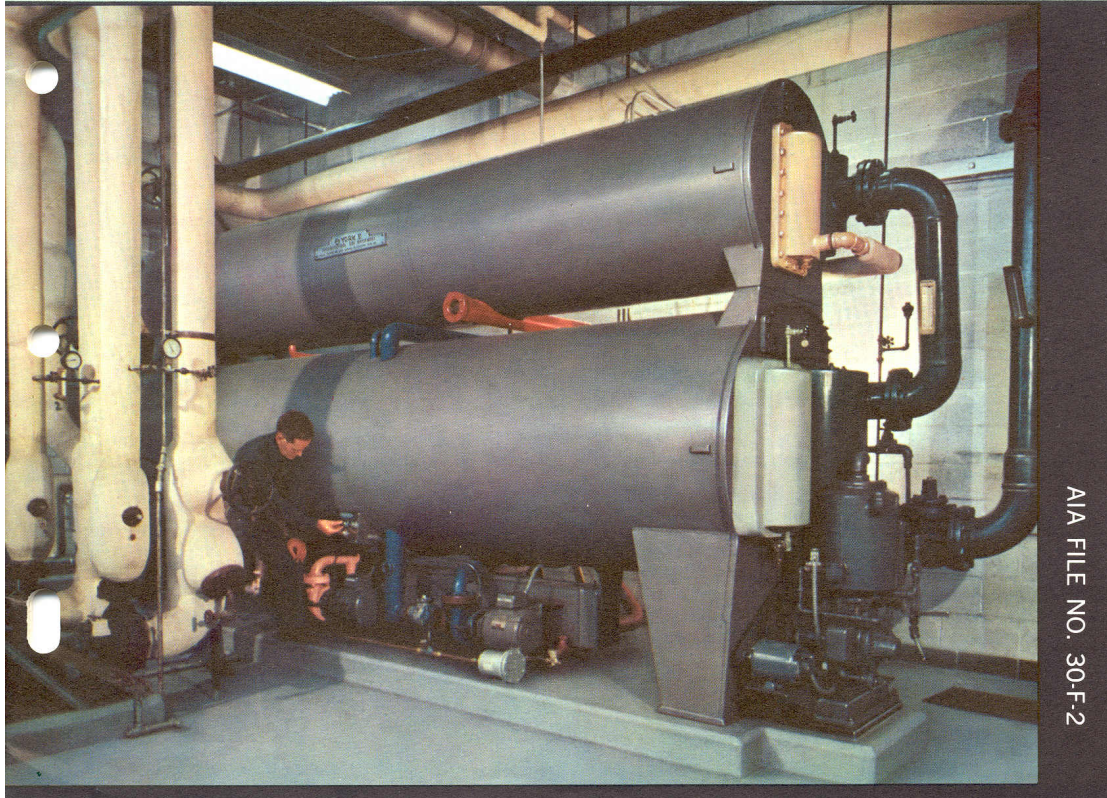
AIA FILE NO. 30-F-2

# YORK<sup>®</sup>

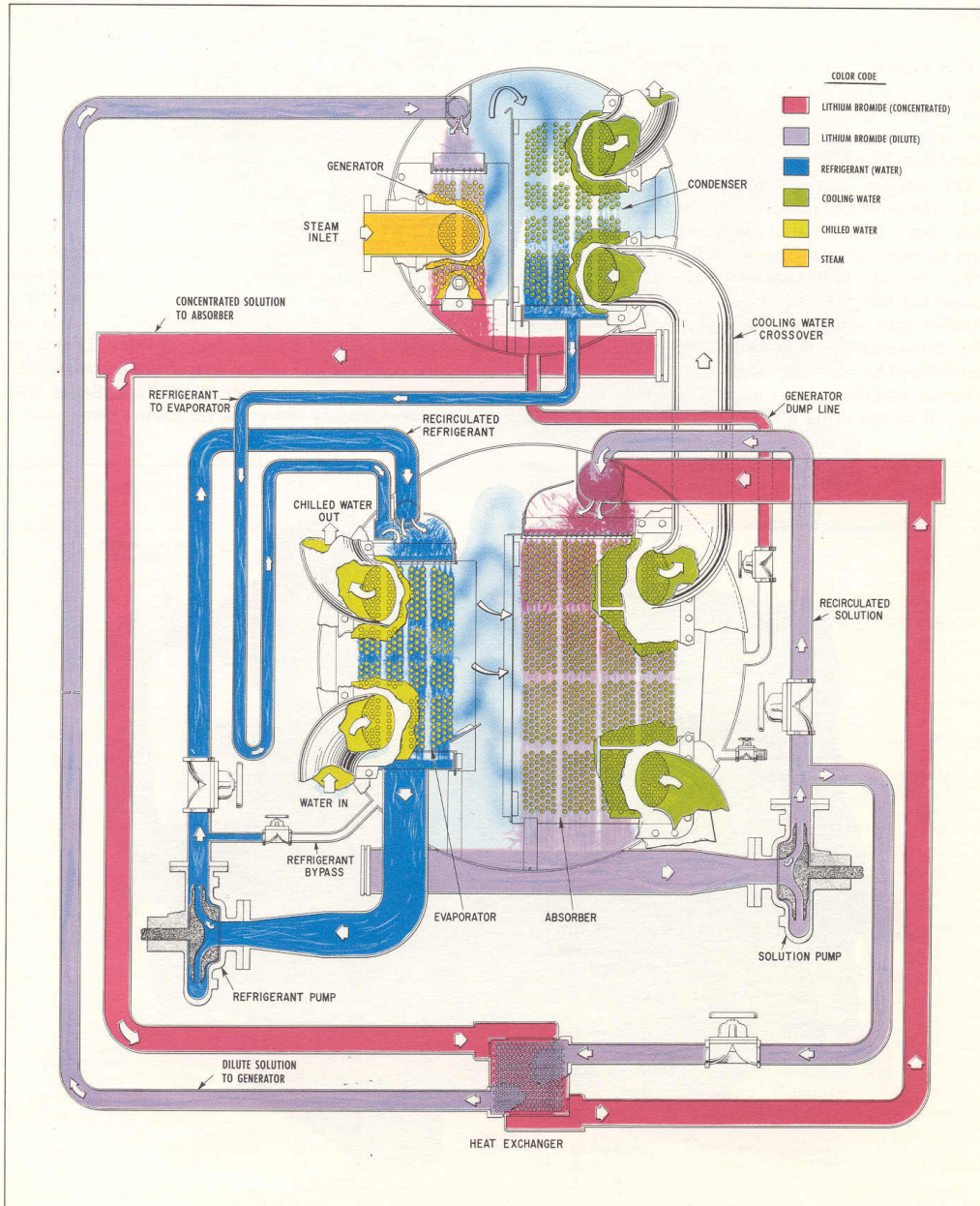
## ABSORPTION

### WATER CHILLING SYSTEMS

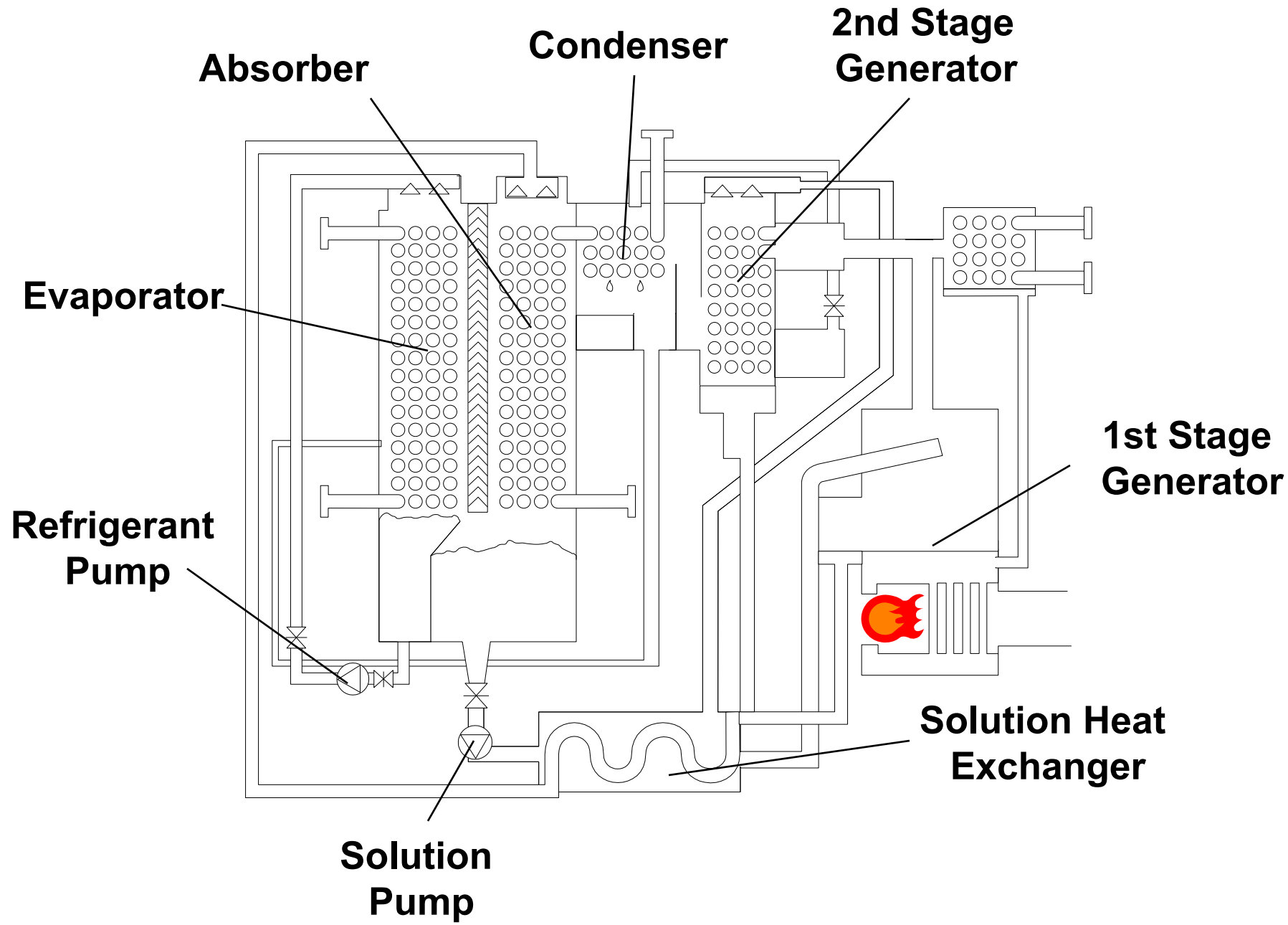
100 TO 740 TONS CAPACITY  
cooling with steam or hot water

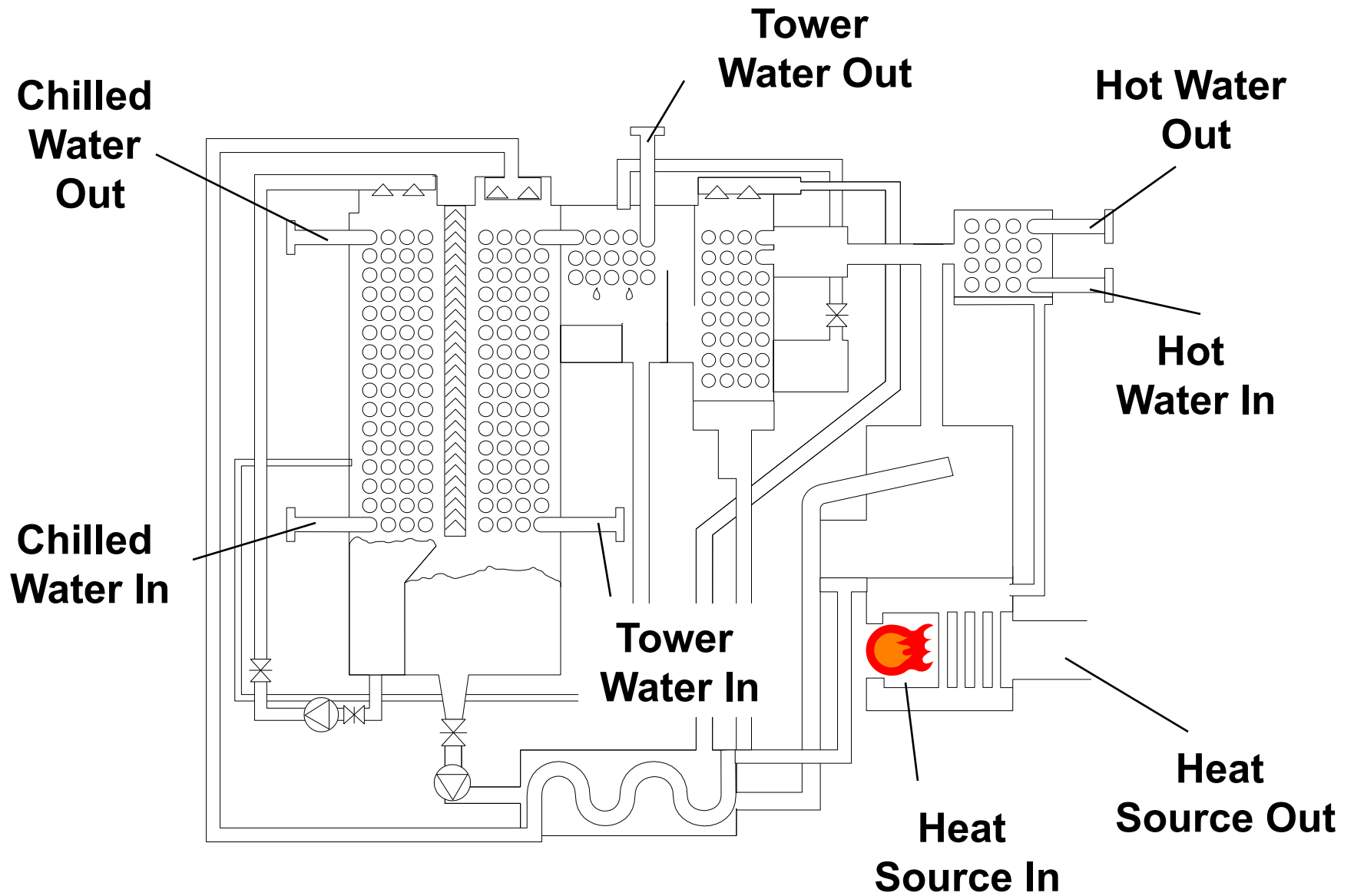


AIA FILE NO. 30-F-2

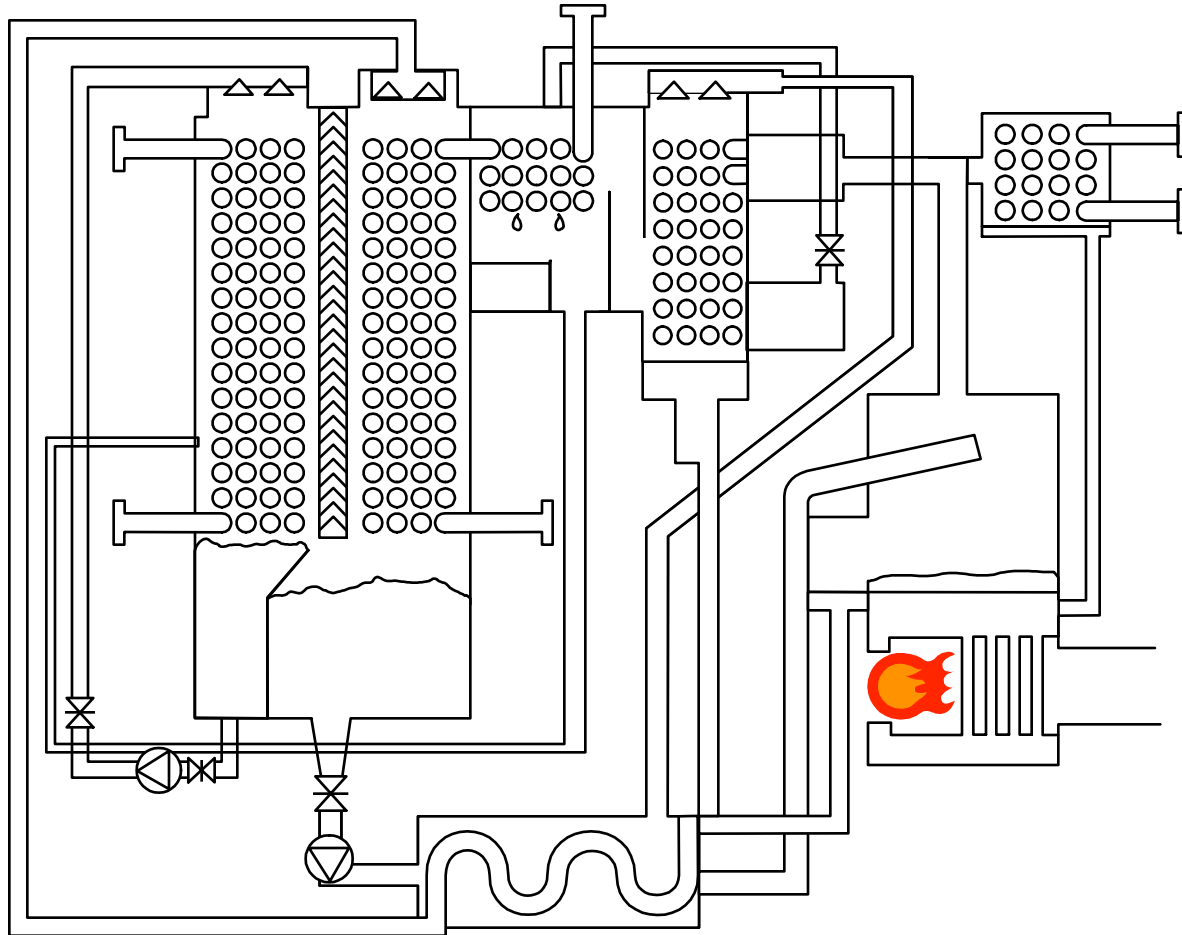


York Absorption System Cycle Diagram

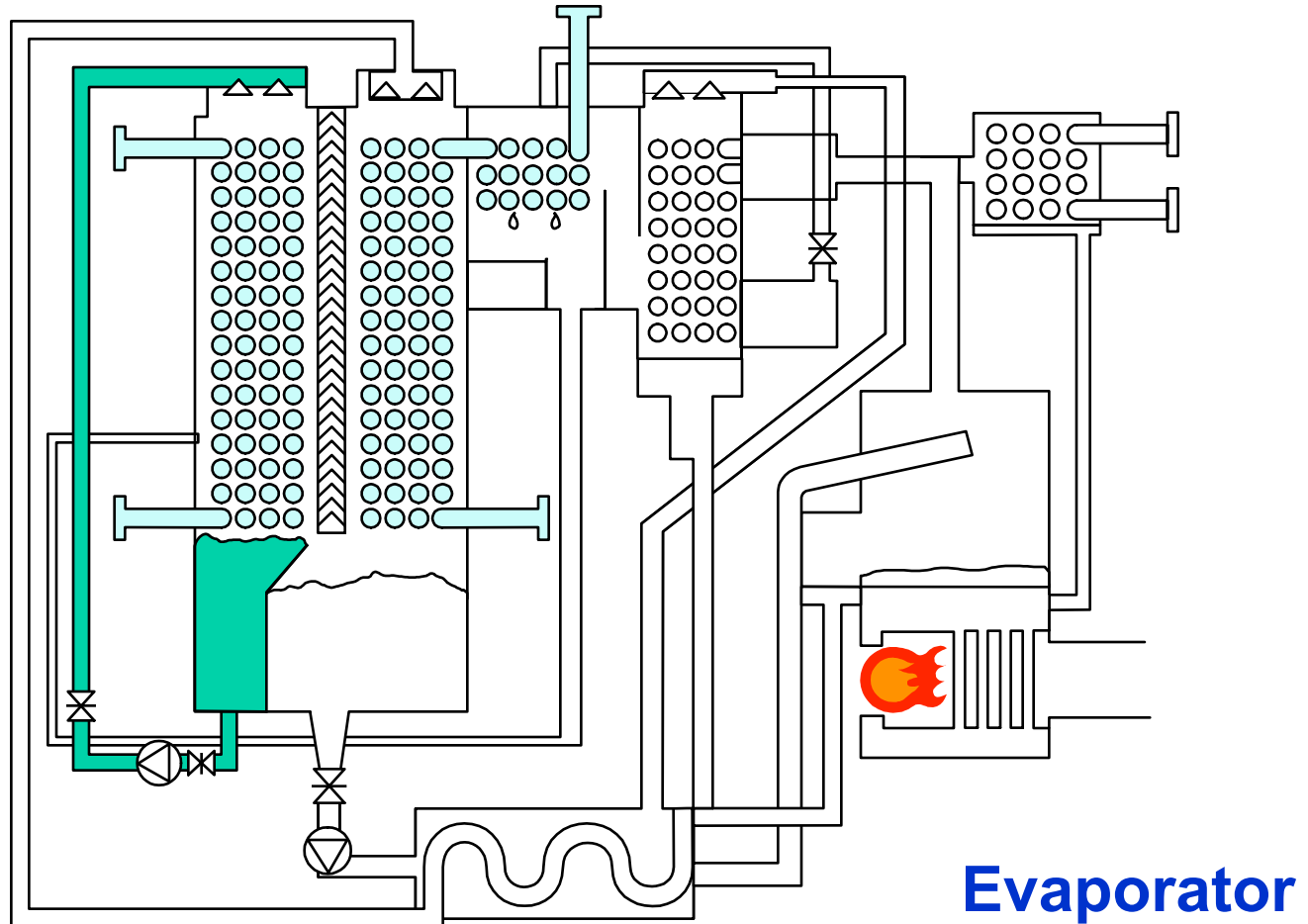




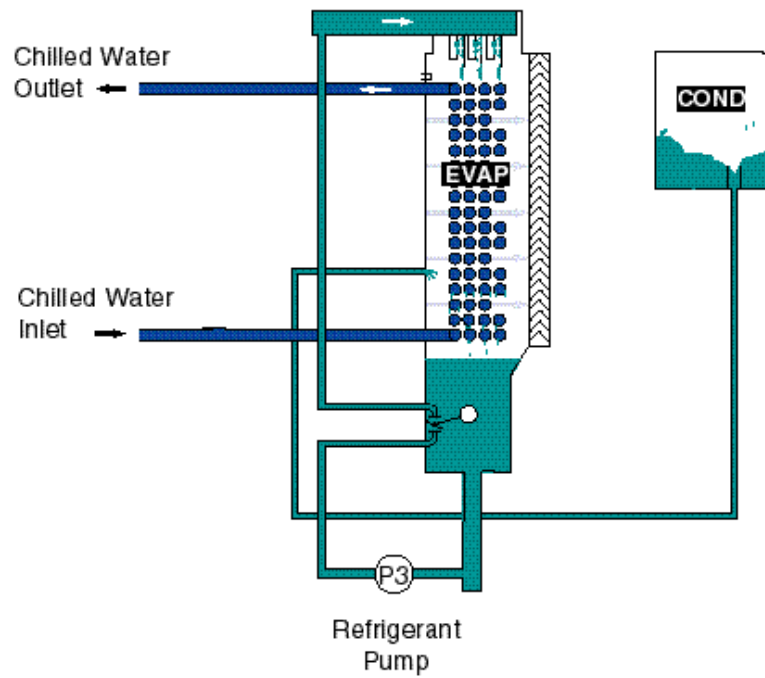
# Two-Stage Parallel Flow Absorption Cycle



# Two-Stage Parallel Flow Absorption Cycle



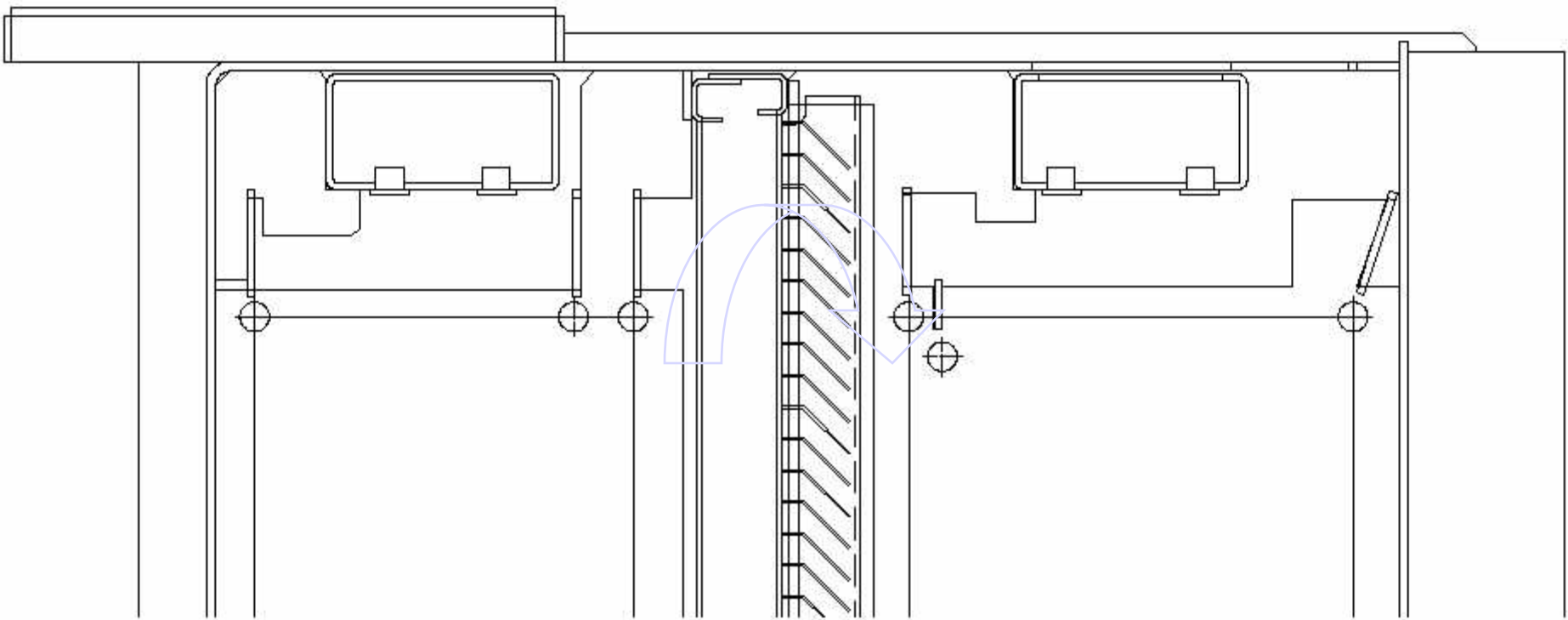
## Evaporator and Condenser Sections



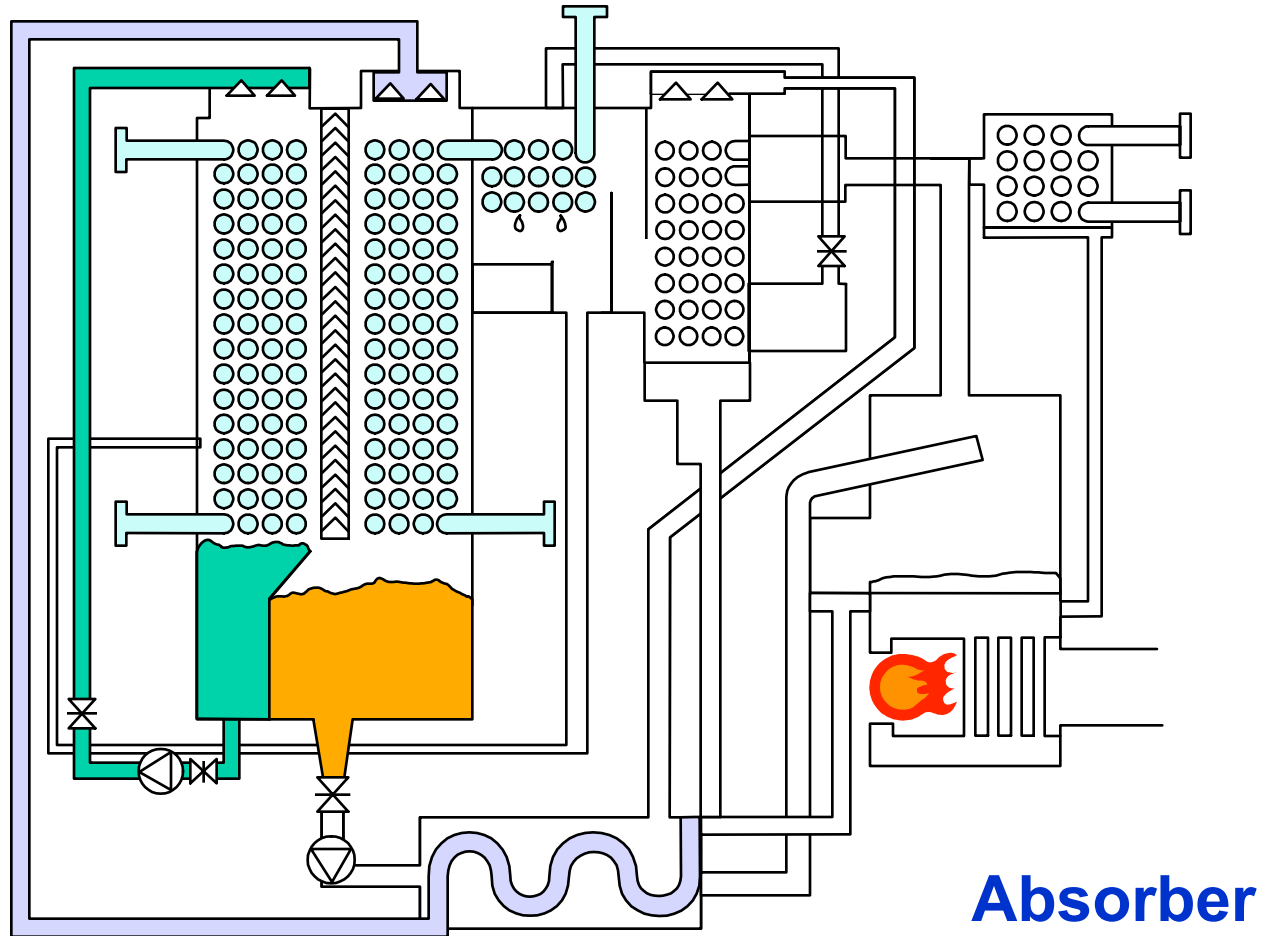


Soparfor P. 81 mm  
N° 222  
16/04/2005

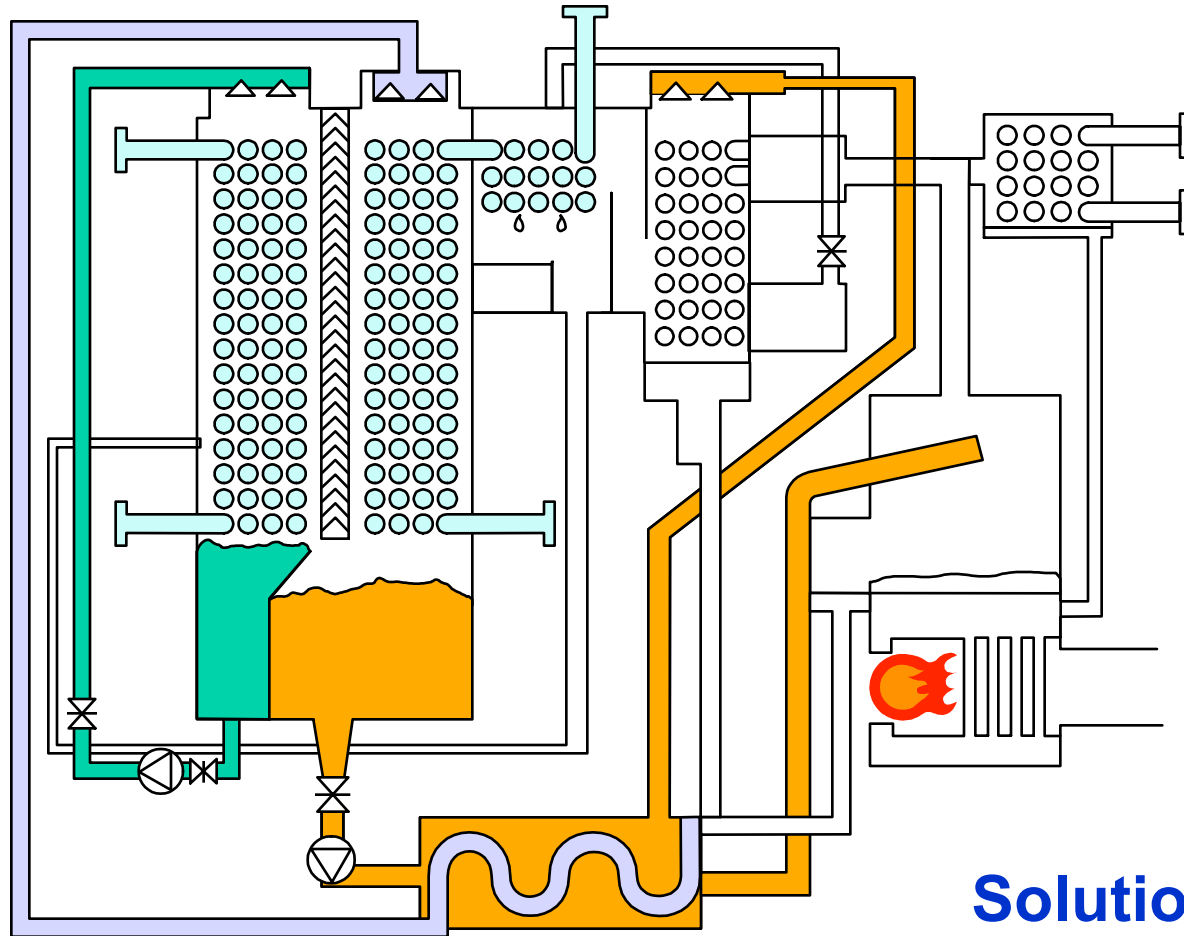
16/04/2005



# Two-Stage Parallel Flow Absorption Cycle

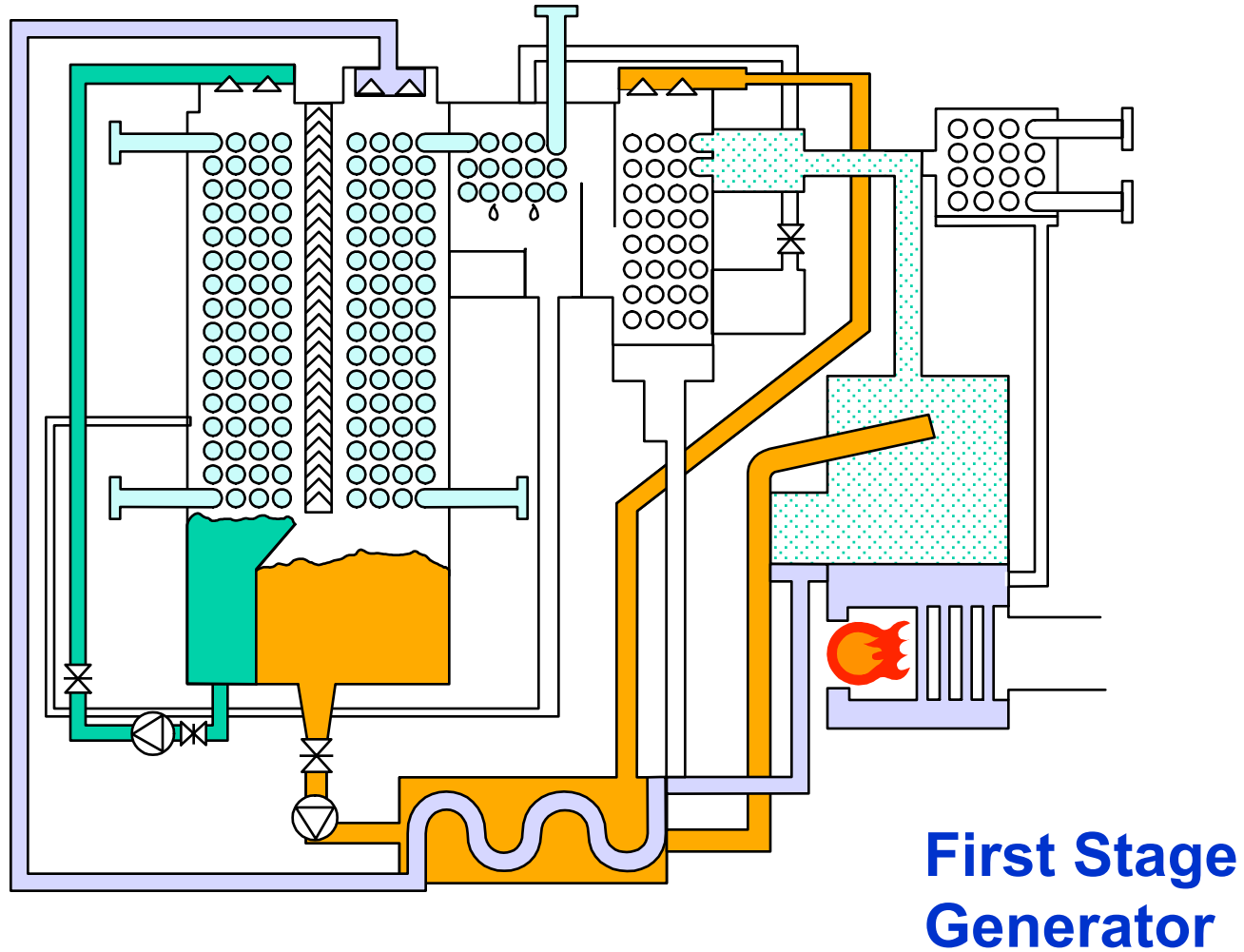


# Two-Stage Parallel Flow Absorption Cycle

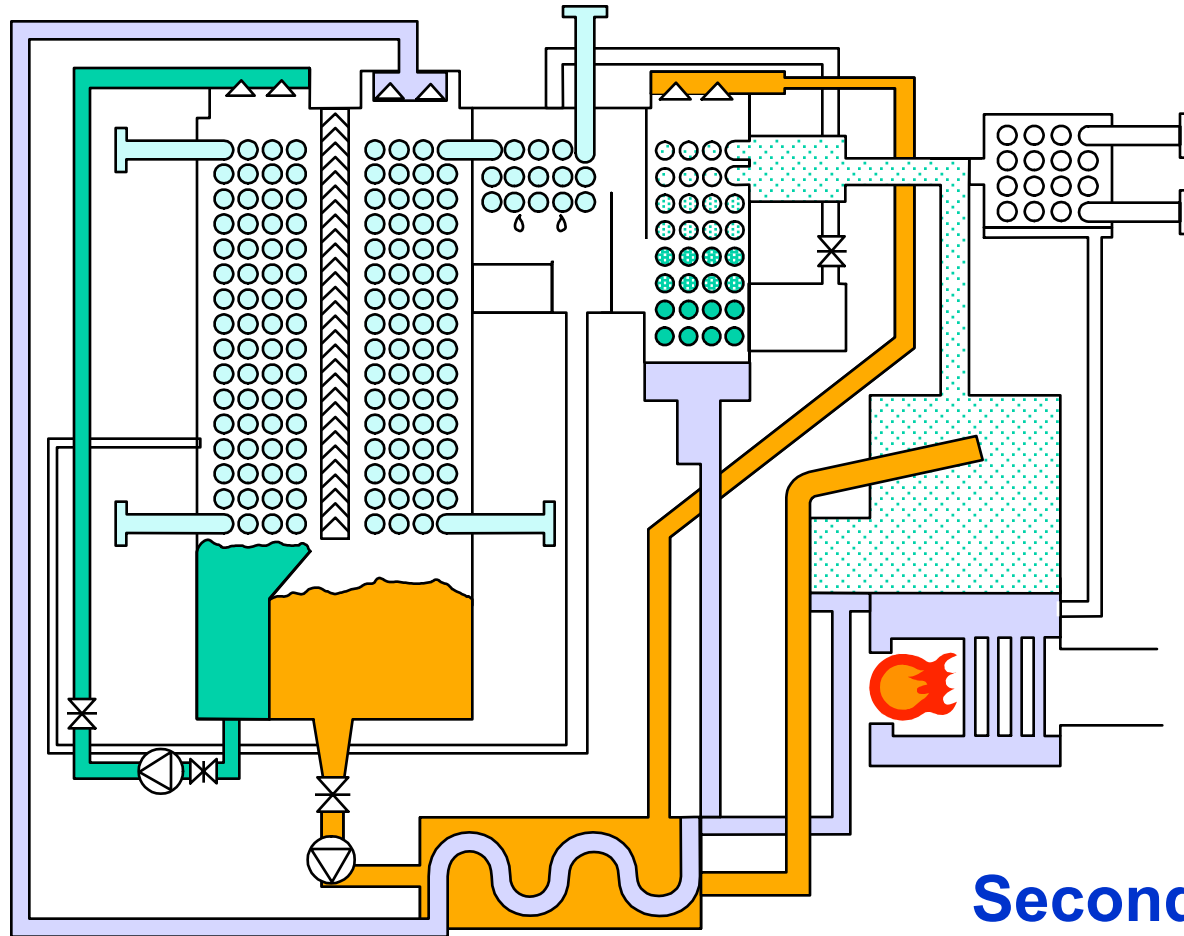


**Solution Heat Exchangers**

# Two-Stage Parallel Flow Absorption Cycle



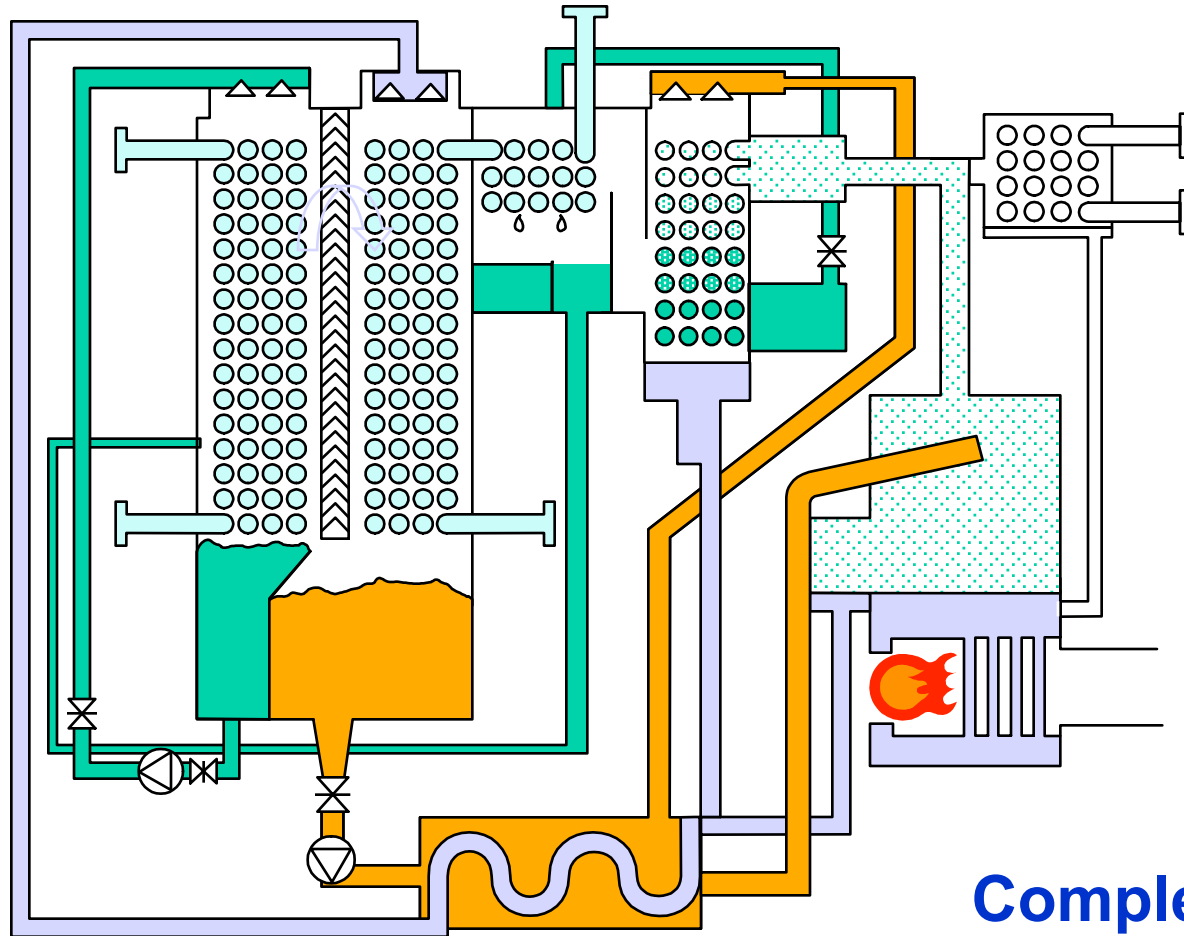
# Two-Stage Parallel Flow Absorption Cycle



**Second Stage  
Generator**

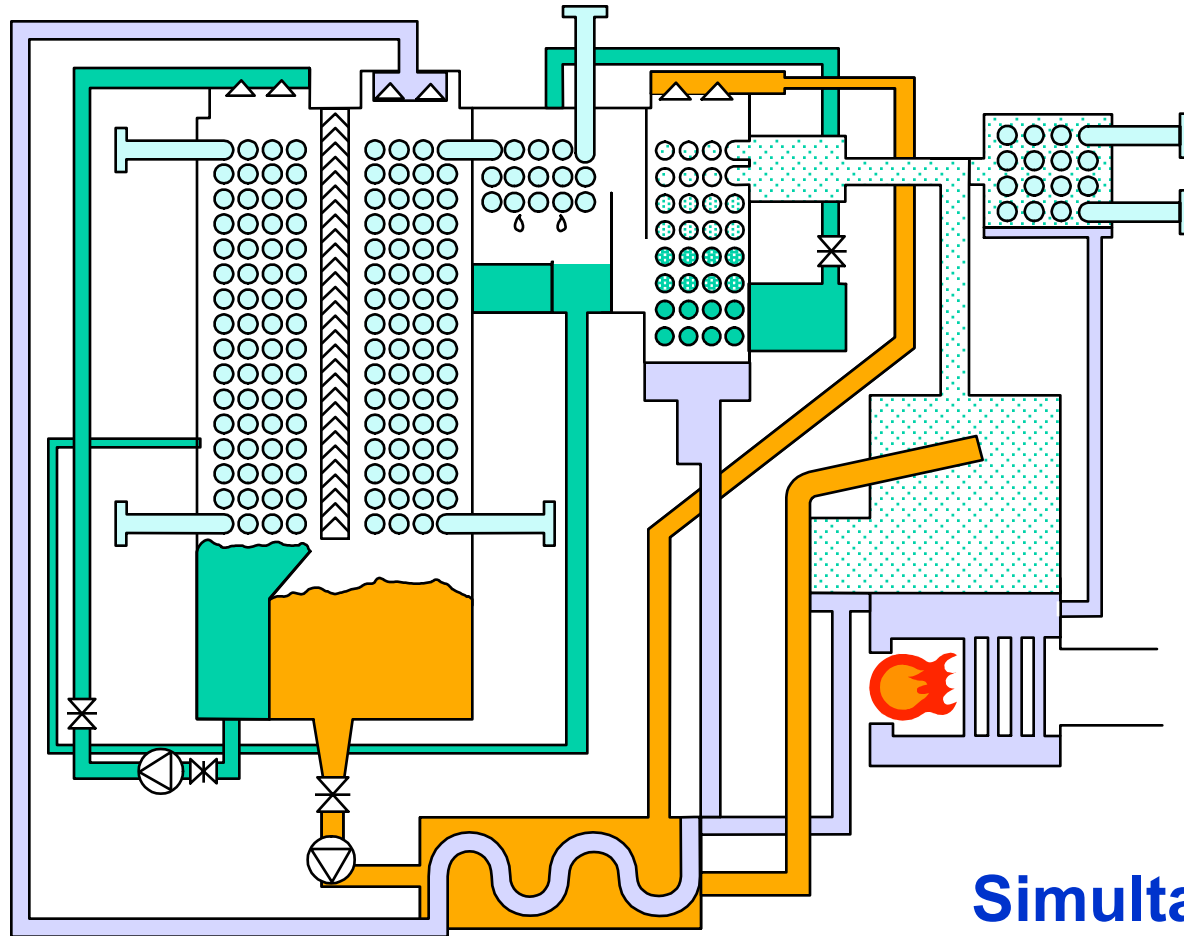


# Two-Stage Parallel Flow Absorption Cycle



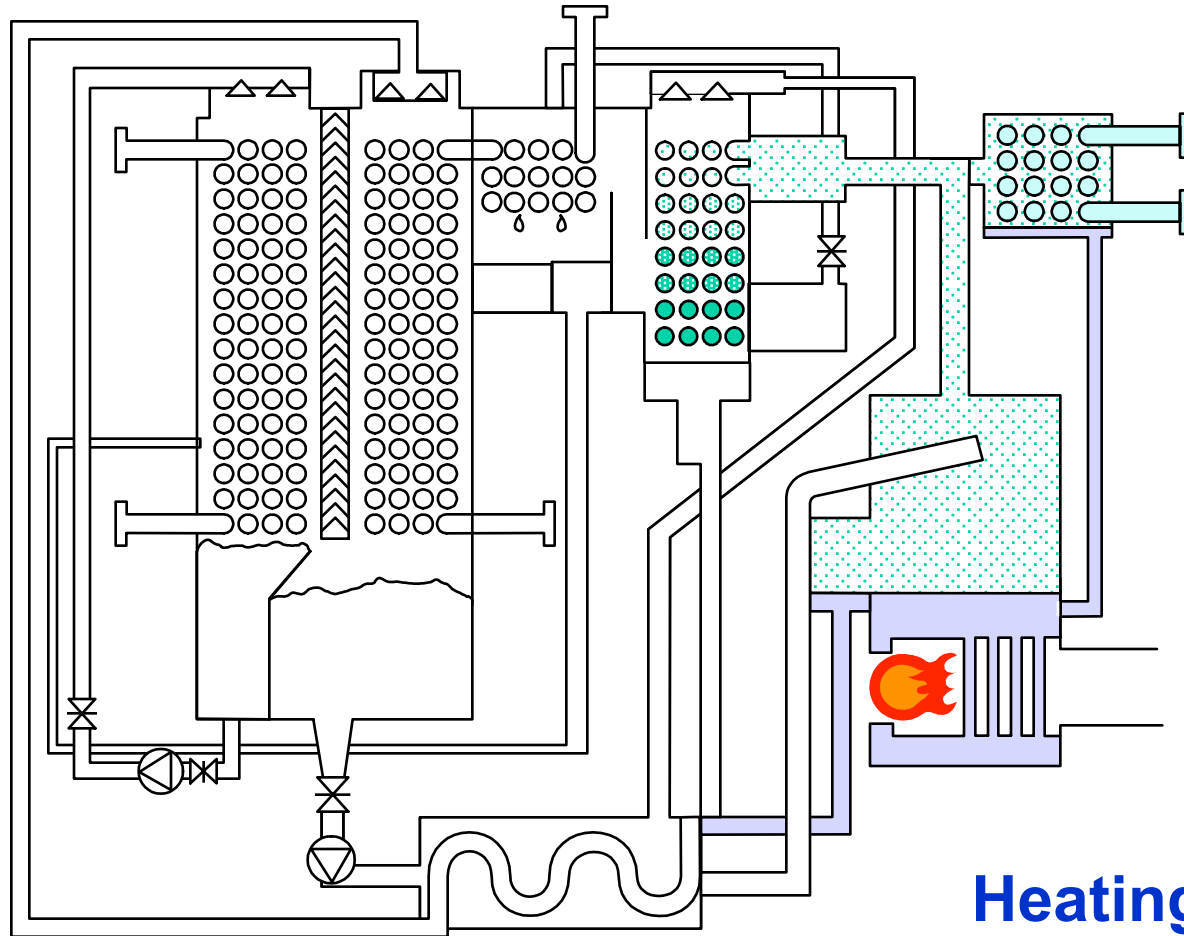
**Complete  
Cooling Cycle**

# Two-Stage Parallel Flow Absorption Cycle



**Simultaneous  
Heating and  
Cooling**

# Two-Stage Parallel Flow Absorption Cycle



# Absorption Chiller Cooling Tower Requirements

- Absorbers are heat-driven chillers
- Absorbers have more heat to reject per ton of capacity compared to a mechanical refrigeration chiller
- Cooling towers must be larger
- Cooling tower capacity control more critical with an absorber (bypass, fan speed control)

CHILLER TYPE	CENTRIFUGAL	TWO-STAGE ABSORBER	SINGLE-STAGE ABSORBER
Typical Condenser Water Flow	3 cgpm/ton, 85f to 95f	3 cgpm/ton, 85f to 100f or 4.5 cgpm, 85f to 95f	3.6 cgpm/ton, 85f to 101.4f
Relative Quantity of Condenser Water Flow	1.0	1.5	2.0

# Avoiding Absorption Chiller Corrosion

- Add inhibitors
- Lithium chromate
- Lithium nitrate . Lithium Molybdate
- Test solution chemistry\* every year
- Test chiller for leaks every year
- Automatic purge

**\*Amount of inhibitors, solution PH, amount of copper in solution, amount of steel in solution**

# Inhibitors and Environmental Concerns

- Lithium bromide solution, chromate inhibited
  - Classed as a hazardous waste
  - Must be disposed of according to EPA, city and state regulations\*
- Lithium bromide solution, molybdate inhibited
- New Inhibitors are even more effective and are environmentally friendly
  - Not a hazardous waste

**\*Avoid contact when servicing. Have adequate ventilation or respirator. Wear rubber gloves and goggles. Chromate is a potential carcinogen (not known for sure yet). Follow ASHRAE guidelines on machine room ventilation**

# Avoiding Crystallization

- Automatic dilution cycle before shutdown
- Automatic decrystallization control
- Operate chiller with lithium bromide solution as far away from the crystallization line as possible (efficiency tradeoff)

## Single-Stage Versus Two-Stage Absorption Chillers

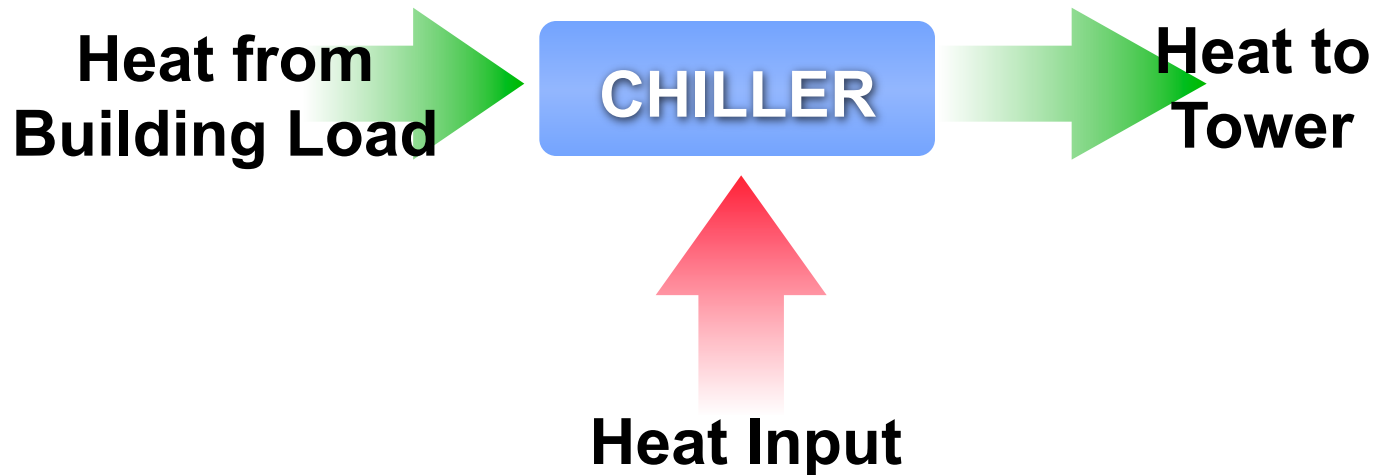
- Two-stage absorber uses refrigerant vapor heat produced in the first stage generator to drive off refrigerant in second stage generator
- As much as 40% more reconcentrated refrigerant is produced by a two stage absorber using the same energy as a single stage absorber

# Single-Stage Versus Two-Stage Absorption Chillers

- Types of single-stage absorbers
  - Steam fired
  - Hot water fired
- Types of two-stage absorbers
  - Steam fired
  - Natural gas fired
  - Oil fired
  - Propane fired
- Single-stage COP (coefficient of performance)  $\cong 0.65$
- Two-stage COP  $\cong 0.98$

# COP Calculation

$$\text{Coefficient of Performance} = \frac{\text{Heat from Building Load}}{\text{Heat Input}}$$



# Typical COP' s

CHILLER TYPE	COP
Absorption	
Single Stage	0.68
Two stage direct fired	1.00
Two stage steam fired	1.19
Gas engine driven centrifugal	1.90
Electric motor driven centrifugal	6.00

# Good Absorption Cooling Applications

- Peak electrical load shaving
- Gas company load balancing
- Simultaneous heating and cooling requirement (two-stage)
- Inadequate electrical service
- Limited machine room space for chillers and boilers (two-stage)
- CFC elimination
- Utility deregulation

# Questions

Thank You

